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PCT

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English

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(71) Applicant (for all designated States except US): JANSSEN PHARMACEUTICA N.V. [BE/BE]; Turnhoutseweg 30, B-2340 Beerse (BE).

(72) Inventors; and

(75) Inventors/Applicants (for US only): JANSSENS, Frans, Eduard [BE/BE]; Janssen Pharmaceutica N.V., Turnhoutseweg 30, B-2340 Beerse (BE). LACRAMPE, Jean, Fernand, Armand [FR/FR]; Janssen-Cilag S.A., 1, rue Camille Desmoulins, TSA 91003, F-92787 Issy-les-Moulineaux Cedex 9 (FR). GUILLEMONT, Jérôme, Emile, Georges [FR/FR]; Janssen-Cilag S.A., 1, rue Camille Desmoulins, TSA 91003, F-92787 Issy-les-Moulineaux Cedex 9 (FR). VENET, Marc, Gaston [FR/FR]; Janssen-Cilag S.A., 1, rue Camille Desmoulins, TSA 91003, F-92787 Issy-les-Moulineaux Cedex 9 (FR). ANDRIES, Koenraad, Jozef, Lodenwijk, Marcel [BE/BE]; Janssen Pharmaceutica N.V., Turnhoutseweg 30, B-2340 Beerse (BE).

(74) Agent: QUAGHEBEUR, Luc; Janssen Pharmaceutica N.V., Patent Department - 3547, Turnhoutseweg 30, B-2340 Beerse (BE).

(81) Designated States (national): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CR, CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZW.

[Continued on next page]

(54) Title: RESPIRATORY SYNCYTIAL VIRUS REPLICATION INHIBITORS

(57) Abstract: The present invention concerns compounds of formula (I), prodrugs, N-oxides, addition salts, quaternary amines, metal complexes and stereochemically isomeric forms thereof wherein -a1=a2-a3=a4- represents a radical of formula -CH=CH-CH=CH-; -N=CH-CH=CH-; -CH=N-CH=CH-; -CH=CH-N=CH-; -CH=CH-CH=N-; wherein each hydrogen atom may optionally be substituted; Q is a radical of formulae (b-1), (b-2), (b-3), (b-4), (b-5), (b-6), (b-7) and (b-8), wherein Alk is C₁₋₆alkanediyl; Y¹ is a bivalent radical of formula -NR²- or -CH(NR²R⁴)-; X¹ is NR⁴, S, S(=O), S(=O)₂, O, CH₂, C(=O), CH(=CH₂), CH(OH), CH(CH₃), CH(OCH₃), CH(SCH₃), CH(NR^{5a}R^{5b}), CH₂-NR⁴ or NR⁴-CH₂; X² is a direct bond, CH₂, C(=O), NR⁴, C₁₋₄alkyl-NR⁴, NR⁴-C₁₋₄alkyl, t is 2 to 5; u is 1 to 5; v is 2 or 3; and whereby each hydrogen in Alk and in (b-3), (b-4), (b-5), (b-6), (b-7) and (b-8), may optionally be replaced by R3; provided that when R3 is hydroxy or C1-6alkyloxy, then R3 cannot replace a hydrogen atom in the α position relative to a nitrogen atom; G is a direct bond or optionally substituted C_{1-10} alkanediyl; R^1 is an optionally substituted bicyclic heterocycle; R² is hydrogen, formyl, C₁₋₆alkylcarbonyl, Hetcarbonyl, pyrrolidinyl, piperidinyl, homopiperidinyl, C_{3-7} cycloalkyl or C_{1-10} alkyl substituted with $N(R^6)_2$ and optionally with another substituent; R^3 is hydrogen, hydroxy, C₁₋₆alkyl, C₁₋₆alkyloxy, arylC₁₋₆alkyl or arylC₁₋₆alkyloxy, R⁴ is hydrogen, C₁₋₆alkyl or arylC₁₋₆alkyl; R^{5a}, R^{5b}, R^{5c} and R^{5d} are hydrogen or C1-6alkyl; or R5a and R5b, or R5c and R5d taken together from a bivalent radical of formula -(CH2)s- wherein S is 4 or 5; R6 is hydrogen, C1-alkyl, formyl, hydroxyC1-alkyl, C1-alkylcarbonyl or C1-alkyloxycarbonyl; aryl is optionally substituted phenyl; Het is pyridyl, pyrimidinyl, pyryzinyl, pyridazinyl; as respiratory syncytial virus replication inhibitors; their preparation, compositions containing them and their use as a medicine.

WO 01/00615 A1



(84) Designated States (regional): ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG).

Published:

- With international search report.

— Before the expiration of the time limit for amending the claims and to be republished in the event of receipt of amendments.

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.



INTERNATIONAL SEARCH REPORT

(PCT Article 18 and Rules 43 and 44)

Applicant's or agent's file reference	L (Form PCT/ISA/2	of Transmittal of International Search Report 20) as well as, where applicable, item 5 below.							
JAB 1500-PCT	ACTION								
International application No.	International filing date (day/month/year)	(Earliest) Priority Date (day/month/year)							
PCT/EP 00/05677	20/06/2000	28/06/1999							
Applicant									
JANSSEN PHARMACEUTICA N.V									
CANSSEN THANNACEUTICA W.V	•								
This International Search Report has been according to Article 18. A copy is being tra	n prepared by this International Searching Auth ansmitted to the International Bureau.	nority and is transmitted to the applicant							
This International Search Report consists It is also accompanied by	of a total of sheets. a copy of each prior art document cited in this	report.							
Basis of the report									
	international search was carried out on the bases otherwise indicated under this item.	sis of the international application in the							
the international search w Authority (Rule 23.1(b)).	as carried out on the basis of a translation of the	he international application furnished to this							
b. With regard to any nucleotide an was carried out on the basis of the		ternational application, the international search							
	nal application in written form.								
filed together with the inte	rnational application in computer readable for	n.							
furnished subsequently to	furnished subsequently to this Authority in written form.								
· · ·	furnished subsequently to this Authority in computer readble form.								
	the statement that the subsequently furnished written sequence listing does not go beyond the disclosure in the international application as filed has been furnished.								
the statement that the info furnished	the statement that the information recorded in computer readable form is identical to the written sequence listing has been								
2. X Certain claims were four	nd unsearchable (See Box I).								
3. Unity of Invention is lac	· ·								
4. With regard to the title ,									
4. With regard to the title, the text is approved as su	hmitted by the applicant								
	hed by this Authority to read as follows:								
	,								
5. With regard to the abstract,									
X the text is approved as su									
	hed, according to Rule 38.2(b), by this Authorite date of mailing of this international search rep								
6. The figure of the drawlngs to be publ									
as suggested by the appli	•	None of the figures.							
because the applicant fail									
because this figure better	characterizes the invention.								

FURTHER INFORMATION CONTINUED FROM PCT/ISA/ 210

Continuation of Box I.2

Present claims 1 to 15. relate to an extremely large number of possible compounds. Support within the meaning of Article 6 PCT and/or disclosure within the meaning of Article 5 PCT is to be found, however, for only a very small proportion of the compounds claimed. In the present case, the claims so lack support, and the application so lacks disclosure, that a meaningful search over the whole of the claimed scope is impossible. Consequently, the search has been carried out for those parts of the claims which appear to be supported and disclosed, namely those parts relating to the compounds of formula I where Q is a 1-R2-piperidyl-4-amino or amino(cyclo)alkylamino group and their intermediates as described in the examples of tables 3 to 13.

The applicant's attention is drawn to the fact that claims, or parts of claims, relating to inventions in respect of which no international search report has been established need not be the subject of an international preliminary examination (Rule 66.1(e) PCT). The applicant is advised that the EPO policy when acting as an International Preliminary Examining Authority is normally not to carry out a preliminary examination on matter which has not been searched. This is the case irrespective of whether or not the claims are amended following receipt of the search report or during any Chapter II procedure.

INTERNATIONAL SEARCH REPORT



International Application No EP 00/05677

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 CO7D A61K A61P

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, hearch terms used)

EPO-Internal, WPI Data, CHEM ABS Data

Category °	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Α	WO 92 01697 A (JANSSEN PHARMACEUTICA NV) 6 February 1992 (1992-02-06) page 21, line 9 - line 12; claim 1	1,7
Ρ,Χ	WO 99 44596 A (JANSSEN PHARMACEUTICA) 10 September 1999 (1999-09-10) page 11, line 18 - line 19; claim 1	1,7

Further documents are listed in the continuation of box C.	Patent family members are listed in annex.		
 Special categories of cited documents: "A" document defining the general state of the art which is not considered to be of particular relevance "E" earlier document but published on or after the international filling date "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "O" document referring to an oral disclosure, use, exhibition or other means "P" document published prior to the international filing date but later than the priority date claimed 	 "T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art. "8" document member of the same patent family 		
Date of the actual completion of the international search	Date of mailing of the international search report		
13 October 2000	27/10/2000		
Name and mailing address of the ISA	Authorized officer		
European Patent Office, P.B. 5818 Patentlaan 2 NL – 2280 HV Rijswijk Tel. (+31-70) 340-2040, Tx. 31 651 epo nl, Fax: (+31-70) 340-3016	Alfaro Faus, I		

INTERNATIONAL SEARCH REPORT



International	Application No
EP EP	00/05677

A. CLASSI IPC 7	A. CLASSIFICATION OF SUBJECT MATTER IPC 7 //(C07D471/04,221:00,221:00),(C07D471/04,235:00,221:00), (C07D491/056,319:00,221:00)						
	to International Patent Classification (IPC) or to both national classification (IPC) are to both national classif	assification and IPC					
	S SEAHCHED documentation searched (classification system followed by class	sification symbols)					
Documenta	ation searched other than minimum documentation to the extent	that such documents are included in the fields searched					
Electronic d	data base consulted during the international search (name of da	ata base and, where practical, search terms used)					
C. DOCUM	MENTS CONSIDERED TO BE RELEVANT						
Category °	Citation of document, with indication, where appropriate, of t	the relevant passages Relevant to clain	n No.				
Furth	ther documents are listed in the continuation of box C.	Patent family members are listed in annex.					
"A" docume consid "E" earlier of filing d "L" docume which itation "O" docume other r "P" docume later th	ategories of cited documents: lent defining the general state of the art which is not dered to be of particular relevance document but published on or after the international date ent which may throw doubts on priority claim(s) or is cited to establish the publication date of another on or other special reason (as specified) enent referring to an oral disclosure, use, exhibition or means lent published prior to the international filing date but than the priority date claimed actual completion of the international search	 "T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art. "&" document member of the same patent family Date of mailing of the international search report 					
<u> </u>	mailing address of the ISA	Authorized officer					
	European Patent Office, P.B. 5818 Patentlaan 2 NL – 2280 HV Rijswijk Tel. (+31-70) 340-2040, Tx. 31 651 epo nl, Fax: (+31-70) 340-3016	Alfaro Faus, I					

INTERNATIONAL SEARCH REPORT

on on patent family members

EP 00/05677

Patent document cited in search repo	rt	Publication date		Patent family member(s)	Publication date
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			ZA 	9105654 A	31-03-1993
WO 9944596	Α	10-09-1999	AU	3408999 A	20-09-1999

PATENT COOPERATION TREATY



RECEIVED

1 1 JUL 2001

From the

INTERNATIONAL PRELIMINARY EXAMINING AUTHORITY

QUAGHEBEUR, Luc JANSSEN PHARMACEUTICA N.V. Patent Department Turnhoutseweg 30 B-2340 Beerse **BELGIQUE**

Patent department

NOTIFICATION OF TRANSMITTAL OF THE INTERNATIONAL PRELIMINARY **EXAMINATION REPORT**

(PCT Rule 71.1)

Date of mailing (day/month/year)

09.07.2001

Applicant's or agent's file reference **JAB 1500-PCT**

International filing date (day/month/year)

IMPORTANT NOTIFICATION

International application No. PCT/EP00/05677

20/06/2000

Priority date (day/month/year)

28/06/1999

Applicant

JANSSEN PHARMACEUTICA N.V.

- 1. The applicant is hereby notified that this International Preliminary Examining Authority transmits herewith the international preliminary examination report and its annexes, if any, established on the international application.
- 2. A copy of the report and its annexes, if any, is being transmitted to the International Bureau for communication to all the elected Offices.
- 3. Where required by any of the elected Offices, the International Bureau will prepare an English translation of the report (but not of any annexes) and will transmit such translation to those Offices.

4. REMINDER

The applicant must enter the national phase before each elected Office by performing certain acts (filing translations and paying national fees) within 30 months from the priority date (or later in some Offices) (Article 39(1)) (see also the reminder sent by the International Bureau with Form PCT/IB/301).

Where a translation of the international application must be furnished to an elected Office, that translation must contain a translation of any annexes to the international preliminary examination report. It is the applicant's responsibility to prepare and furnish such translation directly to each elected Office concerned.

For further details on the applicable time limits and requirements of the elected Offices, see Volume II of the PCT Applicant's Guide.

Name and mailing address of the IPEA/

Authorized officer

European Patent Office D-80298 Munich Tel. +49 89 2399 - 0 Tx: 523656 epmu d THORNTON, J

Fax: +49 89 2399 - 4465

Tel.+49 89 2399-8072





PCT

INTERNATIONAL PRELIMINARY EXAMINATION REPORT

(PCT Article 36 and Rule 70)

		nt's file reference	FOR FURTHER ACT	rion		ation of Transmittal of International Examination Report (Form PCT/IPEA/416)
JAB 150			International filing data (de	w/month	(voar)	Priority date (day/month/year)
Internation	• •		International filing date (da 20/06/2000	ty/111Ortub	year, .	28/06/1999
PCT/EP						20,00,1000
Internation C07D40		nt Classification (IPC) o	r national classification and IPC			
Applicant			:			
JANSSE	=N PH	ARMACEUTICA N	.V.			
1. This and	interna is trans	ational-preliminary.ex smitted to the applica	amination_report_has_been_p nt according to Article 36.	<u>repa</u> red	by this Inte	rnational Preliminary Examining Authori
2. This	REPO	RT consists of a tota	of 9 sheets, including this	cover st	neet.	
	heen a	mended and are the	nied by ANNEXES, i.e. shee basis for this report and/or s n 607 of the Administrative I	sheets c	ontaining re	n, claims and/or drawings which have ectifications made before this Authority ne PCT).
Tho	co ann	exes consist of a tota	al of sheets.			
me	se arm	exes consist of a total	i or sneets.			•
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3. This	report	contains indications	relating to the following item	ıs:		
ı	ı 🛛	Basis of the report				
U		Priority				
11		Non-establishment	of opinion with regard to nov	velty, inv	entive step	and industrial applicability
١٧	<i>'</i> \square	Lack of unity of inve				
V	/ ⊠	Reasoned statement citations and explanations	nt under Article 35(2) with re nations suporting such state	gard to ment	novelty, inv	entive step or industrial applicability;
V	ı 🛛	Certain documents	s cited			
VI	ı 🗆	Certain defects in t	he international application			
VII	, (Certain observation	ns on the international applic	ation		
L						
Date of s	ubmissi	on of the demand		Date of	completion o	f this report
20/11/2	2000			09.07.2	001	
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INTERNATIONAL PRELIMINARY EXAMINATION REPORT

International application No. PCT/EP00/05677

	Bas	is of the report	
1. With regard to the elements of the international application (Replacement sheets which have been furnished the receiving Office in response to an invitation under Article 14 are referred to in this report as "originally file and are not annexed to this report since they do not contain amendments (Rules 70.16 and 70.17)): Description, pages:			
	1-83	3	as originally filed
	Cla	ims, No.:	
	1-15	5	as originally filed
		•••	
2.	With	n regard to the lang guage in which the	guage, all the elements marked above were available or-furnished to this Authority in-the international application was filed, unless otherwise indicated under this item.
	The	ese elements were	available or furnished to this Authority in the following language: , which is:
		the language of a	translation furnished for the purposes of the international search (under Rule 23.1(b)).
			ublication of the international application (under Rule 48.3(b)).
		the language of a 55.2 and/or 55.3).	translation furnished for the purposes of international preliminary examination (under Rule
3.	. Witl	h regard to any nu rnational prelimina	cleotide and/or amino acid sequence disclosed in the international application, the ry examination was carried out on the basis of the sequence listing:

filed together with the international application in computer readable form.
furnished subsequently to this Authority in written form.
furnished subsequently to this Authority in computer readable form.
The statement that the subsequently furnished written sequence listing does not go beyond the disclosure in the international application as filed has been furnished.
The statement that the information recorded in computer readable form is identical to the written sequence listing has been furnished.

4. The amendments have resulted in the cancellation of:

contained in the international application in written form.

	the description,	pages
	the claims,	Nos.:
П	the drawings	sheets

5. This report has been established as if (some of) the amendments had not been made, since they have been considered to go beyond the disclosure as filed (Rule 70.2(c)):

INTERNATIONAL PRELIMINARY EXAMINATION REPORT

International application No. PCT/EP00/05677

(Any replacement sheet containing such amendments must be referred to under item 1 and annexed to this report.)

		Topon.,			
6.	Add	itional observations, if ne	cessary	:	
111.	Nor	n-establishment of opini	on with	regard t	to novelty, inventive step and industrial applicability
	The	questions whether the cl	aimed ii	nvention a	appears to be novel, to involve an inventive step (to be non- not been examined in respect of:
		the entire international a	pplicatio	on.	
	×	claims Nos. 1-15 partiall	y.		
be	caus	se:			·
		the said international app not require an internation	plication nal preli	n, or the s minary ex	aid claims Nos. relate to the following subject matter which does xamination (specify):
		the description, claims o that no meaningful opinion	r drawir on could	ngs (<i>indic</i> d be form	ate particular elements below) or said claims Nos. are so unclea ed (specify):
		the claims, or said claim could be formed.	s Nos.	are so ina	adequately supported by the description that no meaningful opinion
	☒	no international search i	eport h	as been e	established for the said claims Nos. 1-15 partially.
2.	and	neaningful international pr Vor amino acid sequence ructions:	relimina listing t	ry examir o comply	nation cannot be carried out due to the failure of the nucleotide with the standard provided for in Annex C of the Administrative
		the written form has not	been fu	ırnished d	or does not comply with the standard.
		the computer readable f	orm has	s not bee	n furnished or does not comply with the standard.
V.	Rea	asoned statement under ations and explanations	r Article suppo	e 35(2) w rting suc	ith regard to novelty, inventive step or industrial applicability
1.	Sta	tement			
	No	velty (N)	Yes: No:	Claims Claims	1-15
	Inv	entive step (IS)	Yes: No:	Claims Claims	1-15
	Ind	ustrial applicability (IA)	Yes:	Claims	1-15

INTERNATIONAL PRELIMINARY EXAMINATION REPORT

International application No. PCT/EP00/05677

No: Claims

2. Citations and explanations see separate sheet

VI. Certain documents cited

1. Certain published documents (Rule 70.10)

and / or

2. Non-written disclosures (Rule 70.9)

see separate sheet

VIII. Certain observations on the international application

grande de la companya de la companya

The following observations on the clarity of the claims, description, and drawings or on the question whether the claims are fully supported by the description, are made: see separate sheet

EXAMINATION REPORT - SEPARATE SHEET

Reference is made to the following documents; they have all been cited in the 1. written opinion:

D1: WO 92 01697 A; 6 February 1992 (1992-02-06)

D2: WO 99 44596 A; 10 September 1999 (1999-09-10)

D3: EP 0005318 A (cited by the Applicant)

D4: EP 0297661 A (cited by the Applicant)

D5: EP 0307014 A (cited by the Applicant)

D6: WO 9831363 A

D7: EP 0747363 A

D8: WO 9855120 A

D9: WO-9810764 A

- D10 R.R. TIDWELL ET AL: 'Aromatic amidines: comparison of their ability to block respiratory syncytial virus induced cell fusion and to inhibit plasmin, urokinase, thrombin and trypsin', JOURNAL OF MEDICINAL CHEMISTRY, vol. 26, no. 2, pages 294 to 298
- D11 CHIBA ET AL.: 'Inhibitory effect of pyridobenzazoles on virus replication in vitro', BIOLOGICAL & PHARMACEUTICAL BULLETIN, vol. 18, no. 2, pages 1081 to 1083
- Non-establishment of opinion with regard to novelty, inventive st p and 2. industrial applicability (Reference to section III)

The Applicant's attention is drawn to the fact that claims or parts of claims, relating to inventions in respect of which no international search report has been established need not be the subject of an international preliminary examination (cf. Rule 66.1(e) PCT).

With respect to the international search report dated 13 October 2000, the international preliminary examination of the present application has been carried out for those parts relating to the compounds of formula (I) where radical Q is a 1-R2piperidyl-4-amino- or amino(cyclo)alkylamino-group as described in the examples of tables 3 to 13.

EXAMINATION REPORT - SEPARATE SHEET

3. R asoned stat ment under Rule 66.2(a)(ii) with regard to novelty, inv ntive step or industrial applicability; citations and explanations supporting such statement (Reference to section V)

3.1 Novelty

Documents D1 and D3-D5 disclose bicyclic heterocycles as anti-allergic compounds, whereas documents D6-D8 disclose benzimidazoles as anti-viral compounds. The novelty rendering feature of the present application with respect to the subject-matter of documents D1 and D3-D8 is the definition of substituent R¹ as a bicyclic heterocycle.

Document D9 discloses a piperazine-derivative, known as anti-allergic compound, as useful for treating anti-viral diseases. The novelty rendering feature of the present application in view of this document is the fused heterocyclic core-molecule.

Documents D10 and D11 disclose fused heterocyclic core-molecules as anti-viral compounds. The novelty rendering feature of the present application with respect to the subject-matter of these documents is inter alia the definition of radical Q.

Accordingly, the present application meets the requirements set forth in Articl 33(2) PCT.

3.2 Inventive step

Documents D6-D8 are considered as to represent the respective closest prior art for some of the claimed families of compounds. These documents disclose N1-C2-substituted benzimidazoles or pyridoimidazoles as anti-viral agents (see D6, abstract and page 2, lines 14-22; see D8, abstract and page 2, last paragraph; see D7 page 2, lines 24ff).

In view of these documents, the problem to be solved can be regarded as the provision of further fused 5,6-membered heterocyclic compounds with unexpected effects.

The Applicant's attention is drawn to the point, that in contrast to the description, which alleges anti-viral activities for the subject-matter of the present application, the claimed activity (see claim 7-9 and 15) is at present considered to be "therapeutically effective".

The solution to this problem provided by the present application consists in the analogisation of the N1- and C2-substituents of fused heterocyclic core-molecules or their bioisosteric analogues known from documents D6-D8.

However, the combined technical teaching of documents D1 and D3-D5 clearly indicates a fused heterocycle, substituted at least at positions N1 and C2 of the imidazole-moiety, as therapeutically active lead compound. As a consequence thereof, the man skilled in the art having knowledge of this technical teaching would not be surprised to obtain therapeutically active compounds by broadening the group of possible substituents represented by radical Q or R¹ according to the present application.

Moreover, underlying the principles of structure-activity relationship (SAR), it is stressed that for structural similar compounds a similar biological activity can be expected. As a consequence thereof, SAR allow to predict that for formal analogisations a pharmaceutical activity will be maintained.

Therefore, the feature of broadening the group of possible substituents represented by radical Q or R¹ starting from documents D6-D8 is merely one of several straightforward possibilities from which the skilled person would select, in accordance with circumstances, without any exercise of inventive skill, in order to solve the problem defined above.

As far as the alleged anti-viral activity mentioned in the description of the present application is concerned, attention is drawn to documents D10 and D11.

For the man skilled in the art, having knowledge of the combined technical teaching of

- document D10 (benzimidazoles as anti-viral lead compounds, see page 295, first column, last paragraph),
- document D11 (fused benzimidazoles as compounds exhibiting an inhibitory effect on RSV virus replication; see compounds 1-4, table 1, page 1082),

the feature of the provision of substituted benzimidazoles and their isosteric analogues is merely one of several straightforward possibilities from which the skilled person would select, in accordance with circumstances, without the exercise of inventive skill, in order to solve the problem posed.

Furthermore, the inhibiting effect of the well known histamine H1 receptor antagonist cetirizine on viral replication together with an inhibiting effect of RSV-induced cell

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modifications disclosed in document D9, page 2, lines 17-22 and page 3, lines 10-13, is a strong indication for a man skilled in the art having knowledge of the technical teaching of documents D7 and D8 (benzimidazole as lead compound for anti-viral agents) to examine, if known anti-allergic compounds bearing a benzimidazole-core exhibit also anti-viral properties, thereby arriving to the solution proposed by the present application.

It is noted, that SAR does not allow to predict whether the activity is better or worse. As a consequence thereof, an unexpected effect can be considered as an indication for inventive step. However, the applicant has not shown that the claimed compounds are likely to have any unexpected effects compared to those in the cited documents, in particular the nearest-possible compounds.

As far as the scope of the claims is concerned, attention is drawn to the point, that only such compounds can be claimed which represent a solution of the problem underlying the application in suit. The extent of a reasonable generalisation depends on the credibility that substantially all the alternatives claimed must be a solution to the problem. Extremely broad generalisations like e.g. the definition of radical G are in contradiction to the basis of even qualitative structure-activity- relationships. Taking into account the relevant state of the art and the common knowledge, it appears to be not predictable, that all alternatives would achieve the technical effect.

Accordingly, the present application does not fulfill the requirements set forth in Article 33(3) PCT.

Certain documents cited (Reference to section VI) 4.

The following document disclosing glycine transport inhibitors, which structures appear to overlap with the subject-matter of the present application, may become relevant when entering the national/regional phase.

Application No Patent No.

Publication date (day/month/year)

Filing date (day/month/year) Priority date (valid claim) (day/month/year)

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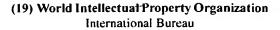
INTERNATIONAL PRELIMINARY

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- Certain observations in th international application (Reference to section VIII) 5.
- The terms "prodrug" and "metal complex" in claim 1 and 8 does not fulfill the 5.1 requirements of Article 6 PCT. The mere term "prodrug" is a functional expression attempting to define the subject-matter in terms of a desired property instead of indicating precisely the technical measures specifically designed to solve the problem. Functional terms will only be allowable if the solution is one which can directly be verified by tests or procedures adequately specified of known to the person skilled in the art and which verification does not need undue experimentation (cf. Guidelines C-III, 4.7). This requirement is presently not fulfilled.
- 5.2 Contrary to the requirements of Rule 5.1(a)(ii) PCT, documents D6-11 are not identified nor is the relevant background art disclosed therein mentioned.
- Attention is drawn to the fact that dependent claims are only admissible in the case of a allowable independent claim (cf. Rule 6.4 PCT).

CORRECTED VERSION







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(72) Inventors; and

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Frans, Eduard [BE/BE]: Janssen Pharmaceutica N.V..

Turnhoutseweg 30, B-2340 Beerse (BE). LACRAMPE, Jean, Fernand, Armand [FR/FR]; Janssen-Cilag S.A.,

rue Camille Desmoulins, TSA 91003, F-92787
 Issy-les-Moulineaux Cedex 9 (FR). GUILLEMONT,
 Jérôme, Emile, Georges [FR/FR]: Janssen-Cilag S.A.,

I, rue Camille Desmoulins, TSA 91003, F-92787 Issy-les-Moulineaux Cedex 9 (FR). VENET, Marc,

Gaston [FR/FR]; Janssen-Cilag S.A., 1, rue Camille Desmoulins, TSA 91003, F-92787 Issy-les-Moulineaux

Cedex 9 (FR). ANDRIES, Koenraad, Jozef, Lodewijk, Marcel [BE/BE]; Janssen Pharmaceutica N.V., Turnhout-

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(74) Agent: QUAGHEBEUR, Luc: Janssen Pharmaceutica N.V., Patent Department - 3547, Turnhoutseweg 30, B-2340 Beerse (BE).

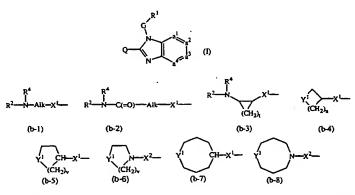
[Continued on next page]

- (71) Applicant (for all designated States except US): JANSSEN PHARMACEUTICA N.V. [BE/BE]; Turn-houtseweg 30, B-2340 Beerse (BE).
- [Conting

seweg 30, B-2340 Beerse (BE).

(75) Inventors/Applicants (for US only):

(54) Title: RESPIRATORY SYNCYTIAL VIRUS REPLICATION INHIBITORS



(57) Abstract: The present invention concerns compounds of formula (I), prodrugs, N-oxides, addition salts, quaternary amines, metal complexes and stereochemically isomeric forms thereof wherein -a1=a2-a3=a4- represents a radical of formula -CH=CH-CH=CH-; -N=CH-CH=CH-; -CH=N-CH=CH-; -CH=CH-N=CH-; -CH=CH-CH=N-; wherein each hydrogen atom may optionally be substituted; Q is a radical of formulae (b-1), (b-2), (b-3), (b-4), (b-5), (b-6), (b-7) and (b-8), wherein Alk is C₁₋₆alkanediyl; Y¹ is a bivalent radical of formula -NR²- or -CH(NR²R⁴)-; X¹ is NR⁴, S, S(=O), S(=O)₂, O, CH₂, C(=O), CH(=CH₂), CH(OH), CH(CH₃), CH(OCH₃), CH(SCH₃), CH(NR^{5a}R^{5b}), CH₂-NR⁴ or NR⁴-CH₂; X² is a direct bond, CH₂, C(=O), NR⁴, C₁₋₄alkyl-NR⁴, NR⁴-C₁₋₄alkyl, t is 2 to 5; u is 1 to 5; v is 2 or 3; and whereby each hydrogen in Alk and in (b-3), (b-4), (b-5), (b-6), (b-7) and (b-8), may optionally be replaced by R3; provided that when R3 is hydroxy or C1.6alkyloxy, then R3 cannot replace a hydrogen atom in the α position relative to a nitrogen atom; G is a direct bond or optionally substituted C_{1-10} alkanediyl; R^1 is an optionally substituted bicyclic heterocycle; R2 is hydrogen, formyl, C16alkylcarbonyl, Hetcarbonyl, pyrrolidinyl, piperidinyl, homopiperidinyl, C3.7cycloalkyl or C1.10alkyl substituted with N(R6)2 and optionally with another substituent; R3 is hydrogen, hydroxy, C₁₋₆alkyl, C₁₋₆alkyloxy, arylC₁₋₆alkyl or arylC₁₋₆alkyloxy, R⁴ is hydrogen, C₁₋₆alkyl or arylC₁₋₆alkyl; R^{5a}, R^{5b}, R^{5c} and R^{5d} are hydrogen or C₁₋₆alkyl; or R^{5a} and R^{5b}, or R^{5c} and R^{5d} taken together from a bivalent radical of formula -(CH₂)_s- wherein S is 4 or 5; R6 is hydrogen, C1-4alkyl, formyl, hydroxyC1-6alkyl, C1-6alkylcarbonyl or C1-6alkyloxycarbonyl; aryl is optionally substituted phenyl; Het is pyridyl, pyrimidinyl, pyryzinyl, pyridazinyl; as respiratory syncytial virus replication inhibitors: their preparation, compositions containing them and their use as a medicine.









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RESPIRATORY SYNCYTIAL VIRUS REPLICATION INHIBITORS

- The present invention is concerned with benzimidazoles and imidazopyridines having antiviral activity, in particular, they have an inhibitory activity on the replication of the respiratory syncytial virus. It further concerns their preparation and compositions comprising them, as well as their use as a medicine.
- Human RSV or Respiratory Syncytial Virus is a large RNA virus, member of the family of Paramyxoviridae, subfamily pneumovirinae together with bovine RSV virus. Human RSV is responsible for a spectrum of respiratory tract diseases in people of all ages throughout the world. It is the major cause of lower respiratory tract illness during infancy and childhood. Over half of all infants encounter RSV in their first year of life, and almost all within their first two years. The infection in young children can cause lung damage that persists for years and may contribute to chronic lung disease in later life (chronic wheezing, asthma). Older children and adults often suffer from a (bad) common cold upon RSV infection. In old age, susceptibility again increases, and RSV has been implicated in a number of outbreaks of pneumonia in the aged resulting in significant mortality.
 - Infection with a virus from a given subgroup does not protect against a subsequent infection with an RSV isolate from the same subgroup in the following winter season. Re-infection with RSV is thus common, despite the existence of only two subtypes, A and B.
 - Today only three drugs have been approved for use against RSV infection. Ribavirin, a nucleoside analogue, provides an aerosol treatment for serious RSV infection in hospitalized children. The aerosol route of administration, the toxicity (risk of teratogenicity), the cost and the highly variable efficacy limit its use. The other two drugs, RespiGam® and palivizumab, polyclonal and monoclonal antibody immunostimulants, are intended to be used in a preventive way.
- Other attempts to develop a safe and effective RSV vaccine have all met with failure
 thus far. Inactivated vaccines failed to protect against disease, and in fact in some cases
 enhanced disease during subsequent infection. Life attenuated vaccines have been tried
 with limited success. Clearly there is a need for an efficacious non-toxic and easy to
 administer drug against RSV replication.

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EP-A-0,005,318, EP-A-0,099,139, EP-A-0,145,037, EP-A-0,144,101, EP-A-0,151,826, EP-A-0,151,824, EP-A-0,232,937, EP-A-0,295,742, EP 0,297,661, EP-A-0,307,014, WO 92 01697 describe benzimidazole and imidazopyridine substituted piperidine and piperazine derivatives as antihistaminics, antiallergics or serotonine antagonists.

Thus, the present invention concerns the compounds of formula (I)

$$Q = \begin{bmatrix} R^1 & & & \\ &$$

their prodrugs, N-oxides, addition salts, quaternary amines, metal complexes and stereochemically isomeric forms wherein

-a¹=a²-a³=a⁴- represents a bivalent radical of formula

-CH=CH-CH=CH-

(a-1);

-N=CH-CH=CH-

(a-2);

-CH=N-CH=CH-

(a-3);

-CH=CH-N=CH-

(a-4); or

-CH=CH-CH=N-

(a-5);

wherein each hydrogen atom in the radicals (a-1), (a-2), (a-3), (a-4) and (a-5) may optionally be replaced by halo, C₁₋₆alkyl, nitro, amino, hydroxy,

C₁₋₆alkyloxy, polyhaloC₁₋₆alkyl, carboxyl, aminoC₁₋₆alkyl, mono- or di(C₁₋₄alkyl)aminoC₁₋₆alkyl, C₁₋₆alkyloxycarbonyl, hydroxyC₁₋₆alkyl, or a radical of formula

wherein =Z is =O, =CH-C(=O)-NR 5a R 5b , =CH2, =CH-C $_{1\text{-}6}$ alkyl, =N-OH or =N-O-C $_{1\text{-}6}$ alkyl;

Q is a radical of formula

$$Y_{(CH_2)_v}^1$$
 $Y_{(CH_2)_v}^1$ $Y_{($

wherein Alk is C₁₋₆alkanediyl;

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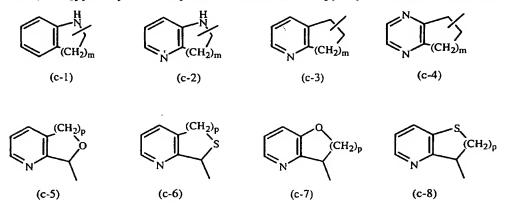
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Y¹ is a bivalent radical of formula $-NR^2$ - or $-CH(NR^2R^4)$ -; X¹ is NR^4 , S, S(=O), S(=O)₂, O, CH₂, C(=O), C(=CH₂), CH(OH), CH(CH₃), CH(OCH₃), CH(SCH₃), CH(NR^{5a}R^{5b}), CH₂-NR⁴ or NR⁴-CH₂; X² is a direct bond, CH₂, C(=O), NR⁴, C₁₋₄alkyl-NR⁴, NR⁴-C₁₋₄alkyl; t is 2, 3, 4 or 5; u is 1, 2, 3, 4 or 5; v is 2 or 3; and

whereby each hydrogen atom in Alk and the carbocycles and the heterocycles defined in radicals (b-3), (b-4), (b-5), (b-6), (b-7) and (b-8) may optionally be replaced by R³; with the proviso that when R³ is hydroxy or C₁₋₆alkyloxy, then R³ can not replace a hydrogen atom in the α position relative to a nitrogen atom;

G is a direct bond or C_{1-10} alkanediyl optionally substituted with one, two or three substituents selected from hydroxy, C_{1-6} alkyloxy, aryl C_{1-6} alkyloxy, C_{1-6} alkylthio, aryl C_{1-6} alkylthio, arylcarbonyl, HO(-CH₂-CH₂-O)_n-, C_{1-6} alkyloxy(-CH₂-CH₂-O)_n-, arryl C_{1-6} alkyloxy(-CH₂-CH₂-O)_n-, amino, mono-or di(C_{1-6} alkyloxycarbonylamino and aryl;

R¹ is a bicyclic heterocycle selected from quinolinyl, quinoxalinyl, benzofuranyl, benzothienyl, benzimidazolyl, benzoxazolyl, benzthiazolyl, pyridopyridyl, naphthiridinyl, 1*H*-imidazo[4,5-b]pyridinyl, 3*H*-imidazo[4,5-b]pyridinyl, imidazo[1,2-a]pyridinyl, 2,3-dihydro-1,4-dioxino[2,3-b]pyridyl or a radical of formula



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and said bicyclic heterocycles may optionally be substituted in either of the two cycles with 1 or where possible more, such as 2, 3 or 4, substituents selected from halo, hydroxy, amino, cyano, carboxy, C₁₋₆alkyl, C₁₋₆alkyloxy, C₁₋₆alkylthio, C₁₋₆alkyloxy-C₁₋₆alkyl, arylC₁₋₆alkyl, arylC₁₋₆alkyloxy, hydroxyC₁₋₆alkyl, mono-or di(C₁₋₆alkyl)-amino, mono-or di(C₁₋₆alkyl)aminoC₁₋₆alkyl, polyhaloC₁₋₆alkyl, C₁₋₆alkylcarbonylamino, C₁₋₆alkyl-SO₂-NR^{5c}-, aryl-SO₂-NR^{5c}-, C₁₋₆alkyloxycarbonyl, -C(=O)-NR^{5c}R^{5d}, HO(-CH₂-CH₂-O)_n-, halo(-CH₂-CH₂-O)_n-, C₁₋₆alkyloxy(-CH₂-CH₂-O)_n-, arylC₁₋₆alkyloxy(-CH₂-CH₂-O)_n-; each n independently is 1, 2, 3 or 4;

each n independently is 1, 2, 3 or 4;

10 each m independently is 1 or 2;
each p independently is 1 or 2;
each R² independently is hydrogen, formyl, C₁₋₆alkylcarbonyl, Hetcarbonyl, pyrrolidinyl, piperidinyl, homopiperidinyl, C₃₋₇cycloalkyl substituted with N

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pyrrolidinyl, piperidinyl, homopiperidinyl, C_{3-7} cycloalkyl substituted with $N(R^6)_2$, or C_{1-10} alkyl substituted with $N(R^6)_2$ and optionally with a second, third or fourth substituent selected from amino, hydroxy, C_{3-7} cycloalkyl, C_{2-5} alkanediyl, piperidinyl,

substituent selected from amino, hydroxy, C₃₋₇cycloalkyl, C₂₋₅alkanediyl, piperidinyl mono-or di(C₁₋₆alkyl)amino, C₁₋₆alkyloxycarbonylamino, aryl and aryloxy; R³ is hydrogen, hydroxy, C₁₋₆alkyl, C₁₋₆alkyloxy, arylC₁₋₆alkyl or arylC₁₋₆alkyloxy; R⁴ is hydrogen, C₁₋₆alkyl or arylC₁₋₆alkyl;

 $R^{5a},\,R^{5b},\,R^{5c}$ and R^{5d} each independently are hydrogen or $C_{1\text{-}6}$ alkyl; or

20 R^{5a} and R^{5b}, or R^{5c} and R^{5d} taken together form a bivalent radical of formula -(CH₂)_s-wherein s is 4 or 5;

 R^6 is hydrogen, $C_{1\text{-}4}$ alkyl, formyl, hydroxy $C_{1\text{-}6}$ alkyl, $C_{1\text{-}6}$ alkylcarbonyl or $C_{1\text{-}6}$ alkyloxycarbonyl;

aryl is phenyl or phenyl substituted with 1 or more, such as 2, 3 or 4, substituents selected from halo, hydroxy, C₁₋₆alkyl, hydroxyC₁₋₆alkyl, polyhaloC₁₋₆alkyl, and C₁₋₆alkyloxy;

Het is pyridyl, pyrimidinyl, pyrazinyl, pyridazinyl.

The term prodrug as used throughout this text means the pharmacologically acceptable derivatives, e.g. esters and amides, such that the resulting biotransformation product of the derivative is the active drug as defined in the compounds of formula (I). The reference by Goodman and Gilman (The Pharmacological Basis of Therapeutics, 8th ed., McGraw-Hill, Int. Ed. 1992, "Biotransformation of Drugs", p. 13-15) describing prodrugs generally, is hereby incorporated.

As used herein C_{1-3} alkyl as a group or part of a group defines straight or branched chain saturated hydrocarbon radicals having from 1 to 3 carbon atoms such as methyl, ethyl,

propyl, 1-methylethyl and the like; C14alkyl as a group or part of a group defines straight or branched chain saturated hydrocarbon radicals having from 1 to 4 carbon atoms such as the group defined for C_{1.3}alkyl and butyl and the like; C_{2.4}alkyl as a group or part of a group defines straight or branched chain saturated hydrocarbon 5 radicals having from 2 to 4 carbon atoms such as ethyl, propyl, 1-methylethyl, butyl and the like; C₁₋₆alkyl as a group or part of a group defines straight or branched chain saturated hydrocarbon radicals having from 1 to 6 carbon atoms such as the groups defined for C₁₋₄alkyl and pentyl, hexyl, 2-methylbutyl and the like; C₁₋₉alkyl as a group or part of a group defines straight or branched chain saturated hydrocarbon radicals 10 having from 1 to 9 carbon atoms such as the groups defined for C_{1.6}alkyl and heptyl, octyl, nonyl, 2-methylhexyl, 2-methylheptyl and the like; C_{1.10}alkyl as a group or part of a group defines straight or branched chain saturated hydrocarbon radicals having from 1 to 10 carbon atoms such as the groups defined for C_{1.9}alkyl and decyl, 2-methylnonyl and the like. C3.7cycloalkyl is generic to cyclopropyl, cyclobutyl, cyclopentyl, 15 cyclohexyl and cycloheptyl; C₂₋₅alkanediyl defines bivalent straight and branched chain saturated hydrocarbon radicals having from 2 to 5 carbon atoms such as, for example, 1,2-ethanediyl, 1,3-propanediyl, 1,4-butanediyl, 1,2-propanediyl, 2,3-butanediyl, 1,5pentanediyl and the like, C₂₋₅alkanediyl is substituted on C₁₋₁₀alkyl as provided for in the definition of R², it is meant to be substituted on one carbon atom thus forming a 20 spiro moiety; C₁₋₄alkanediyl defines bivalent straight and branched chain saturated hydrocarbon radicals having from 1 to 4 carbon atoms such as, for example, methylene, 1,2-ethanediyl, 1,3-propanediyl, 1,4-butanediyl and the like; C_{1.6}alkanediyl is meant to include C₁₋₄alkanediyl and the higher homologues thereof having from 5 to 6 carbon atoms such as, for example, 1,5-pentanediyl, 1,6-hexanediyl and the like; 25 C₁₋₁₀alkanediyl is meant to include C₁₋₆alkanediyl and the higher homologues thereof having from 7 to 10 carbon atoms such as, for example, 1,7-heptanediyl, 1,8-octanediyl, 1,9-nonanediyl, 1,10-decanediyl and the like.

As used herein before, the term (=O) forms a carbonyl moiety when attached to a carbon atom, a sulfoxide moiety when attached to a sulfur atom and a sulfonyl moiety when two of said terms are attached to a sulfur atom. The term (=N-OH) forms a hydroxylimine moiety when attached to a carbon atom.

The term halo is generic to fluoro, chloro, bromo and iodo. As used in the foregoing and hereinafter, polyhaloC₁₋₆alkyl as a group or part of a group is defined as mono- or polyhalosubstituted C₁₋₆alkyl, in particular methyl with one or more fluoro atoms, for example, difluoromethyl or trifluoromethyl. In case more than one halogen atoms are

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attached to an alkyl group within the definition of polyhaloC₁₋₄alkyl, they may be the same or different.

As described hereinabove, R¹ defines a bicyclic heterocycle which may optionally be substituted. The substituents may be divided over both rings or they may be attached to one and the same ring.

When any variable (e.g. aryl, R²,R³, R⁴, R^{5a}, R^{5b} etc.) occurs more than one time in any constituent, each definition is independent.

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It will be appreciated that some of the compounds of formula (I) and their prodrugs, N-oxides, addition salts, quaternary amines, metal complexes and stereochemically isomeric forms may contain one or more centers of chirality and exist as stereochemically isomeric forms.

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The term "stereochemically isomeric forms" as used hereinbefore defines all the possible stereoisomeric forms which the compounds of formula (I), and their prodrugs, N-oxides, addition salts, quaternary amines or physiologically functional derivatives may possess. Unless otherwise mentioned or indicated, the chemical designation of compounds denotes the mixture of all possible stereochemically isomeric forms, said mixtures containing all diastereomers and enantiomers of the basic molecular structure as well as each of the individual isomeric forms of formula (I) and their prodrugs, N-oxides, salts, solvates or quaternary amines substantially free, i.e. associated with less than 10%, preferably less than 5%, in particular less than 2% and most preferably less than 1% of the other isomers. Stereochemically isomeric forms of the compounds of formula (I) are obviously intended to be embraced within the scope of this invention. As used hereinafter the terms trans or cis are well-known by the person skilled in the art.

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The pharmaceutically acceptable acid and base addition salts as mentioned hereinabove are meant to comprise the therapeutically active non-toxic acid and base addition salt forms which the compounds of formula (I) are able to form. The pharmaceutically

counterion is pharmaceutically acceptable. However, salts of acids and bases which are non-pharmaceutically acceptable may also find use, for example, in the preparation or purification of a pharmaceutically acceptable compound. All salts, whether pharmaceutically acceptable or not are included within the ambit of the present invention.

For therapeutic use, salts of the compounds of formula (I) are those wherein the

acceptable acid addition salts can conveniently be obtained by treating the base form with such appropriate acid. Appropriate acids comprise, for example, inorganic acids such as hydrohalic acids, e.g. hydrochloric or hydrobromic acid, sulfuric, nitric, phosphoric and the like acids; or organic acids such as, for example, acetic, propanoic, hydroxyacetic, lactic, pyruvic, oxalic (i.e. ethanedioic), malonic, succinic (i.e. butanedioic acid), maleic, fumaric, malic (i.e. hydroxybutanedioic acid), tartaric, citric, methanesulfonic, ethanesulfonic, benzenesulfonic, p-toluenesulfonic, cyclamic, salicylic, p-aminosalicylic, pamoic and the like acids.

10 Conversely said salt forms can be converted by treatment with an appropriate base into the free base form.

The compounds of formula (I) containing an acidic proton may also be converted into their non-toxic metal or amine addition salt forms by treatment with appropriate organic and inorganic bases. Appropriate base salt forms comprise, for example, the ammonium salts, the alkali and earth alkaline metal salts, e.g. the lithium, sodium, potassium, magnesium, calcium salts and the like, salts with organic bases, e.g. the benzathine, *N*-methyl-D-glucamine, hydrabamine salts, and salts with amino acids such as, for example, arginine, lysine and the like.

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The term addition salt as used hereinabove also comprises the solvates which the compounds of formula (I) as well as the salts thereof, are able to form. Such solvates are for example hydrates, alcoholates and the like.

The term "quaternary amine" as used hereinbefore defines the quaternary ammonium salts which the compounds of formula (I) are able to form by reaction between a basic nitrogen of a compound of formula (I) and an appropriate quaternizing agent, such as, for example, an optionally substituted alkylhalide, arylhalide or arylalkylhalide, e.g. methyliodide or benzyliodide. Other reactants with good leaving groups may also be used, such as alkyl trifluoromethanesulfonates, alkyl methanesulfonates, and alkyl ptoluenesulfonates. A quaternary amine has a positively charged nitrogen. Pharmaceutically acceptable counterions include chloro, bromo, iodo, trifluoroacetate and acetate. The counterion of choice can be introduced using ion exchange resins.

It will be appreciated that the compounds of formula (I) may have metal binding, chelating, complexating properties and therefore may exist as metal complexes or metal chelates. Such metalated derivatives of the compounds of formula (I) are intended to be included within the scope of the present invention.

Some of the compounds of formula (I) may also exist in their tautomeric form. Such forms although not explicitly indicated in the above formula are intended to be included within the scope of the present invention.

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A special group of com pounds are those compounds of formula (I) wherein one or more of the following restrictions apply:

- Q is a radical of formula (b-1), (b-3), (b-4), (b-5), (b-6), (b-7) or (b-8);
- X² is a direct bond, CH₂ or C(=O);
- R¹ is a bicyclic heterocycle selected from quinolinyl, quinoxalinyl, benzofuranyl, benzothienyl, benzimidazolyl, benzoxazolyl, benzthiazolyl, pyridopyridyl, naphthiridinyl, 1*H*-imidazo[4,5-b]pyridinyl, 3*H*-imidazo[4,5-b]pyridinyl, imidazo[1,2-a]pyridinyl, or a radical of formula

$$(CH_{2})_{m} \qquad (CH_{2})_{m} \qquad (CH_$$

and said bicyclic heterocycles may optionally be substituted in either of the two cycles with 1 or where possible more, such as 2, 3 or 4, substituents selected from halo, hydroxy, amino, cyano, carboxy, C₁₋₆alkyl, C₁₋₆alkyloxy, C₁₋₆alkylthio, C₁₋₆alkyloxy-C₁₋₆alkyl, arylC₁₋₆alkyl, arylC₁₋₆alkyloxy, hydroxyC₁₋₆alkyl, mono-or di(C₁₋₆alkyl)-amino, mono-or di(C₁₋₆alkyl)aminoC₁₋₆alkyl, polyhaloC₁₋₆alkyl, C₁₋₆alkylcarbonylamino,

C₁₋₆alkyl-SO₂-NR^{5c}-, aryl-SO₂-NR^{5c}-, C₁₋₆alkyloxycarbonyl, -C(=O)-NR^{5c}R^{5d}, HO(-CH₂-CH₂-O)_n-, halo(-CH₂-CH₂-O)_n-, C₁₋₆alkyloxy(-CH₂-CH₂-O)_n-, arylC₁₋₆alkyloxy(-CH₂-CH₂-O)_n-, arylC₁₋₆alkyloxy(-CH₂-CH₂-O)_n-; each n independently is 1, 2, 3 or 4; each m independently is 1 or 2;

each p independently is 1 or 2;

- each R^2 independently is hydrogen, pyrrolidinyl, piperidinyl, homopiperidinyl, C_{3-7} cycloalkyl substituted with NHR⁶, or C_{1-10} alkyl substituted with NHR⁶ and optionally with a second, third or fourth substituent selected from amino, hydroxy,

 C_{3-7} cycloalkyl, C_{2-5} alkanediyl, piperidinyl, mono-or di(C_{1-6} alkyl)amino, C_{1-6} alkyloxycarbonylamino, aryl and aryloxy;

- R³ is hydrogen, hydroxy, C₁₋₆alkyl, C₁₋₆alkyloxy or arylC₁₋₆alkyl;
- R^6 is hydrogen, C_{1-6} alkyl, formyl, C_{1-6} alkylcarbonyl or C_{1-6} alkyloxycarbonyl.

Another special group of compounds are those compounds wherein $-a^1=a^2-a^3=a^4$ is a radical of formula (a-1), (a-2) or (a-3).

Yet another special group of compounds are those compounds wherein Q is a radical of formula (b-5) wherein v is 2, and Y¹ is -NR²-.

Also interesting compounds are those compounds wherein R^2 is C_{1-10} alkyl substituted with NHR⁶.

- Other interesting compounds are those compounds wherein G is a direct bond or C₁₋₁₀alkanediyl optionally substituted with one two or three substituents selected from hydroxy, C₁₋₆alkyloxy, arylC₁₋₆alkyloxy, HO(-CH₂-CH₂-O)_n-, C₁₋₆alkyloxy(-CH₂-CH₂-O)_n-, arylC₁₋₆alkyloxy(-CH₂-CH₂-O)_n-.
- 20 Preferred compounds are:
 - (\pm) -N-[1-(2-aminoethyl)-4-piperidinyl]-4-methyl-1-[1-(8-quinolinyl)ethyl]-IH-benzimidazol-2-amine monohydrate;
 - (\pm)-*N*-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-1-(2-bromo-5,6,7,8-tetrahydro-8-quinolinyl)-*1H*-benzimidazol-2-amine trihydrochloride trihydrate;
- 25 (±)-N-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-1-[(2-ethoxyethoxy)-8-quinolinyl-methyl]-4-methyl-1H-benzimidazol-2-amine;
 - (\pm)-*N*-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-1-(2-chloro-5,6,7,8-tetrahydro-5-quinoxalinyl)-*1H*-benzimidazol-2-amine trihydrochloride trihydrate;
 - $(\pm)-N-[1-(2-a\min o-3-methylb\mu tyl)-4-piperidinyl]-1-[(1-methyl-1H-benzimidazol-4-methylbutyl)-4-piperidinyl]-1-[(1-methylbutyl)-1-[(1-methylbutyl)-1-[(1-methylbutyl)-1-[(1-methylbutyl)-1-[(1-methylbutyl)-1-[(1-methylbutyl)-1-[(1-methylbutyl)-1-[(1-methylbutyl)-1-[(1-methylbutyl)-1-[(1-methylbutyl)-1-[(1-methylb$
- 30 yl)methyl]-1H-benzimidazol-2-amine;
 - (±)-N-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-1-(ethoxy-8-quinolinylmethyl)-1H-benzimidazol-2-amine;
 - (±)-*N*-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-4-methyl-1-(5,6,7,8-tetrahydro-5-quinoxalinyl)-*1H*-benzimidazol-2-amine;
- 35 (±)-*N*-[1-(2-aminoethyl)-4-piperidinyl]-7-methyl-3-(8-quinolinylmethyl)-3*H*-imidazo-[4,5-b]pyridin-2-amine tetrahydrochloride trihydrate;
 - $\label{eq:continuity} (\pm)-N-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-7-methyl-3-(8-quinolinylmethyl)-3H-imidazo[4,5-b]pyridin-2-amine tetrahydrochloride monohydrate;$

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- (±)-N-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-1-(8-quinolinylmethyl)-1H-imidazo[4,5-c]pyridin-2-amine trihydrochloride dihydrate;
- *N*-[1-(2-aminoethyl)-4-piperidinyl]-4-methyl-1-(8-quinolinylmethyl)-*1H*-benzimidazol-2-amine;
- 5 *N*-[1-(8-quinolinylmethyl)-1*H*-benzimidazol-2-yl]-1,3-propanediamine trihydrochloride monohydrate;
 - (\pm)-N-[1-(2-aminoethyl)-4-piperidinyl]-1-[(2-ethoxyethoxy)-8-quinolinylmethyl]-IH-benzimidazol-2-amine trihydrochloride dihydrate;
 - $(\pm)-N-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-1-(8-quinolinylmethyl)-1H-imidazo-1-(8-quinolinylmethyl)-1H-imidazo-1-(8-quinolinylmethyl)-1H-imidazo-1-(8-quinolinylmethyl)-1-(8-quinolinylmethyl)-1H-imidazo-1-(8-quinolinylmethyl)-1-(8-quinolinylmethylmethyl)-1-(8-quinolinylmethylmet$
- 10 [4,5-b]pyridine-2-amine trihydrochloride dihydrate;
 - (\pm)-N-[1-[1-(aminomethyl)-2-methylpropyl]-4-piperidinyl]-1-[(2-ethoxyethoxy)-8-quinolinylmethyl]-IH-benzimidazol-2-amine;
 - (\pm)-*N*-[1-(2-aminoethyl)-4-piperidinyl]-3-(2-quinolinylmethyl)-3*H*-imidazo-[4,5-b]pyridin-2-amine trihydrochloride trihydrate;
- (±)-N-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-1-(1-isoquinolinylmethyl)-1H-benzimidazol-2-amine trihydrochloride trihydrate;
 N-[1-(2-aminoethyl)-4-piperidinyl]-1-(5,6,7,8-tetrahydro-5-quinoxalinyl)-1H-benzimidazol-2-amine trihydrochloride trihydrate;
 the prodrugs, the N-oxides, the addition salts, the quaternary amines, the metal
- 20 complexes and the stereochemically isomeric forms thereof.

Most preferred compounds are:

- (±)-*N*-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-3-(quinolinylmethyl)-3*H*-imidazo[4,5-b]pyridin-2-amine;
- 25 (±)-*N*-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-4-methyl-1-(8-quinolinylmethyl)-1*H*-benzimidazol-2-amine;
 - (\pm)-N-[1-(2-aminoethyl)-4-piperidinyl]-1-(2-chloro-5,6,7,8-tetrahydro-5-quinoxalinyl)-4-methyl-IH-benzimidazol-2-amine trihydrochloride.trihydrate;
 - (±)-N-[1-(2-aminoethyl)-4-piperidinyl]-1-(5,6,7,8-tetrahydro-2,3-dimethyl-5-
- 30 quinoxalinyl)-1H-benzimidazol-2-amine trihydrochloride trihydrate;
 - (\pm)-N-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-1-[(2-ethoxyethoxy)-8-quinolinyl-methyl]-IH-benzimidazol-2-amine;
 - (±)-*N*-[1-(2-amino-3-methylbutyl)-4-piperidinyl-1-(3-chloro-5,6,7,8-tetrahydro-5-quinoxalinyl)-*1H*-benzimidazol-2-amine trihydrochloride monohydrate;
- 35 (±)-*N*-[1-(2-aminoethyl)-4-piperidinyl]-1-(3-chloro-5,6,7,8-tetrahydro-5-quinoxalinyl)-4-methyl-*1H*-benzimidazol-2-amine trihydrochloride dihydrate;

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(±)-N-[1-(2-aminoethyl)-4-piperidinyl]-1-[(2-ethoxyethoxy)-8-quinolinylmethyl]-4methyl-1H-benzimidazol-2-amine monohydrate;

- (±)-N-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-3-(8-quinolinylmethyl)-3H-imidazo-[4,5-c]pyridin-2-amine trihydrochloride tetrahydrate;
- $(\pm)-N-[1-(2-aminoethyl)-4-piperidinyl]-3-(8-quinolinylmethyl)-3H-imidazo[4,5-b]$ pyridin-2-amine;
 - (\pm) -N-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-4-methyl-1-[(1-methyl-1Hbenzimidazol-4-yl)methyl]-1H-benzimidazol-2-amine;
- (±)-N-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-1-(2-chloro-5,6,7,8-tetrahydro-5-10 quinoxalinyl)-4-methyl-1H-benzimidazol-2-amine; the prodrugs, the N-oxides, the addition salts, the quaternary amines, the metal

complexes and the stereochemically isomeric forms thereof.

In general, compounds of formula (I) can be prepared by reacting an intermediate of 15 formula (II-a) or (II-b), wherein P represents a protecting group, such as, for example C₁₋₄alkyloxycarbonyl, or those protecting groups mentioned in Chapter 7 of 'Protective Groups in Organic Synthesis' by T Greene and P. Wuyts (John Wiley & Sons Inc., 1991), with an intermediate of formula (III), wherein W_1 is a suitable leaving group, such as a halo atom, e.g. chloro, bromo, in the presence of a suitable base, such as, e.g. 20 sodium hydride. Said reaction can be performed in a reaction-inert solvent, such as N,N-dimethylformamide.

$$Q = \begin{pmatrix} R^{1} & R^{1} & G - W_{1} \\ N & a^{1} & a^{2} \\ N & a^{4} & a^{3} \end{pmatrix}$$

$$(II-a) \qquad \qquad (III-b) \qquad (III) \qquad (III)$$

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Compounds of formula (I) wherein, in the definition of O, R² or at least one R⁶ substituent is hydrogen, said Q being represented by H-Q₁, and said compounds being represented by formula (I-a), can be prepared by deprotecting an intermediate of formula (IV) wherein P represents a protecting group, for example C₁. 4alkyloxycarbonyl, benzyl, or those protecting groups mentioned in Chapter 7 of

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'Protective Groups in Organic Synthesis' by T Greene and P. Wuyts (John Wiley & Sons Inc., 1991).

$$P = Q_1 = \begin{pmatrix} R^1 \\ Q \\ N \end{pmatrix} \begin{pmatrix} A^1 \\ A^2 \\ A^3 \end{pmatrix}$$

$$(IV)$$

$$H = Q_1 + \begin{pmatrix} R^1 \\ N \\ A^2 \\ A^3 \end{pmatrix}$$

$$(I-a)$$

When P represents, for example, C_{1.4}alkyloxycarbonyl, said deprotection reaction can be performed by, for example, acidic hydrolysis in the presence of a suitable acid, such 5 as hydrobromic, hydrochloric, sulfuric, acetic, or trifluoroacetic acid or a mixture of said acids, or by alkaline hydrolysis in the presence of a suitable base, such as, for example potassium hydroxide, in a suitable solvent such as water, alcohol, a mixture of water-alcohol, methylene chloride. Suitable alcohols are methanol, ethanol, 10 2-propanol, 1-butanol and the like. In order to enhance the rate of the reaction, it is advantageous to heat the reaction mixture, in particular up to the reflux temperature. Alternatively, when P represents, for example, benzyl, the deprotection reaction can be performed by catalytic hydrogenation in the presence of hydrogen and an appropriate catalyst in a reaction-inert solvent. A suitable catalyst in the above reaction is, for 15 example, platinum-on-charcoal, palladium-on-charcoal, and the like. An appropriate reaction-inert solvent for said reaction is, for example, an alcohol, e.g. methanol, ethanol, 2-propanol and the like, an ester, e.g. ethylacetate and the like, an acid, e.g. acetic acid and the like.

20 The catalytic hydrogenation reaction described above can also be used to prepare a compound of formula (I-a) by deprotecting and reducing an intermediate of formula (IV) wherein Q₁ comprises an unsaturated bond, said Q₁ being represented by Q_{1a}(CH=CH), and said intermediate being represented by formula (IV-a).

P—Q_{1a}(CH=CH)
$$\stackrel{a_1}{\longrightarrow} \stackrel{a_2}{\longrightarrow} \stackrel{a_3}{\longrightarrow} \stackrel{a_1}{\longrightarrow} \stackrel{a_1}{\longrightarrow} \stackrel{a_2}{\longrightarrow} \stackrel{a_2}{\longrightarrow} \stackrel{a_1}{\longrightarrow} \stackrel{a_2}{\longrightarrow} \stackrel{a_2}{\longrightarrow} \stackrel{a_1}{\longrightarrow} \stackrel{a_2}{\longrightarrow} \stackrel{a_1}{\longrightarrow} \stackrel{a_2}{\longrightarrow} \stackrel{a_1}{\longrightarrow} \stackrel{a_2}{\longrightarrow} \stackrel{a_2}{\longrightarrow} \stackrel{a_1}{\longrightarrow} \stackrel{a_2}{\longrightarrow} \stackrel{a_1}{\longrightarrow} \stackrel{a_2}{\longrightarrow} \stackrel{a_1}{\longrightarrow} \stackrel{a_2}{\longrightarrow} \stackrel{a_2}{\longrightarrow} \stackrel{a_1}{\longrightarrow} \stackrel{a_2}{\longrightarrow} \stackrel{a_2}{\longrightarrow} \stackrel{a_1}{\longrightarrow} \stackrel{a_2}{\longrightarrow} \stackrel{a_1}{\longrightarrow} \stackrel{a_2}{\longrightarrow} \stackrel{a_2}{\longrightarrow} \stackrel{a_1}{\longrightarrow} \stackrel{a_2}{\longrightarrow} \stackrel{a_2}{\longrightarrow} \stackrel{a_1}{\longrightarrow} \stackrel{a_2}{\longrightarrow} \stackrel{a_2}{\longrightarrow} \stackrel{a_1}{\longrightarrow} \stackrel{a_1}{\longrightarrow} \stackrel{a_2}{\longrightarrow} \stackrel{a_1}{\longrightarrow} \stackrel{a_2}{\longrightarrow} \stackrel{a_1}{\longrightarrow} \stackrel{a_1}{\longrightarrow}$$

25 Compounds of formula (I) wherein, in the definition of O, both R⁶ substituents are hydrogen or R² and R⁴ are both hydrogen, said O being represented by H₂N-O₂, and said compounds being represented by formula (I-a-1), can also be prepared by deprotecting an intermediate of formula (V).

Said deprotection reaction can be performed in the presence of a suitable base such as, for example hydrazine, or in the presence of a suitable acid, such as hydrochloric acid and the like, in a suitable solvent, such as an alcohol, acetic acid and the like.

Compounds of formula (I-a-1) can also be prepared by deprotecting an intermediate of formula (VI) according to the procedure described for the preparation of compounds of formula (I-a).

Compounds of formula (I-a) or (I-a-1), wherein Q₁ or Q₂ comprise a hydroxy substituent, said Q₁ or Q₂ being represented by Q₁·(OH) or Q₂·(OH), and said compounds being represented by formula (I-a-2) or (I-a-1-1), can be prepared by deprotecting an intermediate of formula (VII) or (VIII) as described hereinabove for the preparation of compounds of formula (I-a).

$$P = Q_{1} \cdot (OP) \longrightarrow \begin{pmatrix} R^{1} & & & \\ & A^{2} & A^{2} & A^{3} \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\$$

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Compounds of formula (I) wherein, in the definition of Q, both R⁶ substituents are hydrogen or R² and R⁴ are both hydrogen, and the carbon adjacent to the nitrogen carrying the R⁶, or R² and R⁴ substituents contains at least one hydrogen, said Q being represented by H₂N-Q₃H, and said compounds being represented by formula (I-a-1-2) can also be obtained by reductive amination of intermediates of formula (IX) in the presence of a suitable amination reagent, such as, for example, ammonia, hydroxylamine, or benzylamine, and in the presence of a suitable reducing agent, e.g. hydrogen, and an appropriate catalyst. An appropriate catalyst in the above reaction is, for example, platinum-on-charcoal, palladium-on-charcoal, rhodium-on-Al₂O₃, and the like, optionally in the presence of a catalyst poison, such as a thiophene solution. A suitable reaction-inert solvent for the above reaction is, for example, an alcohol, e.g. methanol, ethanol, 2-propanol and the like.

$$(O=)Q_3 \xrightarrow{R^1} a_{a_1}^{1} a_{a_2}^{2} \qquad \text{amination}$$

$$(IX) \qquad H_2N = Q_3H \xrightarrow{N} a_{a_4}^{1} a_{a_2}^{1}$$

Compounds of formula (I), wherein Q comprises a -CH₂NH₂ moiety, said Q being represented by H₂N-CH₂-Q₄, and said compounds being represented by formula (I-a-1-3) can be prepared by reducing an intermediate of formula (X).

NC-Q₄

$$\stackrel{A}{=}$$
 $\stackrel{A_1}{=}$
 $\stackrel{A_2}{=}$
 $\stackrel{A_1}{=}$
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 $\stackrel{A_1}{=}$
 $\stackrel{A_1}{=}$

Said reduction can be performed with a suitable reducing agent, such as lithium aluminium hydride or hydrogen, optionally in the presence of a suitable catalyst, such as Raney Nickel. A suitable solvent for the above reaction is, for example, tetrahydrofuran, or a solution of ammonia in an alcohol. Suitable alcohols are methanol, ethanol, 2-propanol and the like. Said reduction reaction performed in a solution of ammonia in an alcohol can also be used to prepare compounds of formula (I-a-1-3), wherein R¹ is substituted with C₁₋₆alkyloxyC₁₋₆alkyl, said R¹ being represented by R¹'-C₁₋₆alkyloxyC₁₋₆alkyl, and said compounds being represented by formula (I-a-1-3-1) starting from an intermediate of formula (X-a).

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NC-Q₄

$$= \begin{array}{c} R^{1} - C_{1-6}alkyl - OH \\ N - C_{1-$$

Compounds of formula (I), wherein Q comprises a -CH₂-CHOH-CH₂-NH₂ moiety, said Q being represented by H₂N-CH₂-CHOH-CH₂-Q₄, and said compounds being represented by formula (I-a-1-3-2), can be prepared by reacting an intermediate of formula (XI) with ammonia in the presence of a suitable reaction-inert solvent, such as an alcohol, e.g. methanol.

Compounds of formula (I), wherein, in the definition of Q, R² or one R⁶ substituent is formyl, said Q being represented by H-C(=O)-Q₁, and said compounds being represented by formula (I-b), can be prepared by reacting an intermediate of formula (XII) with formic acid, formamide and ammonia.

$$C_{1^{-4}alkyl} = C_{1^{-4}alkyl} = C_{1^{-4}a$$

Compounds of formula (I), wherein, in the definition of Q, R² is other than hydrogen, said R² being represented by R^{2a}, R⁴ is hydrogen, and the carbon atom adjacent to the nitrogen atom carrying the R² and R⁴ substituents, carries also at least one hydrogen atom, said Q being represented by R^{2a}-NH-HQ₅, and said compounds being represented by formula (I-c), can be prepared by reductive amination of an intermediate of formula (XIII) with an intermediate of formula (XIV) in the presence of a suitable reducing agent, such as hydrogen, and a suitable catalyst, such as palladium-on-charcoal, platinum-on-charcoal, and the like. A suitable reaction-inert solvent for the above reaction is, for example, an alcohol, e.g. methanol, ethanol, 2-propanol and the like.

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$$(O=)Q_{5} \xrightarrow{R^{1}} A^{2} A^{2} A^{3} + R^{2a} - NH_{2} \xrightarrow{amination} R^{2a} - NH - HQ_{5} \times A^{2a} A^{2a} A^{3}$$

$$(XIII) \qquad (XIV)$$

Compounds of formula (I-c), wherein R^{2a} represents C_{1-10} alkyl substituted with $N(R^6)_2$ and with hydroxy, and the carbon atom carrying the hydroxy, carries also two hydrogen atoms, said R^{2a} being represented by $[(C_{1-9}alkyl)CH_2OH]-N(R^6)_2$, and said compounds being represented by formula (I-c-1), can be prepared by reducing an intermediate of formula (XV) in the presence of a suitable reducing agent, such as lithium aluminium hydride, in a suitable reaction-inert solvent, such as tetrahydrofuran.

$$(R^{6})_{2}N-(C_{1}-\text{palkyl})-NH-HQ_{5}-N$$

$$C(=O)OC_{1}-\text{palkyl}$$

$$(XV)$$

$$R^{1}$$

$$G$$

$$R^{1}$$

$$G$$

$$R^{1}$$

$$G$$

$$(C_{1}-\text{palkyl})-NH-HQ_{5}$$

$$CH_{2}OH$$

$$(I-c-1)$$

Compounds of formula (I) wherein, in the definition of Q, R² or one R⁶ substituent is hydrogen, said Q being represented by H-Q₁, and wherein R¹ is a bicyclic heterocycle substituted with 1 or more substituents selected from hydroxy, hydroxyC₁₋₆alkyl, or HO(-CH₂-CH₂-O)_n-, said substituents being represented by formula A-OH, said R¹ being represented by R^{1a}-(A-OH)_w, with w being the amount of substituents on R^{1a} ranging from 1 to 4, and said compounds being represented by formula (I-d), can be prepared by deprotecting an intermediate of formula (XVI) with a suitable acid, such as hydrochloric acid and the like, optionally in the presence of a suitable solvent, such as an alcohol. Suitable alcohols are methanol, ethanol, 2-propanol and the like. Alternatively, one protecting group may also protect more than one substituent of R^{1a}, said protecting group being represented by P₁, as represented by formula (XVI-a). The two ways of protecting the substituents of R^{1a}, i.e. with a separate, as in formula (XVI), or a combined, as in formula (XVI-a), protecting group, may also be combined in the same intermediate, as represented by formula (XVI-b).

Compounds of formula (I), wherein Q is a radical of formula (b-2), said compounds being represented by formula (I-e), can be prepared by reacting an intermediate of formula (XVII) with an intermediate of formula (XVIII) in the presence of sodium cyanide and a suitable reaction-inert solvent, such as an alcohol, e.g. methanol and the like.

$$C_{1^{-4}alkyl} - O - C_{-Alk} - X^{1} - Alk - X^{1} - A$$

Compounds of formula (I), wherein in the definition of Q, X² is C₂₋₄alkyl-NR⁴, said Q being represented by Q₆N-CH₂-C₁₋₃alkyl-NR⁴, and said compounds being represented by formula (I-p), can be prepared by reacting an intermediate of formula (XIX) with an intermediate of formula (XX) in the presence of isopropyl titanate (IV) and a suitable

reducing agent, such as NaBH₃CN, and in the presence of a suitable reaction-inert solvent, such as methylene chloride and an alcohol, e.g. ethanol.

Compounds of formula (I-p), wherein R^2 is C_{1-6} alkylcarbonyl, and Q is a radical of formula (b-6), wherein Y^1 is NR^2 , said compounds being represented by formula (I-p-1), can be prepared by reacting an intermediate of formula (XIX) with an intermediate of formula (XX-a) according to the procedure described for the preparation of a compound of formula (I-p).

$$\begin{array}{c} O \\ H - C - C_{1-3}alkyl - NR^4 - A_{1-6}a^{1/3} \\ (XIX) \end{array} + \begin{array}{c} C_{1-6}alkyl - C - N \\ (XIX) \end{array} + \begin{array}{c} C_{1-6}alkyl - C - N \\ (XIX) \end{array}$$

(I-p-1)

Compounds of formula (I), wherein G is substituted with hydroxy or HO(-CH₂CH₂O)_n-, said G being represented by G₁-OH, and said compounds being represented by formula (I-q), may be prepared by deprotecting an intermediate of formula (XXI), wherein P represents a suitable protecting group, for example, benzyl. Said deprotection reaction can be performed by catalytic hydrogenation in the presence of hydrogen and an appropriate catalyst in a reaction-inert solvent. A suitable catalyst in the above reaction is, for example, platinum-on-charcoal, palladium-on-charcoal, and the like. An appropriate reaction-inert solvent for said reaction is, for example, an alcohol, e.g. methanol, ethanol, 2-propanol and the like, an ester, e.g. ethylacetate and the like, an acid, e.g. acetic acid and the like.

$$\begin{array}{c}
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P \longrightarrow G_1 \\
Q \longrightarrow N \longrightarrow a^1 \longrightarrow a^2 \\
N \longrightarrow a^1 \longrightarrow a^2
\end{array}$$
(XXI)
$$(I-q)$$

Compounds of formula (I), wherein G is substituted with hydroxy and the carbon atom carrying the hydroxy substituent carries also at least one hydrogen, said G being represented by H-G₂-OH, and said compounds being represented by formula (I-q-1), can also be prepared by reducing an intermediate of formula (XXII).

$$Q = \begin{pmatrix} R^1 \\ H - G_2 - OH \\ N - A_3 - A_3 \end{pmatrix}$$
reduction
$$Q = \begin{pmatrix} R^1 \\ H - G_2 - OH \\ N - A_3 - A_3 \\ N - A_3 - A_3 \end{pmatrix}$$
(XXII)

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Said reduction reaction can be performed in the presence of a suitable reducing agent, such as, for example sodium borohydride, in a reaction-inert solvent, such as an alcohol or tetrahydrofuran or a mixture thereof. Suitable alcohols are methanol, ethanol, 2-propanol and the like.

Compounds of formula (I) may be converted into each other following art-known functional group transformation reactions, comprising those described hereinafter.

The compounds of formula (I) may be converted to the corresponding N-oxide forms following art-known procedures for converting a trivalent nitrogen into its N-oxide form. Said N-oxidation reaction may generally be carried out by reacting the starting material of formula (I) with an appropriate organic or inorganic peroxide. Appropriate inorganic peroxides comprise; for example, hydrogen peroxide, alkali metal or earth alkaline metal peroxides, e.g. sodium peroxide, potassium peroxide; appropriate organic peroxides may comprise peroxy acids such as, for example, benzenecarboperoxoic acid or halo substituted benzenecarboperoxoic acid, e.g. 3-chlorobenzenecarboperoxoic acid, peroxoalkanoic acids, e.g. peroxoacetic acid, alkylhydroperoxides, e.g. t.butyl hydro-peroxide. Suitable solvents are, for example, water, lower alcohols, e.g. ethanol and the like, hydrocarbons, e.g. toluene, ketones, e.g. 2-butanone, halogenated hydrocarbons, e.g. dichloromethane, and mixtures of such solvents.

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Compounds of formula (I), wherein R¹ is a bicyclic heterocycle substituted with C₁₋₆alkyloxycarbonyl, said R¹ being represented by R¹-C(=0)OC₁₋₆alkyl, and said compounds being represented by formula (I-f), can be prepared by esterification of a compound of formula (I-g) in the presence of a suitable alcohol, e.g. methanol, ethanol, propanol, butanol, pentanol, hexanol and the like, and in the presence of a suitable acid, such as hydrochloric acid and the like.

Compounds of formula (I-a) may be converted into compounds of formula (I) wherein, in the definition of Q, R^2 or at least one R^6 substituent is other than hydrogen, said R^2 or R^6 being represented by Z_1 , said Q being represented by Z_1 - Q_1 , and said compounds being represented by formula (I-h), by reaction with a reagent of formula (XXIII), wherein W_2 is a suitable leaving group, such as a halo atom, e.g. bromo, or 4-methylbenzenesulphonate, in the presence of a suitable base, such as, for example disodium carbonate, dipotassium carbonate, sodium hydroxide and the like, in a reaction-inert solvent, e.g. 3-methyl-2-butanone, acetonitrile, N,N-dimethylformamide.

Compounds of formula (I-h), wherein, in the definition of Z_1 , R^2 is CH_2 - C_{1-9} alkyl substituted with $N(R^6)_2$, said compounds being represented by formula (I-h-1), can also be prepared by reacting a compound of formula (I-a) wherein, in the definition of H- Q_1 , R^2 is hydrogen, said H- Q_1 being represented by H- Q_{1b} , and said compounds being represented by formula (I-a-3), with an intermediate of formula (XXIV), in the presence of a suitable reducing agent, such as sodium cyanoborohydride, in a suitable reaction-inert solvent, such as an alcohol.

Compounds of formula (I-h), wherein Z₁ comprises formyl, C₁₋₆alkylcarbonyl, Hetcarbonyl or C₁₋₆alkyloxycarbonyl, said Z₁ being represented by Z_{1a}, and said compounds being represented by formula (I-h-2), can be converted into compounds of formula (I-a), by acidic hydrolysis in the presence of a suitable acid, such as hydrobromic, hydrochloric, sulfuric, acetic, or trifluoroacetic acid or a mixture of said acids, or by alkaline hydrolysis in the presence of a suitable base, such as, for example potassium hydroxide, in a suitable solvent such as water, alcohol, a mixture of water-alcohol, methylene chloride. Suitable alcohols are methanol, ethanol, 2-propanol, 1-butanol, sec. butanol and the like. In order to enhance the rate of the reaction, it is advantageous to work at elevated temperatures.

$$Z_{1a}$$
 Q_1 Q_1 Q_1 Q_1 Q_1 Q_1 Q_1 Q_1 Q_1 Q_2 Q_3 Q_4 Q

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Compounds of formula (I-b) can be prepared by reacting a compound of formula (I-a) with formic acid.

$$H = Q_1 = \begin{pmatrix} R^1 & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & &$$

Compounds of formula (I) wherein R^1 is a bicyclic heterocycle substituted with hydroxy, said R^1 being represented by HO- $R^{1'}$, and said compounds being represented by formula (I-i), can be prepared by deprotecting a compound of formula (I-j), wherein R^1 is a bicyclic heterocycle substituted with C_{1-6} alkyloxy or aryl C_{1-6} alkyloxy, said C_{1-6} alkyl or aryl C_{1-6} alkyl being represented by Z_2 , and said R^1 being represented by Z_2 -O- $R^{1'}$. Said deprotection can be performed in a reaction-inert solvent, such as, for

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example methylene chloride, in the presence of a suitable deprotecting agent, e.g. tribromoborane.

$$Q = \begin{bmatrix} O - Z_2 & O H & O H \\ R^{1'} & G & O H \\ R^$$

Compounds of formula (I) wherein R¹ is a bicyclic heterocycle substituted with halo(-CH₂-CH₂-O)_n, said compounds being represented by formula (I-k), can be converted into compounds of formula (I-l-1) or (I-l-2) by reaction with an appropriate amine of formula (XXV) or (XXVI) in a suitable reaction-inert solvent, e.g. tetrahydrofuran.

Compounds of formula (I) wherein R¹ is a bicyclic heterocycle substituted with halo, said compounds being represented by formula (I-m) can be converted into compounds of formula (I) by reaction with 1-butanethiol in the presence of palladium-on-charcoal and CaO in a suitable reaction-inert solvent, such as tetrahydrofuran.

Q
$$R^{1'}$$
 Q R^{1} A^{2} A^{3} A^{2} A^{3} A^{3} A^{3} A^{3} A^{4} A^{3} A^{4} A^{3} A^{4} A^{3}

Compounds of formula (I) wherein a hydrogen atom in the radicals of formula (a-1), (a-2), (a-3), (a-4) or (a-5) is replaced by nitro, said compounds being represented by formula (I-n) may be reduced to a compound of formula (I-o) in the presence of a suitable reducing agent, such as hydrogen, optionally in the presence of a suitable catalyst, such as platinum-on-charcoal, and optionally in the presence of a suitable catalyst poison, e.g. a thiophene solution. The reaction may be performed in a suitable reaction-inert solvent, such as an alcohol.

In the following paragraphs, there are described several methods of preparing the intermediates in the foregoing preparations. A number of intermediates and starting materials are commercially available or are known compounds which may be prepared according to conventional reaction procedures generally known in the art or analogous to the procedures described in EP-A-0005318, EP-A-0099139, EP-A-0151824, EP-A-0151826, EP-A-0232937, EP-A-0295742, EP-A-0297661, EP-A-0539420, EP-A-0539421, US 4,634,704, US 4,695,569.

In the foregoing and the following preparations, the reaction mixture is worked up following art-known methods and the reaction product is isolated and, if necessary, further purified.

Intermediates of formula (III) can be prepared by reacting an intermediate of formula (XXVII) with a suitable leaving group, i.e. W₁, introducing agent, e.g. 1-halo-2,5-pyrrolidinedione in the presence of dibenzoyl peroxide, in a reaction-inert solvent,

e.g. tetrachloromethane.

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$$R^1-G-H$$
 O
 N
 R^1-G-W_1
 $(XXVII)$
 (III)

Intermediates of formula (XXVII), wherein R¹ is a bicyclic heterocycle substituted with chloro, said R¹ being represented by Cl-R¹ and said intermediates being represented by formula (XXVII-a) can be prepared by reacting an intermediate of formula (XXVIII),

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wherein (O=)R^{1b}H is defined as a carbonyl derivative of R¹ wherein one carbon or nitrogen, adjacent to the carbonyl, carries at least one hydrogen, with phosphorus oxychloride. Intermediates of formula (XXVIII) may also react as their enol tautomeric forms.

$$(O=)R^{1b}H - G - H \qquad POCl_3$$

$$(XXVIII) \qquad (XXVII-a)$$

Intermediates of formula (XXVII), wherein R¹ is 2-trifluoromethyl-3-methyl (3*H*)-imidazo[4,5-b]pyridine, and G is CH₂, said intermediates being represented by formula (XXVII-b), can be prepared by reacting N-2,6-dimethyl-2,3-pyridinediamine (Heterocycles, 38, p 529, 1994), with trifluoroacetic acid.

Intermediates of formula (III) wherein W_1 is chloro, which is attached to a carbon atom carrying at least one hydrogen, said G being represented by G_3H , and said intermediates being represented by formula (III-a) can also be prepared by reacting an intermediate of formula (XXIX) with thionylchloride in a reaction-inert solvent, e.g. methylenechloride.

 R^1 — G_3H —OH \longrightarrow R^1 — G_3H —O

(XXIX) (III-a)

Intermediates of formula (XXIX) can be prepared by reducing an intermediate of formula (XXX) in a reaction-inert solvent, e.g. an alcohol, in the presence of a suitable reducing agent, e.g. sodium borohydride.

$$R^1$$
— $G_3(=O)$ reduction R^1 — G_3H — OH (XXX)

Alternatively, intermediates of formula (XXIX) can also be prepared by deprotecting an intermediate of formula (XXXI), wherein P is a suitable protecting group, e.g. C₁₋₄alkylcarbonyl, in a reaction-inert solvent, such as an alcohol, in the presence of a suitable base, e.g. sodium hydroxide.

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$$R^1 - G_3H - O - P$$
 $R^1 - G_3H - OH$
(XXXI) (XXIX)

Intermediates of formula (XXX), wherein $G_3(=0)$ is CH(=0), said intermediates being represented by formula (XXX-a), can be prepared by reacting an intermediate of formula (XXXII), wherein W_3 is a suitable leaving group, such as a halo atom, e.g. bromo, with N_1N_2 -dimethylformamide in the presence of butyllithium in a reaction-inert solvent, e.g. tetrahydrofuran, diethylether or a mixture thereof.

$$R^{\perp}W_3$$
 $R^{\perp}CH(=0)$ (XXX-a)

Intermediates of formula (XXX-a) can also be prepared by oxidizing an intermediate of formula R¹-CH₂-OH in the presence of a suitable oxidizing agent, e.g. MnO₂ in a reaction-inert solvent, e.g. methylenechloride.

$$R^1$$
— CH_2 — OH R^1 — $CH(=O)$ (XXX-a)

Intermediates of formula R¹-CH₂-OH, wherein R¹ is 2,3-dimethylquinoxaline, said intermediates being represented by formula (XCI) can be prepared by reducing an intermediate of formula (XCII) in a reaction-inert solvent, e.g. tetrahydrofuran, in the presence of a suitable reducing agent, e.g. potassium borohydride in the presence of lithium chloride.

Intermediates of formula (XCII) can be prepared by reacting ethyl 2,3-diaminobenzoate (Tetrahydron, 28, 3271, 1972) with 2,3-butanedione in the presence of disodium disulfite.

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Intermediates of formula (XXXI), wherein R^1 is 5,6,7,8-tetrahydroquinoline, which can optionally be substituted, G_3H is CH_2 , and P is C_{1-4} alkylcarbonyl, said intermediates being represented by formula (XXXI-a) can be prepared by reacting an intermediate of formula (XCIII) with C_{1-4} alkylacid anhydride at elevated temperatures in the presence of a suitable base, e.g. sodium hydroxide.

(XCIII)
$$CH_3$$
 CH_2 CH_2

Intermediates of formula (XCIII) can be prepared by oxidizing an intermediate of formula (XCIV) with a suitable oxidizing agent, e.g. a peroxide such as 3-chlorobenzenecarboperoxoic acid, in a reaction-inert solvent, e.g. methylene chloride.

Intermediates of formula (XCIV) can be prepared by reducing an intermediate of formula (XCV) (Org. Prep. Proced. Int., 23, p 386-387, 1991) with an appropriate reducing agent, e.g. hydrogen, in the presence of a suitable catalyst, e.g. palladium-on-charcoal, and a suitable acid, e.g. trifluoroacetic acid.

Intermediates of formula (IV) can be prepared by reacting an intermediate of formula (XXXIII-a) or (XXXIII-b), wherein P represents a suitable protecting group, such as, for example, C₁₋₄alkyloxycarbonyl, with an intermediate of formula (III) according to the reaction described for the general preparation of compounds of formula (I).

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$$P = Q_{1} \longrightarrow \begin{pmatrix} A_{1} & A_{2} & A_{3} & A_{4} & A_{3} & A_{4} & A_{4}$$

Intermediates of formula (IV) can also be prepared by reacting an intermediate of formula (XXXIII-a) with an intermediate of formula (XXXIV) that has reacted with methanesulfonyl chloride, in the presence of a suitable base, such as sodium hydride, and in the presence of a suitable reaction-inert solvent, e.g. N,N-dimethylformamide.

$$P = Q_{1} = \begin{bmatrix} A_{1} & A_{2} & A_{3} & A_{4} & A_{4}$$

Intermediates of formula (IV) can also be prepared by a cyclization reaction of an intermediate of formula (XXXV) in a reaction-inert solvent, e.g. an alcohol or *N,N*-dimethylformamide, in the presence of mercury oxide and sulphur.

Intermediates of formula (IV) wherein Q_1 comprises an unsaturated bond, said Q_1 being represented by $Q_{1a}(CH=CH)$, and said intermediates by formula (IV-a), can be prepared by reacting an intermediate of formula (XXXVI) with an intermediate of formula (III) in the presence of a suitable base, such as dipotassium carbonate.

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$$P = Q_{1a}(CH = CH)$$

$$N = A^{1} - G - W_{1}$$

$$(III)$$

$$(IV-a)$$

Intermediates of formula (IV) wherein, in the definition of Q_1 , the X^1 or X^2 moieties in the radicals of formula (b-1) to (b-8) represent NH, said Q_1 being represented by Q_{1c} -NH, and said intermediates by formula (IV-b), may also be prepared by reacting an intermediate of formula (XXXVII) with an intermediate of formula (XXXVIII).

halo
$$= \begin{pmatrix} R^1 \\ N \\ A \\ A \\ A \end{pmatrix} \begin{pmatrix} A^1 \\ A^2 \\ A \end{pmatrix} \begin{pmatrix} A^1 \\ A$$

Intermediates of formula (IV) wherein R¹ is a bicyclic heterocycle substituted with amino or mono- or di(C₁₋₆alkyl)amino, said R¹ being represented by R^{5a}R^{5b}N-R¹, wherein R^{5a} and R^{5b} are defined as described above, and said intermediates being represented by formula (IV-c), can be prepared by reacting an intermediate of formula (XXXIX) with an appropriate amine, represented by formula (XL), in the presence of an appropriate catalyst, e.g. palladium, and (R)-(+)-2,2'-bis(diphenyl-phosphino)-1,1'-binaphtyl, in a suitable reaction-inert solvent, e.g. tetrahydrofuran.

halo
$$R^{1'}$$
 R^{5a}
 R^{5a}
 R^{5a}
 $R^{1'}$
 R^{5a}
 R^{5a}

Intermediates of formula (IV) wherein R¹ is a bicyclic heterocycle substituted with C(=O)-NR^{5a}R^{5b}, wherein R^{5a} and R^{5b} are defined as described above, said R¹ being represented by R^{5a}R^{5b}N-C(=O)-R¹, and said intermediates being represented by formula (IV-d), can be prepared by reacting an intermediate of formula (XXXIX) with an appropriate amine, represented by formula (XL), under an atmosphere of carbon monoxide, in the presence of a suitable catalyst, e.g. palladium (II) acetate, and

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1,3-bis(diphenylphosphino)propane, in a suitable reaction-inert solvent, e.g. tetrahydrofuran.

halo—
$$R^{1'}$$
 R^{5a}
 R^{5a}

Intermediates of formula (IV) wherein P-Q₁ comprises C₁₋₁₀alkyl or C₃₋₇cycloalkyl substituted with NR⁶-P, said C₁₋₁₀alkyl or C₃₋₇cycloalkyl being represented by Z₃, said P-Q₁ being represented by P-NR⁶-Z₃-Q_{1b}, and said intermediates being represented by formula (IV-e), can be prepared by reacting a compound of formula (I-a-3) with an intermediate of formula (XLI), wherein W₄ represents a suitable leaving group, such as p-toluenesulphonate. Said reaction can be performed in a reaction-inert solvent, e.g. acetonitrile, in the presence of a suitable base, e.g. dipotassium carbonate.

Intermediates of formula (IV-e), wherein R⁶ is hydroxyC₁₋₆alkyl, said intermediates being represented by formula (IV-e-1), can be prepared by reacting an intermediate of formula (XLII) with an intermediate of formula (XLIII) in the presence of a suitable base, e.g. dipotassium carbonate, and a suitable solvent, e.g. acetonitrile.

$$Q = Q_{16} = Q_{16}$$

Intermediates of formula (XXXIII-a) or (XXXIII-b) can be prepared by protecting an intermediate of formula (XLIV) with a suitable protecting group, such as, for example, C₁₋₄alkyloxycarbonyl, in a reaction-inert solvent, such as methylene chloride or an alcohol, e.g. methanol, ethanol, 2-propanol and the like, in the presence of a suitable

reagent, e.g. di C₁₋₄alkyl dicarbonate and optionally in the presence of a suitable base, e.g. sodium acetate.

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Alternatively, intermediates of formula (XXXIII-a) or (XXXIII-b) can be converted into an intermediate of formula (XLIV) by reaction with a suitable acid, such as hydrochloric acid or hydrobromic acid and the like or mixtures thereof, in the presence of a suitable solvent, e.g. water.

Intermediates of formula (XXXIII-a) or (XXXIII-b), wherein in the definition of Q₁, the X¹ or X² moieties in the radicals of formula (b-1) to (b-8) represent NH, said Q₁ being represented by Q_{1c}-NH, and said intermediates by formula (XXXIII-a-1) or (XXXIII-b-1), can be prepared by reacting an intermediate of formula (XLV-a) or (XLV-b), wherein W₅ represents a suitable leaving group, such as for example a halo atom, e.g. chloro, with an intermediate of formula (XLVI).

$$W_{5} = \begin{pmatrix} A_{1} & A_{2} & A_{3} & A_{4} & A_{3} & A_{4} & A_$$

Intermediates of formula (XLV-a) or (XLV-b) can be prepared by reacting an intermediate of formula (XLVII-a) or (XLVII-b) with $H_2P(=O)(W_5)_3$ in the presence of a suitable acid, e.g. hydrochloric acid.

$$O = \bigvee_{N=1}^{H} \bigvee_{a^4 = a^3}^{a^2} \bigvee_{a^4 = a^3}^{H_2P(=O)(W_5)_3} \bigvee_{N=1}^{H} \bigvee_{a^4 = a^3}^{A^3} \bigvee_{N=1}^{A^3} \bigvee_{N=1}^{A^3$$

Intermediates of formula (XLVII-a) or (XLVII-b) can be prepared by reacting an intermediate of formula (XLVIII-a) or (XLVIII-b) with an intermediate of formula (IL).

$$\begin{array}{c} H \\ HN \\ H2N \\ \hline \\ (XLVIII-a) \end{array} + H2N \\ \hline \\ (IL) \\ \hline \\ (XLVIII-a) \end{array} + H2N \\ \hline \\ (IL) \\ \hline \\ (XLVIII-b) \end{array}$$

$$\begin{array}{c} H \\ N \\ 1 \\ 2 \\ 3 \\ 1 \\ 3 \\ 1 \end{array}$$

$$(XLVIII-b)$$

Intermediates of formula (XXXIII-a) can also be prepared by reacting an intermediate of formula (XLVIII-a) with P-Q₁-C(=NH)-O-CH₂-CH₃ in a reaction-inert solvent, such as an alcohol.

Intermediates of formula (XXXV) can be prepared by reacting an intermediate of formula (L) with an intermediate of formula P-Q₁=C=S, which is synthesized according to the procedures described in EP 0005318, in a reaction-inert solvent, such as an alcohol, e.g. ethanol. To increase the reaction rate, the reaction may be performed at elevated temperatures.

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$$R^{1}$$
 G HN A^{1} A^{2} A^{2} A^{3} A^{2} A^{2} A^{3} A^{2} A^{3} A^{2} A^{3} A^{3} A^{4} A^{3} A^{3} A^{4} A^{3} A^{4} A^{3} A^{4} A^{3} A^{4} $A^{$

Intermediates of formula (L) can be obtained by reducing an intermediate of formula (LI) in a reaction-inert solvent, e.g. an alcohol, in the presence of a suitable reducing agent, e.g. hydrogen, and an appropriate catalyst, e.g. Raney Nickel.

Intermediates of formula (LII) can be prepared by reacting an intermediate of formula (LII) with an intermediate of formula (LIII), in which W₆ represents a suitable leaving group, such as a halo atom, e.g. chloro. This reaction may be performed in a reaction-inert solvent, e.g. acetonitrile, in the presence of a suitable base, e.g. dipotassium carbonate.

$$R^{\perp} - G - NH_2 + W_6 - A_2 - A_3 - A_4 - A_3 - A_4 - A_3 - A_4 - A_3 - A_4 - A_4$$

Intermediates of formula (LII) can be prepared by reacting an intermediate of formula (LIV) with a suitable acid, such as hydrochloric acid, in the presence of a suitable solvent, e.g. an alcohol, e.g. ethanol.

$$R^{1} - G - N$$

$$C = O$$

$$H$$

$$(LIV)$$

$$R^{1} - G - NH_{2}$$

$$(LII)$$

Intermediates of formula (LIV) can be prepared by reacting an intermediate of formula (III) with $NaN[C(=O)H]_2$.

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$$R^{\perp}G-W_1$$
 + NaN[C(=O)H]₂ $R^{\perp}G-N$ $C=O$
(III)
(LIV)

Intermediates of formula (LI) can also be prepared by reacting an intermediate of formula (LIII) with an intermediate of formula (LV) (J. Org. Chem., 25, p 1138, 1960) in a reaction-inert solvent, e.g. N,N-dimethylformamide, in the presence of an appropriate base, e.g. sodium hydride.

Intermediates of formula (XXXVI) can be prepared by dehydrating an intermediate of formula (LVI) with a suitable acid, such as sulfuric acid.

$$P \longrightarrow Q_{1a}(CH_2-CHOH) \longrightarrow N \longrightarrow a^{1 \atop 1}a^{2}$$

$$(LVI) \qquad P \longrightarrow Q_{1a}(CH=CH) \longrightarrow N \longrightarrow a^{1 \atop 1}a^{2}$$

$$(XXXVI)$$

Intermediates of formula (LVI) wherein, in the definition of Q_{1a}, the X¹ or X² moieties are CH₂, said Q_{1a} being represented by Q_{1a}, and said intermediates being represented by formula (LVI-a), can be prepared by reacting a carbonyl moiety of formula (LVII) with an intermediate of formula (LVIII) in the presence of N,N-diisopropylamine and butyl lithium, in a suitable reaction-inert solvent, e.g. tetrahydrofuran.

P—Q_{1a}(CH₂·C=O) + CH₃
$$\stackrel{\text{H}}{\underset{\text{N}}{\bigvee}} \stackrel{\text{a}^{1}}{\underset{\text{a}^{4}}{\bigvee}} \stackrel{\text{a}^{1}}{\underset{\text{a}^{3}}{\bigvee}} = P$$

$$(LVII)$$
(LVIII) (LVI-a)

Intermediates of formula (IV), wherein G is C_{1-10} alkanediyl substituted with C_{1-6} alkyloxy, aryl C_{1-6} alkyloxy, HO(-CH₂CH₂O)_n-, C_{1-6} alkyloxy(-CH₂CH₂O)_n-, or aryl C_{1-6} alkyloxy(-CH₂CH₂O)_n-, said group of substituents being represented by O-Z₄, said G being represented by Z₄-O-G₁, and said intermediates being represented by formula (IV-f), can be prepared by reacting an intermediate of formula (XXXIII-a), with an intermediate of formula (LIX), optionally in the presence of a suitable acid, such as p-toluenesulfonic acid and the like, and optionally in the presence of a suitable solvent,

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such as N,N-dimethylacetamide. To increase the reaction rate, the reaction may be carried out at elevated temperatures.

$$P = Q_{1} = \begin{bmatrix} R^{1} & & & & \\ & & & \\ & & & \\ N & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & & \\ &$$

Intermediates of formula (LIX) can be prepared by reacting an intermediate of formula (LX) with a reagent of formula (LXI) or (LXII) in a reaction-inert solvent, such as an alcohol, or toluene, in the presence of an acid, e.g. 4-methylbenzenesulphonic acid.

$$Z_4$$
-O-H (LXI) or R^1 — G_1 (=O) Q - Z_4
(LX) Q - Z_4
 Z_4 —O-CH—O-C₁₋₂alkyl (LXII) (LIX)

Intermediates of formula (LX) can be prepared by oxidizing an intermediate of formula (LXIII) with a suitable oxidizing agent, e.g. MnO_2 , in a reaction-inert solvent, such as methylene chloride.

$$R^{1}$$
— G_{1} H—OH R^{1} — G_{1} (=O)

Intermediates of formula (IV-f) can also be prepared by reacting an intermediate of formula (IV) wherein G is C_{1-10} alkanediyl substituted with hydroxy, said G being represented by G_1 -OH, and said intermediates being represented by formula (IV-g), with an intermediate of formula (LXIV), wherein W_7 is a suitable leaving group, such as a halo atom, e.g. iodo, in the presence of a suitable base, e.g. sodium hydride, in a reaction-inert solvent, e.g. tetrahydrofuran.

$$P = Q_{1}$$

$$(IV-g)$$

$$Z_{4} = O - G_{1}$$

$$A_{1} = A_{2}$$

$$A_{2} = A_{3}$$

$$A_{3} = A_{4} = A_{3}$$

$$A_{4} = A_{3}$$

$$A_{5} = A_{5} = A_{5}$$

$$A_{7} = A_{1} = A_{2}$$

$$A_{1} = A_{2} = A_{3}$$

$$A_{2} = A_{3} = A_{4} = A_{3}$$

$$A_{3} = A_{4} = A_{3}$$

$$A_{4} = A_{5} = A_{5} = A_{5}$$

$$A_{5} = A_{5} = A_{5} = A_{5} = A_{5}$$

$$A_{5} = A_{5} = A_{5} = A_{5} = A_{5}$$

$$A_{7} = A_{7} = A_{7}$$

Intermediates of formula (IV-g), wherein the carbon atom of G₁ carrying the hydroxy, also carries a hydrogen atom, said G₁-OH being represented by H-G₂-OH, and said intermediates being represented by formula (IV-g-1), can be prepared by reducing an

intermediate of formula (LXV) in the presence of a suitable reducing agent, e.g. sodium borohydride, in a reaction-inert solvent, such as an alcohol, tetrahydrofuran or a mixture thereof. Intermediates of formula (LXV) can also first be deprotected, e.g. in the presence of a suitable acid, such as hydrochloric acid and the like, resulting in intermediates of formula (LXVI), followed by a reduction, resulting in a compound of formula (I-q-1) wherein Q represents H-Q₁, said compounds being represented by formula (I-q-1-1).

$$P = Q_1 = \begin{pmatrix} R^1 \\ G_2(=O) \end{pmatrix}$$

$$(LXV)$$

$$(IV-g-1)$$

$$R^1 \\ Q_2(=O) \\ R^1 \\ G_2(=O) \\ R^1 \\ R^1 \\ (IV-g-1)$$

$$R^1 \\ R^1 \\ R^1 \\ (IV-g-1)$$

$$R^1 \\ R^1 \\ R^1 \\ R^2 \\ (IV-g-1)$$

Intermediates of formula (IV), wherein G is ethyl substituted with hydroxy, said intermediates being represented by formula (IV-g-2) can also be prepared by reacting an intermediate of formula (XXXIII-a) with an intermediate of formula (LXVII) in the presence of a suitable base, such as sodium hydride, in a reaction-inert solvent, such as N,N-dimethylformamide.

$$P = Q_1 = \begin{pmatrix} R \\ N \end{pmatrix} \begin{pmatrix} A \\ A \end{pmatrix} \begin{pmatrix} A$$

A subgroup of intermediates of formula (IV-g-2), represented by formula (IV-g-2-1), can also be prepared by reacting an intermediate of formula (LXVIII) with an

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intermediate of formula (LXIX) in the presence of 1,3-dicyclohexylcarbodiimide, in a reaction-inert solvent, e.g. toluene.

Intermediates of formula (LXV) can be prepared by reacting an intermediate of formula (XXXIII-a) with an intermediate of formula (LXX), wherein W₈ is a suitable leaving group, such as a halo atom, e.g. bromo, in the presence of a suitable base, e.g. sodium hydride, in a reaction-inert solvent, e.g. N,N-dimethylformamide.

$$P = Q_{1} = \begin{pmatrix} R^{1} \\ Q_{2}(=0) \\ N \end{pmatrix} = \begin{pmatrix} R^{1} \\ Q_{2}($$

Intermediates of formula (V) can be prepared by reacting an intermediate of formula (LXXI) with 1*H*-isoindole-1,3 (2*H*)-dione in the presence of triphenylphosphine and diethyl azodicarboxylate.

$$HO-Q_{2}-\bigvee_{N}^{2}\bigvee_{a^{4}}^{a^{1}}\bigvee_{a^{3}}^{a^{2}}+\bigvee_{O}^{N}\bigvee_{N}^{2}\bigvee_{a^{4}}^{a^{1}}\bigvee_{a^{3}}^{a^{2}}$$

$$(LXXI)$$

Intermediates of formula (V) may also be prepared by reacting an intermediate of formula (LXXII) with 1H-isoindole-1,3 (2H)-dione in the presence of a suitable base, such as sodium hydride, and a suitable solvent, such as N, N-dimethylformamide.

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$$Q_{2} = \begin{pmatrix} R^{1} & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & &$$

Intermediates of formula (LXXII) can be prepared by reacting an intermediate of formula (LXXII) with an intermediate of formula (LXXIII), wherein W₉ represents a suitable leaving group, such as a halo atom, e.g. chloro, in the presence of a suitable base, such as N, N -diethyl-ethanamine, and a suitable solvent, such as methylene chloride.

Intermediates of formula (V), wherein in the definition of Q_2 , R^2 is C_{1-10} alkyl, said Q_2 being represented by C_{1-10} alkyl- Q_{1b} , and said intermediates by formula (V-a), can be prepared by reacting a compound of formula (I-a-3) with an intermediate of formula (LXXIV), wherein W_{10} is a suitable leaving group, such as a halo atom, e.g. chloro, in the presence of a suitable base, such as dipotassium carbonate, and a suitable solvent, such as acetonitrile.

$$H-Q_{16}$$

$$(I-a-3)$$

$$(LXXIV)$$

$$Q_{16}$$

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Intermediates of formula (LXXI) wherein, in the definition of Q₂, the carbon atom carrying the hydroxy, also carries two hydrogen atoms, said HO-Q₂ being represented by HO-CH₂-Q₂, and said intermediates being represented by formula (LXXI-a), can be prepared by reducing an intermediate of formula (LXXV) in the presence of a suitable reducing agent, such as lithium aluminium hydride, in a suitable reaction-inert solvent, e.g. tetrahydrofuran.

$$C_{1-4}$$
alkyl $-O$ — $C(=O)$ — Q_2 — N
 A_1
 A_2
 A_3
 A_4
 A_3
 A_4
 A_4
 A_3
 A_4
 A_4

Intermediates of formula (LXXI), wherein, in the definition of Q_2 , the carbon atom carrying the hydroxy, carries also at least one hydrogen, said HO- Q_2 being represented by HO- Q_3 H, and said intermediates being represented by formula (LXXI-b), can be prepared by reducing an intermediate of formula (IX) with a suitable reducing agent, e.g. sodium borohydride, in a reaction-inert solvent, e.g. an alcohol.

$$(O=)Q_3 \xrightarrow{\qquad \qquad \qquad \qquad } HO - Q_3 H \xrightarrow{\qquad \qquad \qquad } HO - Q_3 H \xrightarrow{\qquad } HO - Q_3 H \xrightarrow{\qquad \qquad } HO - Q_3 H$$

Intermediates of formula (VI) wherein, in the definition of Q₂, R² is C₁₋₁₀alkyl substituted with N(P)₂ and the carbon atom adjacent to the nitrogen atom carrying the R² substituent carries also at least one hydrogen atom, said Q₂ being represented by (P)₂N-C₁₋₁₀alkyl-NH-Q_{2a}H, and said intermediates being represented by formula (VI-a), can be prepared by reductive amination of an intermediate of formula (LXXVI) with an intermediate of formula (LXXVII) in the presence of a suitable reductive agent, such as hydrogen, and a suitable catalyst, such as palladium-on-charcoal, platinum-on-charcoal, and the like, and optionally in the presence of a suitable catalyst poison, such as a thiophene solution. A suitable solvent in this reaction is a reaction-inert solvent, such as an alcohol.

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Intermediates of formula (LXXVI) can be prepared by deprotecting an intermediate of formula (LXXVIII) in the presence of a suitable acid, such as hydrochloric acid and the like, in a suitable solvent, e.g. water.

Intermediates of formula (IX) may be prepared by deprotecting an intermediate of formula (LXXIX) in the presence of a suitable acid, e.g. hydrochloric acid and the like.

Intermediates of formula (LXXIX) can be prepared by reacting an intermediate of formula (LXXX) with an intermediate of formula (III) in the presence of a suitable base, e.g. dipotassium carbonate, in a suitable reaction-inert solvent, e.g. acetonitrile.

Intermediates of formula (LXXX) wherein, in the definition of Q_3 , the X^1 or X^2 moiety of the radicals of formula (b-1) to (b-8) represent NH, said Q_3 being represented by Q_3 -NH, and said intermediates being represented by formula (LXXX-a), may be prepared by cyclizing an intermediate of formula (LXXXI) in the presence of mercury oxide and sulphur, in a suitable reaction-inert solvent, e.g. an alcohol.

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Intermediates of formula (LXXXI) can be prepared by reducing an intermediate of formula (LXXXII) in the presence of a suitable reducing agent, such as hydrogen, in the presence of a suitable catalyst, such as palladium-on-charcoal, platinum-on-charcoal and the like, in a suitable solvent, e.g. a mixture of ammonia in alcohol. Suitable alcohols are methanol, ethanol, 2-propanol and the like.

Intermediates of formula (LXXXII) can be prepared by reacting an intermediate of formula (LXXXIII) with an intermediate of formula (LXXXIV) in a suitable reaction-inert solvent, e.g. ethanol.

Intermediates of formula (IX), wherein, in the definition of Q_3 , R^2 comprises C_{1-10} alkyl, said Q_3 being represented by C_{1-10} alkyl- Q_{1b} , and said intermediates being represented by formula (IX-a), can be prepared by reacting a compound of formula (I-a-3) with a reagent of formula (LXXXV), wherein $(O=)C_{1-10}$ alkyl represents a carbonyl derivative of C_{1-10} alkyl and wherein W_{11} is a suitable leaving group, such as a halo atom, e.g. bromo, in a reaction-inert solvent, e.g. acetonitrile, in the presence of a suitable base, e.g. dipotassium carbonate.

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$$H-Q_{1b} = \begin{pmatrix} R^1 & & & \\ & &$$

Intermediates of formula (X) wherein Q₄ comprises C₁₋₉alkyl, said Q₄ being represented by C₁₋₉alkyl-Q_{1b}, and said intermediates being represented by formula (X-a), can be prepared by reacting a compound of formula (I-a-3) with a reagent of formula (LXXXVI) wherein W₁₂ represents a suitable leaving group, such as a halo atom, e.g. chloro, in a reaction-inert solvent, e.g. 3-methyl-2-butanone, in the presence of a suitable base, e.g. dipotassium carbonate, sodium bicarbonate and the like.

$$H-Q_{1b} \xrightarrow{N} \stackrel{a^1}{=} \stackrel{a^2}{=} ^2 + W_{12}-C_{1-9alkyl}-CN \qquad NC-C_{1-9alkyl}-Q_{1b} \xrightarrow{N} \stackrel{a^1}{=} \stackrel{a^2}{=} ^3$$

$$(LXXXVI) \qquad (X-a)$$

Intermediates of formula (X), wherein NC-Q₄ represents NC-(C_{1.9}alkyl)(R⁴)N-C(=O)-Alk-X¹, said intermediates being represented by formula (X-b), can be prepared by reacting an intermediate of formula (LXXXVII) with an intermediate of formula (LXXXVIII) in the presence of di-1*H*-imidazol-2-yl-methanone, a suitable base, such as N, N-diethyl-ethanamine, and a suitable solvent, such as methylene chloride.

HO-C-Alk-X¹

$$(LXXXVIII)$$

$$(LXXXVIII)$$

$$R^{4}$$

$$NC-C_{1-9}alkyl$$

$$NC-C_{1-9}alkyl$$

$$NC-C_{1-9}alkyl$$

$$(X-b)$$

Intermediates of formula (XI), wherein Q₄ represents Q_{1b}, said intermediates being represented by formula (XI-a), can be prepared by reacting a compound of formula (I-a-3) with an intermediate of formula (LXXXIX), wherein W₁₃ represents a suitable leaving group, such as a halo atom, e.g. chloro, in the presence of a suitable base, such as disodium carbonate, and in the presence of a suitable solvent, such as 3-methyl-2-butanone.

$$H-Q_{1b} \xrightarrow{Q} Q_{1b} \xrightarrow{A^{1} A^{2}} A^{2} + Q_{1b} \xrightarrow{CH_{2}-W_{13}} Q_{1b} \xrightarrow{CH_{2}-Q_{1b}} Q_{1b} \xrightarrow{N} A^{2} A^{2} A^{3}$$

$$(I-a-3) \qquad (XI-a)$$

Intermediates of formula (XIX) can be prepared by reacting an intermediate of formula (XC) with a suitable acid, such as hydrochloric acid.

- Pure stereochemically isomeric forms of the compounds of formula (I) may be obtained by the application of art-known procedures. Diastereomers may be separated by physical methods such as selective crystallization and chromatographic techniques, e.g., countercurrent distribution, liquid chromatography and the like.
- 10 The compounds of formula (I) as prepared in the hereinabove described processes are generally racemic mixtures of enantiomers which can be separated from one another following art-known resolution procedures. The racemic compounds of formula (I) which are sufficiently basic or acidic may be converted into the corresponding diastereomeric salt forms by reaction with a suitable chiral acid, respectively chiral base. Said diastereomeric 15 salt forms are subsequently separated, for example, by selective or fractional crystallization and the enantiomers are liberated therefrom by alkali or acid. An alternative manner of separating the enantiomeric forms of the compounds of formula (I) involves liquid chromatography, in particular liquid chromatography using a chiral stationary phase. Said pure stereochemically isomeric forms may also be derived from the corresponding pure 20 stereochemically isomeric forms of the appropriate starting materials, provided that the reaction occurs stereospecifically. Preferably if a specific stereoisomer is desired, said compound will be synthesized by stereospecific methods of preparation. These methods will advantageously employ enantiomerically pure starting materials.
- The compounds of formula (I) show antiviral properties. Viral infections treatable using the compounds and methods of the present invention include those infections

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brought on by ortho- and paramyxoviruses and in particular by human and bovine respiratory syncytial virus (RSV).

The *in vitro* antiviral activity against RSV of the present compounds was tested in a test as described in the experimental part of the description, and may also be demonstrated in a virus yield reduction assay. The *in vivo* antiviral activity against RSV of the present compounds may be demonstrated in a test model using cotton rats as described in Wyde et al. (Antiviral Research (1998), 38, 31-42).

Due to their antiviral properties, particularly their anti-RSV properties, the compounds of formula (I) or any subgroup thereof, their prodrugs, N-oxides, addition salts, quaternary amines, metal complexes and stereochemically isomeric forms, are useful in the treatment of individuals-experiencing a viral infection, particularly a RSV infection, and for the prophylaxis of these infections. In general, the compounds of the present invention may be useful in the treatment of warm-blooded animals infected with viruses, in particular the respiratory syncytial virus.

The compounds of the present invention or any subgroup thereof may therefore be used as medicines. Said use as a medicine or method of treatment comprises the systemic administration to viral infected subjects or to subjects susceptible to viral infections of an amount effective to combat the conditions associated with the viral infection, in particular the RSV infection.

The present invention also relates to the use of the present compounds or any subgroup thereof in the manufacture of a medicament for the treatment or the prevention of viral infections, particularly RSV infection.

The compounds of the present invention or any subgroup thereof may be formulated into various pharmaceutical forms for administration purposes. As appropriate compositions there may be cited all compositions usually employed for systemically administering drugs. To prepare the pharmaceutical compositions of this invention, an effective amount of the particular compound, optionally in addition salt form or metal complex, as the active ingredient is combined in intimate admixture with a pharmaceutically acceptable carrier, which carrier may take a wide variety of forms depending on the form of preparation desired for administration. These pharmaceutical compositions are desirable in unitary dosage form suitable, particularly, for administration orally, rectally, percutaneously, or by parenteral injection. For example, in preparing the compositions in oral dosage form, any of the usual pharmaceutical

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media may be employed such as, for example, water, glycols, oils, alcohols and the like in the case of oral liquid preparations such as suspensions, syrups, elixirs, emulsions and solutions; or solid carriers such as starches, sugars, kaolin, lubricants, binders, disintegrating agents and the like in the case of powders, pills, capsules, and tablets.

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the skin.

- disintegrating agents and the like in the case of powders, pills, capsules, and tablets. Because of their ease in administration, tablets and capsules represent the most advantageous oral dosage unit forms, in which case solid pharmaceutical carriers are obviously employed. For parenteral compositions, the carrier will usually comprise sterile water, at least in large part, though other ingredients, for example, to aid solubility, may be included. Injectable solutions, for example, may be prepared in which the carrier comprises saline solution, glucose solution or a mixture of saline and glucose solution. Injectable suspensions may also be prepared in which case appropriate liquid carriers, suspending agents and the like may be employed. Also included are solid form preparations which are intended to be converted, shortly before use, to liquid form preparations. In the compositions suitable for percutaneous administration, the carrier optionally comprises a penetration enhancing agent and/or a suitable wetting agent, optionally combined with suitable additives of any nature in minor proportions, which additives do not introduce a significant deleterious effect on
- The compounds of the present invention may also be administered via oral inhalation or insufflation by means of methods and formulations employed in the art for administration via this way. Thus, in general the compounds of the present invention may be administered to the lungs in the form of a solution, a suspension or a dry powder, a solution being preferred. Any system developed for the delivery of solutions, suspensions or dry powders via oral inhalation or insufflation are suitable for the administration of the present compounds.
 - Thus, the present invention also provides a pharmaceutical composition adapted for administration by inhalation or insufflation through the mouth comprising a compound of formula (I) and a pharmaceutically acceptable carrier. Preferably, the compounds of the present invention are administered via inhalation of a solution in nebulized or aerosolized doses.

It is especially advantageous to formulate the aforementioned pharmaceutical compositions in unit dosage form for ease of administration and uniformity of dosage. Unit dosage form as used herein refers to physically discrete units suitable as unitary dosages, each unit containing a predetermined quantity of active ingredient calculated to produce the desired therapeutic effect in association with the required pharmaceutical carrier. Examples of such unit dosage forms are tablets (including scored or coated

tablets), capsules, pills, suppositories, powder packets, wafers, injectable solutions or suspensions and the like, and segregated multiples thereof.

In general it is contemplated that an antivirally effective daily amount would be from 0.01 mg/kg to 500 mg/kg body weight, more preferably from 0.1 mg/kg to 50 mg/kg body weight. It may be appropriate to administer the required dose as two, three, four or more sub-doses at appropriate intervals throughout the day. Said sub-doses may be formulated as unit dosage forms, for example, containing 1 to 1000 mg, and in particular 5 to 200 mg of active ingredient per unit dosage form.

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The exact dosage and frequency of administration depends on the particular compound of formula (I) used, the particular condition being treated, the severity of the condition being treated, the age, weight, sex, extent of disorder and general physical condition of the particular patient as well as other medication the individual may be taking, as is well known to those skilled in the art. Furthermore, it is evident that said effective daily amount may be lowered or increased depending on the response of the treated subject and/or depending on the evaluation of the physician prescribing the compounds of the instant invention. The effective daily amount ranges mentioned hereinabove are therefore only guidelines.

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Also, the combination of another antiviral agent and a compound of formula (I) can be used as a medicine. Thus, the present invention also relates to a product containing (a) a compound of formula (I), and (b) another antiviral compound, as a combined preparation for simultaneous, separate or sequential use in antiviral treatment. The different drugs may be combined in a single preparation together with pharmaceutically acceptable carriers. For instance, the compounds of the present invention may be combined with interferon-beta or tumor necrosis factor-alpha in order to treat or prevent RSV infections.

30 The following examples are intended to illustrate the present invention.

Experimental part

Hereinafter, "DMF" is defined as N,N-dimethylformamide, "DIPE" is defined as diisopropyl ether.

35 A. Preparation of the intermediate compounds

Example A1

a) Sodium methoxide (0.2 mol) was added to a mixture of N-(4-piperidinyl)-1H-benzimidazol-2-amine dihydrobromide (0.1 mol) in methanol (389ml), the mixture

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was cooled on an ice bath and stirred for 2 hours.

Di-tert-butyldicarbonate (0.1mol) was added to a cooled mixture on an ice bath and then stirred for 18 hours at room temperature. The mixture was evaporated and suspended in water/DIPE. The residue was filtered off, washed with water/DIPE and dried. The residue was boiled up in CH₃OH, yielding 17.46g (55.2%) of 1,1-dimethylethyl 4-(1*H*-benzimidazol-2-ylamino)-1-piperidinecarboxylate; mp. 249.4°C (interm. 1).

b) A mixture of intermediate (1) (0.05 mol), 2-(chloromethyl)quinoline monohydrochloride (0.055 mol) and sodium carbonate (0.075 mol) in DMF (250ml) was stirred at 55°C overnight. The solvent was evaporated. The residue was taken up in H₂O and CH₂Cl₂. The organic layer was separated, dried, filtered and the solvent was evaporated. The residue was purified by column chromatography over silica gel (eluent: CH₂Cl₂/CH₃OH 97/3 and 95/5). The pure fractions were collected and the solvent was evaporated. The residue was suspended in DIPE, filtered off and dried, yielding 13.5g (59%) of 1,1-dimethylethyl 4-[[1-(quinolinylmethyl)-1*H*-benzimidazol-2-yl]amino]-1-piperidinecarboxylate (interm. 2).

Example A2

- a) A mixture of 5,6,7,8-tetrahydro-2(1*H*)-quinoxalinone in phosphoryl chloride (200ml) was stirred and refluxed for 3 hours. The solvent was evaporated. The residue was taken up in ice and CH₂Cl₂. The mixture was basified with NH₄OH. The organic layer was separated, dried (MgSO₄), filtered and the solvent was evaporated, yielding 34g (86%) of 2-chloro-5,6,7,8-tetrahydroquinoxaline (interm. 3).
- b) A mixture of intermediate (3), 1-bromo-2,5-pyrolidinedione (0.116 mol) and dibenzoyl peroxide (1.3g) in tetrachloromethane (400ml) was stirred and refluxed for 35 minutes, brought to room temperature and then filtered. The reaction was carried out again using the same quantities. The residues were combined. The solvent was evaporated. The residue (60g) was purified by column chromatography over silica gel (eluent: cyclohexane/EtOAc 85/5; 15-35 μm). Two pure fractions were collected and their solvents were evaporated, yielding 25 g (43%) of (±)-5-bromo-2-chloro-5,6,7,8-tetrahydroguinovoline (interm. 4) and 12 g (21%) of (±) 8 bromo 2 chloro 5.6.7.8
- tetrahydroquinoxaline (interm. 4) and 12 g (21%) of (±)-8-bromo-2-chloro-5,6,7,8-tetrahydroquinoxaline.
 - c) A dispersion of sodium hydride in mineral oil (60%) (0.0518 mol) was added portionwise at 5°C under N_2 flow to a mixture of intermediate (1) (0.0471 mol) in DMF (200ml). The mixture was stirred at 5°C/10°C for 1 hour. A solution of intermediate (4) (0.0565 mol) in DMF (50ml) was added dropwise. The mixture was stirred at room temperature for 3 hours and poured out into H_2O . The precipitate was

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filtered off and taken up in CH₂Cl₂. The organic solution was dried (MgSO₄), filtered and the solvent was evaporated. The residue (32g) was purified by column chromatography over silica gel (eluent: CH₂Cl₂/CH₃OH/ NH₄OH 95/5/0.1; 20-45 µm). The pure fractions were collected and the solvent was evaporated, yielding 13.3g (58%) of (±)-1,1-dimethylethyl 4-[[1-(2-chloro-5,6,7,8-tetrahydro-5-quinoxalinyl)-1*H*-benzimidazol-2-yl]amino]-1-piperidinecarboxylate (interm. 5). Example A3

- a) 2,3-Butanedione (0.0776 mol) was added at room temperature to a solution of sodium pyrosulfite (0.1 mol) in water (75ml). The mixture was heated to 70°C and then added to a solution of ethyl 2,3-diaminobenzoate (0.0776 mol) in water (75ml). The mixture was stirred at 100°C for 12 hours, cooled, basified with K₂CO₃ 10% and extracted with EtOAc. The organic layer was separated, dried (MgSO₄), filtered and the solvent was evaporated. The residue (17.5g) was purified by column chromatography over silica gel (eluent: CH₂Cl₂/EtOAc 93/7; 20-45 µm). The pure fractions were collected and the solvent was evaporated, yielding 12g (67%) of ethyl 2,3-dimethyl-5-quinoxalinecarboxylate (interm. 6).
- b) Lithium chloride (0.6 mol) was added portionwise at 80°C to a mixture of intermediate (6) (0.06 mol) and potassium tetrahydroborate (0.6 mol) in tetrahydrofuran (300ml). The mixture was stirred at 80°C for 5 hours, cooled, poured out into H₂O and extracted with EtOAc. The organic layer was separated, washed with H₂O, dried (MgSO₄), filtered and the solvent was evaporated, yielding 10.5g (91%) of (±)-1,2,3,4-tetrahydro-2,3-dimethyl-5-quinoxaline-methanol (interm. 7).
- c) MnO₂ (100g) was added portionwise at room temperature to a mixture of intermediate (7) (0.0546 mol) in dichloromethane (500ml). The mixture was stirred at room temperature overnight, filtered over celite, washed with CH₂Cl₂ and the filtrate was evaporated. The product was used without further purification, yielding 7.8g (77%) of 2,3-dimethyl-5-quinoxalinecarboxaldehyde (interm. 8).
- d) Sodium tetrahydroborate (0.084 mol) was added portionwise at 5°C to a mixture of intermediate (8) (0.042 mol) in methanol (100ml). The mixture was stirred at 5°C for 30 minutes, hydrolized cold and extracted with EtOAc. The organic layer was separated, dried (MgSO₄), filtered and the solvent was evaporated, yielding 6.7g (85%) 2,3-dimethyl-5-quinoxalinemethanol (interm. 9).
- e) Thionyl chloride (0.045 mol) was added dropwise at 5°C to a mixture of intermediate (9) (0.03 mol) in dichloromethane (50ml). The mixture was stirred at room temperature for 2 hours, poured out on ice and K₂CO₃ 10%. The organic layer was separated, washed with K₂CO₃ 10%, dried (MgSO₄), filtered and the solvent was

evaporated. The product was used without further purification, yielding 6.2g (quant.) of 5-(chloromethyl)-2,3-dimethyl-quinoxaline (interm. 10).

f) A dispersion of sodium hydride in mineral oil (60%) (0.021 mol) was added portionwise at 5°C under N₂ flow to a mixture of intermediate (1) (0.02 mol) in DMF (30ml). The mixture was stirred at 5°C under N₂ flow for 1 hour. A solution of intermediate (10) (0.03 mol) in a small amount of DMF was added dropwise at 5°C. The mixture was stirred at room temperature under N₂ flow for 2 hours, hydrolized and extracted with EtOAc. The organic layer was separated, washed several times with H₂O, dried (MgSO₄), filtered and the solvent was evaporated. The residue (12.5g) was purified by column chromatography over silica gel (eluent: CH₂Cl₂/ CH₃OH/NH₄OH 97.5/2.5/0.1; 20-45 μm). Two pure fractions were collected and their solvents were evaporated, yielding 7.8g (80%) of 1,1-dimethylethyl 4-[[1-[(2,3-dimethyl-5-quino-xalinyl)methyl]-1*H*-benzimidazol-2-yl]amino]-1-piperidinecarboxylate (interm. 11).

Example A4

8-Bromo-2-methylquinoline (0.0675 mol) was added portionwise at -70°C under N₂ flow to a mixture of a solution of butyllithium in hexane (1.6M) (0.135 mol) in tetrahydrofuran (300ml) and diethyl ether (300ml). The mixture was stirred for 30 minutes. A solution of DMF (0.405 mol) in tetrahydrofuran (100ml) was added quickly. The mixture was cooled to -70°C and stirred for 15 minutes. Ethanol (70ml) and a NH₄Cl solution 10% were added. The mixture was brought to room temperature and stirred for 15 minutes. NH₄Cl was added. The mixture was extracted with EtOAc. The organic layer was separated, washed with H₂O, dried (MgSO₄), filtered and the solvent was evaporated. The product was used without further purification, yielding 15g (>100%) of 2-methyl-8-quinolinecarboxaldehyde (interm. 12).

25 Example A5

a) A mixture of 3-methoxy-2-methylquinoline (0.081 mol) in trifluoro-acetic acid (150ml) was hydrogenated at room temperature under a 3-4 bar pressure for 48 hours with palladium on activated carbon (2g) as a catalyst. After uptake of hydrogen (2 equiv.), the catalyst was filtered through celite and washed with H₂O. The filtrate was basified with a concentrated NH₄OH solution and extracted with CH₂Cl₂. The organic layer was separated, dried (MgSO₄), filtered and the solvent was evaporated, yielding 14.3g (quant.) of 5,6,7,8-tetrahydro-3-methoxy-2-methylquinoline (interm. 13). b) 3-Chlorobenzenecarboperoxoic acid (0.1 mol) was added portionwise at 5°C to a mixture of intermediate (13) (0.067 mol) in dichloromethane (300ml). The mixture was stirred at room temperature overnight, basified with K₂CO₃ 10% and separated into its layers. The aqueous layer was extracted with CH₂Cl₂. The combined organic

layer was dried (MgSO₄), filtered and the solvent was evaporated, yielding 13.7g (quant.) of 5,6,7,8-tetrahydro-3-methoxy-2-methylquinoline, 1-oxide (interm. 14). c) A mixture of intermediate (14) (0.067 mol) in acetic anhydride (100ml) was stirred at 90°C for 1 hour, poured out on ice and basified with NaOH 3N. CH₂Cl₂ was added.

- The organic layer was separated, washed with a diluted NaOH solution, dried (MgSO₄), filtered and the solvent was evaporated, yielding 16.8g (quant.) of 5,6,7,8-tetrahydro-3-methoxy-2-quinolinemethanol acetate (ester) (interm. 15).

 d) A mixture of intermediate (15) (0.067 mol) and sodium hydroxide (13g) in methanol (60ml) was stirred and refluxed for 20 minutes, poured out on ice and extracted with CH₂Cl₂. The organic layer was separated, dried (MgSO₄), filtered and the solvent was evaporated, yielding 12.3g (95%) of 5,6,7,8-tetrahydro-3-methoxy-2-quinolinemethanol (interm. 16).
 - In a similar way was also prepared (±)-5,6,7,8-tetrahydro-2-methyl-8-quinolinol (interm. 17).

15 Example A6

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Phosphorus tribromide (0.0105 mol) was added dropwise at 0°C/5°C under N₂ flow to a mixture of (±)-5,6,7,8-tetrahydro-2-methyl-8-quinolinol (intermediate 17) (0.03 mol) in toluene (20ml). The mixture was brought to room temperature and stirred at room temperature overnight. Ice water was added. The mixture was basified with a concentrated NaOH solution and extracted with CH₂Cl₂. The organic layer was separated, dried (MgSO₄), filtered and the solvent was evaporated. The residue (6g) was purified by column chromatography over silica gel (eluent: CH₂Cl₂/CH₃OH 99/1; 20-45 μm). The pure fractions were collected and the solvent was evaporated, yielding 2g (29%) of (±)-8-bromo-5,6,7,8-tetrahydro-2-methylquinoline (interm. 18).

25 Example A7

- a) A mixture of N-2,6-dimetyl-2,3-pyridinediamine (0.122 mol) in trifluoro-acetic acid (250ml) was stirred and refluxed for 6 hours and brought to room temperature. The solvent was evaporated. The residue was taken up in CH_2Cl_2 and K_2CO_3 10%. The organic layer was separated, dried (MgSO₄), filtered and the solvent was evaporated.
- The residue (32g) was purified by column chromatography over silica gel (eluent: CH₂Cl₂/EtOAc 97/3; 20-45 μm). The pure fractions were collected and the solvent was evaporated. The residue was taken up in petroleum ether. The precipitate was filtered off and dried, yielding 15g of residue (fraction 1). The mother layer was evaporated. The residue was combined with 14.1g of fraction 1, yielding 28.9 g of
- 35 1,6-dimethyl-2-(trifluoromethyl)-1*H*-imidazo[4,5-b]pyridine; mp. 100°C (interm. 19).

- b) 1-Bromo-2,5-pyrolidinedione (0.0735 mol) and dibenzoyl peroxide (1.5g) were added at room temperature to a solution of intermediate (19) (0.07 mol) in tetrachloromethane (450ml). The mixture was stirred and refluxed for 7 hours, then brought to room temperature and filtered. The reaction was carried out again using the same quantities. The mixtures were combined. The solvent was evaporated. The residue (50g) was purified by column chromatography over silica gel (eluent: CH₂Cl₂/CH₃OH 100/0 and 98/2; 20-45 μm). The pure fractions were collected and the solvent was evaporated, yielding 20.2g (49%) of 6-(bromomethyl)-1-methyl-2-(trifluoromethyl)-1*H*-imidazo[4,5-b]pyridine (interm. 20).
- c) A mixture of ethyl 4-(1*H*-benzimidazol-2-ylamino)-1-piperidine-carboxylate (0.0464 mol), intermediate (20) (0.051 mol) and potassium carbonate (0.1392 mol) in acetonitrile (250ml) was stirred and refluxed for 90 minutes and then brought to room temperature. Water was added and the mixture was extracted twice with CH₂Cl₂. The combined organic layer was dried (MgSO₄), filtered and the solvent was evaporated.
- The product was used without further purification, yielding 23g (>100%) of ethyl 4- [[1-[[1-methyl-2-(trifluoromethyl)-1*H*-imidazo[4,5-b]pyridin-6-yl]methyl]-1*H*-benzimidazol-2-yl]amino]-1-piperidinecarboxylate (interm. 21).

Example A8

A mixture of ethyl 4-(1*H*-benzimidazol-2-ylamino)-1-piperidine-carboxylate (0.0289 mol), 7-chloro-6,7-dihydro-5*H*-cyclopenta[b]pyridine (0.0289 mol) and potassium carbonate (0.0867 mol) in acetonitrile (250ml) was stirred and refluxed for 48 hours and then brought to room temperature. The reaction was carried out again using the same quantities. The mixtures were combined, poured out into H₂O and extracted with EtOAc. The organic layer was separated, washed with H₂O, dried (MgSO₄), filtered and the solvent was evaporated. The residue (25g) was purified by column chromatography over silica gel (eluent: CH₂Cl₂/ CH₃OH/NH₄OH 97/3/0.5; 20-45 μm). Two fractions were collected and their solvents were evaporated, yielding 8g of ethyl 4-[[1-(6,7-dihydro-5*H*-1-pyrindin-7-yl)-1*H*-benzimidazol-2-yl]amino]-1-piperidinecarboxylate (interm. 22).

30 Example A9

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a) A dispersion of sodium hydride in mineral oil (0.261 mol) was added portionwise at room temperature under N_2 flow to a mixture of N-8-quinolinylformamide (0.174 mol) in DMF (500ml). The mixture was stirred at room temperature for 1 hour. A solution of 1-chloro-2-nitrobenzene (0.53 mol) in DMF (200ml) was added dropwise. The mixture was stirred at 140°C for 12 hours and then brought to room temperature. H_2O was added and the mixture was extracted with CH_2Cl_2 . The organic layer was

separated, dried (MgSO₄), filtered and the solvent was evaporated. The residue (110g) was purified by column chromatography over silica gel (eluent: $CH_2Cl_2/cyclohexane$ 80/20; 20-45 µm). The pure fractions were collected and the solvent was evaporated, yielding 9.8g (21%) of N-(2-nitrophenyl)-8-quinolinamine (interm. 23).

- b) A mixture of 6-quinolinemethanamine (0.074 mol), 2-chloro-3-nitropyridine (0.0888 mol) and potassium carbonate (0.185 mol) in acetronitrile (200ml) was stirred and refluxed for 5 hours and then cooled to room temperature. EtOAc and H₂O were added. The mixture was extracted with HCl 3N. The aqueous layer was basified with K₂CO₃ solid and extracted with CH₂Cl₂. The combined organic layer was dried
- 10 (MgSO₄), filtered and the solvent was evaporated, yielding 17.8g (84%) of *N*-(3-nitro-2-pyridinyl)-8-quinolinemethanamine (interm. 24).

Example A10

- a) A mixture of intermediate (24) (0.064 mol) in methanol (200ml) was hydrogenated under a 3 bar pressure for 2 hours with Raney nickel (10g) as a catalyst. After uptake
- of hydrogen (3 equiv), the catalyst was filtered through celite and the filtrate was evaporated, yielding 14.8g (93%) of N2-(8-quinolinylmethyl)-2,3-pyridinediamine (interm. 25).
 - b) A mixture of intermediate (25) (0.059 mol) and ethyl 4-isothiocyanato-1-piperidine-carboxylate (0.059 mol) in methanol (150ml) was stirred and refluxed for 4 hours and brought to room temperature. The solvent was evaporated. The residue was purified by column chromatography over silica gel (eluent: CH₂Cl₂/CH₃OH 97/3; 20-45 μm). The desired fractions were collected and the solvent was evaporated, yielding 10.5g
 - (37%) of ethyl 4-[[[[2-[(8-quinolinylmethyl)amino]-3-pyridinyl]amino]sulfinyl]-amino]-1-piperidine-carboxylate (interm. 26)
- c) A mixture of intermediate (26) (0.026 mol), mercury(II) oxide (0.052 mol) and sulfur (0.2g) in ethanol (120ml) was stirred and refluxed for 2 hours, brought to room temperature and filtered over celite. The filtrate was evaporated, yielding 8.7g (96%) of 4-[[1-(8-quinolinylmethyl)-1*H*-imidazo[4,5-b]pyridin-2-yl]amino]-1-piperidine-carboxylate (interm. 27).

30 Example A11

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a) A mixture of 8-quinolinecarboxaldehyde (0.092 mol) and 4-methylbenzenesulfonic acid (0.3g) in 2-ethoxyethanol (110ml) was stirred and refluxed for 24 hours using a Dean Stark apparatus. The solvent was evaporated. The reaction was carried out again using the same quantities. The residues were combined and taken up in CH₂Cl₂. The organic solution was washed with K₂CO₃ 10% and decanted. The organic layer was dried (MgSO₄), filtered and the solvent was evaporated. The residue (41g) was

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purified by column chromatography over silica gel (eluent: CH₂Cl₂/CH₃OH 98/2; 20-45 μm). Two pure fractions were collected and their solvents were evaporated, yielding 20g (34%) of 8-[bis(2-ethoxyethoxy)methyl]quinoline (interm. 28). b) A mixture of 8-quinolinecarboxaldehyde (0.248 mol), triethoxymethane (0.4464 mol) and 4-methylbenzenesulfonic acid (4g) in ethanol (250ml) was stirred and refluxed for 1 hour, brought to room temperature, poured out into K₂CO₃ 10% and extracted with EtOAc. The organic layer was separated, dried (MgSO₄), filtered and the solvent was evaporated. The product was used without further purification, yielding 48.5g (80%) of 8-(diethoxymethyl)-quinoline (interm. 29).

c) A mixture of 2-quinolinecarboxaldehyde (0.08 mol) and 4-methylbenzenesulfonic acid (0.25g) in ethanol (100ml) was stirred and refluxed for 48 hours and brought to room temperature. The reaction was carried out again using the same quantities. The mixtures were combined. The solvent was evaporated. The residue was taken up in CH₂Cl₂. The organic solution was washed with K₂CO₃ 10% and with H₂O, then dried (MgSO₄), filtered and the solvent was evaporated. The product was used with seven

(MgSO₄), filtered and the solvent was evaporated. The product was used without further purification, yielding 32.5g of 2-(diethoxymethyl)quinoline (interm. 30).

Example A12

Intermediate (1) (0.0377 mol) and intermediate (29) (0.0755 mol) were heated at 160° C for 1 hour and then purified by column chromatography over silica gel (eluent: CH₂Cl₂/CH₃OH 98/2; 15-35 µm). The pure fractions were collected and the solvent was evaporated, yielding 15g (79%) of (±)-1,1-dimethylethyl 4-[[1-[ethoxy(8-quino-linyl)methyl]-1*H*-benzimidazol-2-yl]amino]-1-piperidinecarboxylate (interm. 31).

Example A13

4-Methylbenzenesulfonyl chloride (0.2222 mol) was added portionwise at 10°C to a mixture of 1,1-dimethylethyl [1-(hydroxymethyl)-2-methylpropyl]carbamic acid (ester) (0.202 mol) in pyridine (65ml). The mixture was stirred at 10°C for 2 hours. H₂O (75ml) was added at 10°C. The precipitate was filtered off, washed with H₂O and taken up in CH₂Cl₂. The organic solution was washed with H₂O, dried, filtered and the solvent was evaporated, yielding 49g (68%) of (±)-1,1-dimethylethyl [1-[[(4-methylphenyl)sulfonyl]oxy]methyl]-2-methylpropyl]carbamate; mp. 85°C(interm. 32).

Example A14

a) A mixture of compound (33) (0.0347 mol), 1-bromo-3-methyl-2-butanone (0.052 mol) and potassium carbonate (0.104 mol) in acetonitrile (255ml) was stirred and refluxed for 2 hours and filtered. The filtrate was evaporated. The residue was taken up in H_2O and the mixture was extracted with EtOAc. The organic layer was separated, dried (MgSO₄), filtered and the solvent was evaporated. The product was

used without further purification, yielding 16.84g of (±)-1-[4-[[1-[ethoxy(8-quinolinyl)methyl]-1*H*-benzimidazol-2-yl]amino]-1-piperidinyl]-3-methyl-2-butanone (interm. 34) (quant.).

In a similar way were also prepared:

- 5 1-[4-(1*H*-benzimidazol-2-ylamino)-1-piperidinyl]-3-methyl-2-butanone; 1-[4-[[1-(8-quinolinyl)-1*H*-benzimidazol-2-yl]amino]-1-piperidinyl]-3-methyl-2-butanone; and 1-[4-[[1-(2-quinolinylmethyl)-1*H*-benzimidazol-2-yl]amino]-1-piperidinyl]-3-methyl-2-butanone.
- b) A mixture of intermediate (34) (0.036 mol) in methanol (200ml) was stirred at 10°C. Sodium tetrahydroborate (0.04 mol) was added portionwise. The mixture was stirred for 90 minutes. H₂O was added. The solvent was evaporated. The residue was extracted with CH₂Cl₂. The organic layer was separated, washed with H₂O, dried (MgSO₄), filtered and the solvent was evaporated, yielding 17g (96%) of (±)-4-[[1-
- 15 [ethoxy(8-quinolinyl)methyl]-1*H*-benzimidazol-2-yl]amino]-alpha-(1-methylethyl)-1-piperidineethanol (interm. 35).
 - c) Diethyl azodicarboxylate (0.015 mol) was added dropwise at 0° C under N_2 flow to a solution of intermediate (35) (0.01 mol), phthalimide (0.015 mol) and triphenylphosphine (0.015 mol) in tetrahydrofuran (100ml). The mixture was stirred at room
- temperature for 2 hours. EtOAc was added. The mixture was extracted with HCl 3N and separated into its layers. The aqueous layer was washed twice with EtOAc, basified with K₂CO₃ solid and extracted with CH₂Cl₂. The combined organic layer was dried (MgSO₄), filtered and the solvent was evaporated. The residue was purified by column chromatography over silica gel (eluent: CH₂Cl₂/CH₃OH/NH₄OH 97/3/0.2;
- 25 20-45 μ m). Two pure fractions were collected and their solvents were evaporated, yielding 2.3g (30%) of (±)-2-[2-[4-[[1-[ethoxy(8-quinolinyl)methyl]-1*H*-benzimidazol-2-yl]amino]-1-piperidinyl]-3-methylbutyl]-1*H*-isoindole-1,3(2*H*)dione (interm.

A solution of

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and Et₃N (0.072 mol) in CH₂Cl₂ (100ml) was cooled to 0°C under N₂ flow. A mixture of methanesulfonyl chloride (0.036 mol) in CH₂Cl₂ (a small amount) was added dropwise. The mixture was allowed to cool to room temperature while stirring for 3 hours. Water was added. The mixture was decanted. The organic layer was dried (MgSO₄), filtered and the solvent was evaporated, yielding 8.5g of intermediate (80) (86%).

e) Preparation of intermediate

A solution of 1*H*-isoindole-1,3(2*H*)-dione (0.0828 mol) in DMF (80ml) was cooled to 10°C. NaH 60% in oil (0.0828 mol) was added portionwise. The mixture was allowed to cool to room temperature while stirring for 1 hour. A mixture of intermediate (80) (0.0207 mol) (prepared according to A14d) in DMF (a small amount) was added dropwise. The mixture was stirred at room temperature for 1.5 hours, at 60°C for 5 hours and at room temperature for the weekend. The residue (9.6g) was crystallized from diethyl ether and CH₃CN. The precipitate was filtered off and dried, yielding 4g of intermediate (81) (42%).

Example A15

- a) A mixture of 1-[4-(1*H*-benzimidazol-2-ylamino)-1-piperidinyl]-3-methyl-2-butanone (0.03 mol) and benzenemethanamine (0.09 mol) in methanol (200ml) was hydrogenated at 40°C under a 3 bar pressure for 48 hours with palladium on activated carbon (1.3g) as a catalyst. After uptake of hydrogen, the catalyst was filtered through celite, washed with CH₃OH and the filtrate was evaporated. Hydrogenation was continued for 24 hours. After uptake of hydrogen, the catalyst was filtered through celite, washed with CH₃OH and the filtrate was evaporated. The residue was purified by column chromatography over silica gel (eluent: CH₂Cl₂/CH₃OH/NH₄OH 85/14/1; 20-45 μm). The desired fractions were collected and the solvent was evaporated. The residue was crystallized from diethyl ether. The precipitate was filtered off and dried,
- yielding 0.4g of (±)-N-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-1H-benzimidazol-2-amine; mp. 138°C (interm. 37).
 b) Di-tert-butyl dicarbonate (0.02 mol) was added at 5°C to a mixture of intermediate
- (37) (0.0186 mol) in dichloromethane (60ml). The mixture was stirred at room temperature for 3 hours and poured out into H₂O. The organic layer was separated, dried (MgSO₄), filtered and the solvent was evaporated. The product was used without

further purification, yielding 5.9g of (±)-1,1-dimethylethyl [1-[[4-[[1-[(1,1-dimethylethoxy)carbonyl]-1*H*-benzimidazol-2-yl]amino]-1-piperidinyl]methyl]-2-methyl-propyl]carbamate (interm. 38).

Example A16

- A mixture of 1-[4-[[1-(8-quinolinyl)-1*H*-benzimidazol-2-yl]amino]-1-piperidinyl]3-methyl-2-butanone (0.0222 mol) and benzenemethanamine (0.0666 mol) in methanol
 (250ml) was hydrogenated at 40°C under a 3 bar pressure for 24 hours with palladium
 on activated carbon (1.5g) as a catalyst. After uptake of hydrogen, the catalyst was
 filtered through celite, washed with CH₂Cl₂ and CH₃OH and the filtrate was
- evaporated. Palladium on activated carbon (1.5g) and methanol (250ml) were added again. Hydrogenation was continued at 40°C under a 3 bar pressure for 24 hours. After uptake of hydrogen, the catalyst was filtered through celite, washed with CH₂Cl₂ and the filtrate was evaporated. The residue (22g) was purified by column chromatography over silica gel (eluent: CH₂Cl₂/CH₃OH/ NH₄OH 95/5/0.1 and
- 85/15/1; 20-45 μm). Three pure fractions were collected and their solvents were evaporated, yielding 2.6g 1-[4-[[1-(1,2,3,4-tetrahydro-8-quinolinyl)-1*H*-benzimidazol-2-yl]amino]-1-piperidinyl]-3-methyl-2-butanone (interm. 40) (fraction 1), 2.9g of fraction 2 and 0.7g of fraction 3. Fraction 2 and 3 were crystallized from CH₃CN. The precipitate was filtered off and dried, yielding 0.82g (±)-N-[1-[3-methyl-2-
- [(phenylmethyl)amino]butyl]-4-piperidinyl]-1-(1,2,3,4-tetrahydro-8-quinolinyl)-1*H*-benzimidazol-2-amine; mp. 126°C and 0.55g of *N*-(4-piperidinyl)-1-(1,2,3,4-tetrahydro-8-quinolinyl)-1*H*-benzimidazol-2-amine; mp. 205°C (comp. 48).

Example A17

- a) A mixture of N-(4-piperidinyl)-1-(4-quinolinylmethyl)-1H-benzimidazol-2-amine (comp. 23) (0.0129 mol), chloroacetonitrile (0.0155 mol), potassium iodide (0.00129 mol) and potassium carbonate (0.0258 mol) in 4-methyl-2-pentanone (80ml) was stirred and refluxed for 5 hours. H₂O was added. The solvent was evaporated. H₂O and CH₂Cl₂ were added. The precipitate was filtered off. The filtrate was separated into its layers. The organic layer was dried (MgSO₄), filtered and the solvent was
- evaporated. The residue (3.5g) was purified by column chromatography over silica gel (eluent: CH₂Cl₂/ CH₃OH/NH₄OH 95/5/0.3; 15-40 μm). The pure fractions were collected and the solvent was evaporated. The residue was crystallized from CH₃CN. The precipitate was filtered off and dried, yielding 0.94g 4-[[1-(4-quinolinylmethyl)-1*H*-benzimidazol-2-yl]amino]-1-piperidineacetonitrile; mp. 190°C (interm. 41).
- b) A mixture of N-(4-piperidinyl)-[1,2'-bi-1H-benzimidazol]-2-amine (comp. 71) (0.01 mol), chloroacetonitrile (0.01 mol) and sodium hydrogen carbonate (0.02 mol) in

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DMF (50ml) was stirred at 50°C overnight. The solvent was evaporated. The residue was taken up in H₂O and the mixture was extracted with CH₂Cl₂. The organic layer was separated, dried, filtered and the solvent was evaporated. The residue was suspended in DIPE, filtered off and dried, yielding 2.3g (63%) of product. This 5 fraction was purified over silica gel on a glass filter (eluent: CH2Cl2/(CH3OH/NH3) 97/3). The pure fractions were collected and the solvent was evaporated, yielding 1.36g (37%) of 4-[(1,2'-bi-1H-benzimidazol-2-yl)amino]-1-piperidine-acetonitrile (interm. 42).

Example A18

Preparation of intermediate

- 10 A mixture of 2-chloro-1*H*-benzimidazole (0.0189 mol) and 1,1-dimethylethyl 2-aminocyclohexanecarbamoate (0.04725 mol) (prepared according to A1a))was stirred at 140°C for 3 hours, then brought to room temperature and taken up in CH₂Cl₂/CH₃OH. The same procedure was repeated 3 times on the same quantities of 2-chloro-1*H*-benzimidazole and 1,1-dimethylethyl 2-aminocyclohexanecarbamoate.
- 15 The mother layers were brought together, dried (MgSO₄), filtered and the solvent was evaporated. The residue (28g) was purified by column chromatography over silica gel (eluent: CH₂Cl₂/CH₃OH/ NH₄OH 96/4/0.1; 15-35μm). Two fractions were collected and the solvent was evaporated, yielding 4.5g of intermediate (84) (24%).

Example A19

Preparation of intermediate

20 A mixture of quantities of ethyl 4-(1H-benzimidazol-2-ylamino)-1-piperidinecarboxylate (0.0154 mol), (0.0154 mol) (prepared

according to A14d) and K₂CO₃ (0.0463 mol) in CH₃CN (50ml) and DMF (5ml) was stirred and refluxed for 6 hours, poured out into H₂O and extracted with EtOAc. The organic layer was separated, dried (MgSO₄), filtered and the solvent was evaporated. The residue was purified by column chromatography over silica gel (eluent:

25 CH₂Cl₂CH₃OH 97/3; 35-70µm). The pure fractions were collected and the solvent was evaporated, yielding: 0.87g of intermediate (76) (13%).

Example A20

a) Preparation of intermediate

A solution of

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A1b) in HCl 6N (60ml) was stirred and refluxed for 12 hours and then brought to room temperature. The solvent was evaporated. The residue was taken up in 2-propanol. The precipitate was filtered off, washed with CH₃CN, washed with diethyl ether and dried, yielding: 4g of intermediate (82) (94%).

b) Preparation of intermediate

Intermediate (82 (0.0094 mol) was added at room temperature to CH₂Cl₂ (70ml). Et₃N (0.0188 mol) was added. 1,1'-carbonylbis-1*H*-imidazole (0.0188 mol) was added. The mixture was stirred at room temperature for 4.5 hours. (Methylamino)acetonitrile .HCl (0.0188 mol) was added. The mixture was stirred at room temperature for 12 hours. The organic layer was separated, washed twice with water, dried (MgSO₄), filtered and the solvent was evaporated. The residue was purified by column chromatography over silica gel (eluent: CH₂Cl₂/CH₃OH 98.5/1.5; 35-70 µm). The pure fractions were collected and the solvent was evaporated. The residue (2.2g) was crystallized from CH₃CN. The precipitate was filtered off and dried, yielding: 1.5g of intermediate (83) (41%).

Example A21

A mixture of intermediate

(prepared according to A1b) in HCl 3 N (200ml) was stirred and refluxed for 1 hour. The solvent was evaporated. The residue was taken up in EtOAc and NH4OH. The mixture was stirred for 30 minutes and filtered. The solvent was evaporated. The product was used

without further purification, yielding 14g of

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Tables 1, 2 and 3 list intermediates which were prepared analogous to one of the above examples.

Table 1

Int. No.	Ex. No.	Rª	R ^b	R ^c	n	а	*	b	R ^d	Re	R ^f	R ^g
43	A10c	Н	H	Н	1	N	2	С	-	Н	Н	Н
44	A12	CH ₃	H	O(CH ₂) ₂ OC ₂ H ₅	1	СН	8	С	H	Н	Н	-
45	A12	CH ₃	H	O(CH ₂) ₂ OC ₂ H ₅	1	СН	2	C	-	H	Н	Н
46	А7с	CH ₃	H	Н	1	CH	2	N	. .	OCH ₃	-	н
47	A7c	Н	H	н .	1	СН	2	С	-	H	н	Cl
48	A7c	Н	Н	Н	1	СН	2	C	-	Н	CI	Н
49	A7c	H	Н	Н	1	CH	2	C	-	H	Н	Н
2	Alb	CH ₃	Н	Н	1	СН	2	C	-	Н	Н	Н
50	A12	CH ₃	CH ₃	OC ₂ H ₅	1	CH	8	С	H	H	Н	-
51	A12	CH ₃	Н	OC ₂ H ₅	1	СН	2	С	-	H	Н	Н
52	A12	CH ₃	H	OC ₂ H ₅	1	СН	2	Ċ	-	OCH ₃	H	Н
31	A12	CH ₃	H	OC ₂ H ₅	1	CH	8	С	Н	H	Н	-
53	A3f	Н	Н	Н	1	CH	8	С	Н	H	Н	_]
54	A3f	CH ₃	H	Н	1	СН	8	N	Н	Н	-	

Int. No.	Ex. No.	Rª	R ^b	R ^c	n	a	*	b	R ^d	Re	R ^f	R ^g
55	A7c	CH ₃	Н	Н	1	СН	8	С	CH ₃	Н	Н	-
11	A3f	СН3	H	Н	1	СН	8	N	CH₃	CH ₃	-	-
56	A7c	H	Н	H	1	CH	4	С	н	н	-	Н
57	A7c	H	CH ₃	H	1	CH	8	С	Н	Н	н	-
27	A10c	Н	н	H	1	N	8	С	Н	н	Н	-
58	A10c	Н	Н	-	0	СН	8	С	Н	H	Н	-
66	A12	CH ₃	CH ₃	$O(C_2H_5)OC_2H_5$	1	CH	8	С	Н	н	Н	-
67	A12	CH ₃	Н	$O(C_2H_5)OC_2H_5$	1	CH	8	С	Н	н	Н	-
68	A1b	CH ₃	CH ₃	CH ₃	1	CH	8	С	Н	Н	H	-
69	A1b	CH ₃	Н	H	1	СН	2	С	-	OCH ₃	H	Н
7.0	Alb	CH₃	H	H	1	CH	2	N -		Н	-	Н
71	Alb	CH ₃	Н	H	1	СН	8	С	OCH ₃	H	Н	-

^{* =} position bicyclic heterocycle

Table 2

					K
Int. No.	Ex. No.	Rª	R ^b	n	L
59	A2c	СН3	Н	0	
60	A8	н	Н	0	
61	A2c	Н	H	0	N N N N N N N N N N N N N N N N N N N
5	A2c	СН3	Н	0	N CI
21	A7c	Н	Н	1	N CF_3 CH_3

Int. No.	Ex. No.	Rª	R ^b	n	L
62	A3f	СН3	Н	1	N OCH3
63	A7c	СН3	Н	1	
64	A7c	Н	н	1	
65	A2c	СН3	Н	0	
22	A8	Н	Н	0	
72	A2c	CH ₃	СН₃	0	N CI
73	A2c	CH ₃	СН₃	0	N CI
74	A2c	CH₃	СН₃	0	N
75	A2c	CH ₃	СН3	0	N CH ₃
76	A19	H	Н	1	

Table 3

2.5 ...

Int.		L	Physical data
No.	No.		
77	Alb	NH C(C(H ₃) ₃	
78	A1b	N C(CH ₃) ₃	
79	Alb	H O C(CH ₃) ₃	trans
80	A14d	ON CH3	
81	A14e	СН3	
82	A20	но он	
83	A20	N CH ₃	

B. Preparation of the final compounds

- a) A mixture of 2-propanol and hydrochloric acid (15ml) was added to a mixture of intermediate (2) (0.0284 mol) in 2-propanol (150ml). The mixture was stirred and refluxed for 90 minutes and cooled. The precipitate was filtered off, washed with 2-propanol and DIPE and dried, yielding 10.36g of N-(4-piperidinyl)-1-(2-quinolinyl-methyl)-1H-benzimidazol-2-amine dihydrochloride (comp.1).
- b) A mixture of compound (1) (0.01 mol) and sodium carbonate (0.03 mol) in
 4-methyl-2-pentanone (250ml) was stirred and refluxed for a few hours using a water separator (until gas development stops). 2-Bromoethyl carbamic acid 1,1-dimethylethyl ester (0.015 mol) was added. The mixture was stirred and refluxed for 18 hours using a water separator, then cooled, washed with H₂O, dried, filtered and the solvent was evaporated. The residue was purified by column chromatography over silica gel
 (eluent: CH₂Cl₂/C₂H₅OH 95/5 and 90/10). The pure fractions were collected and the solvent was evaporated, yielding 3.8g of 1,1-dimethylethyl [2-[4-[[1-(2-quinolinylmethyl)-1H-benzimidazol-2-yl]amino]-1-piperidinyl]ethyl]carbamate (comp. 2).
 c) A mixture of compound (2) (0.0076 mol) in a mixture of 2-propanol and
 - c) A mixture of compound (2) (0.0076 mol) in a mixture of 2-propanol and hydrochloric acid (10ml) and 2-propanol (100ml) was stirred and refluxed for 1 hour

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and then cooled. The precipitate was filtered off, washed with 2-propanol and DIPE and dried, yielding 3.08g of N-[1-(2-aminoethyl)-4-piperidinyl]-1-(2-quinolinyl-methyl)-1H-benzimidazol-2-amine tetrahydrochloride monohydrate (comp. 3).

- d) A mixture of compound (115) (0.00305 mol) in HBr/HOAc 33% (34ml) was stirred at room temperature for 2 hours, poured out on ice, basified with a concentrated NH₄OH solution and extracted with CH₂Cl₂. The organic layer was separated, dried (MgSO₄), filtered and the solvent was evaporated. The residue was purified by column chromatography over silica gel (eluent: CH₂Cl₂/CH₃OH/NH₄OH 96/4/0.2; 15-40 μm). Two fractions (F1 and F2) were collected and their solvents were evaporated, yielding
- 0.56g F1 (46%) and 0.69g F2 (50%). F1 was crystallized from diethyl ether. The precipitate was filtered off and dried, yielding 0.27g of (±)-N-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-4-methyl-1-(8-quinolinylmethyl)-1H-benzimidazol-2-amine (comp. 116).
- e) A mixture of compound (155) (0.0024 mol) in CH₃OH (3ml) and 2-propanol (15ml)
 15 was stirred and refluxed for 2 hours, filtered, washed with 2-propanol and dried. The residue (1.05g) was taken up in CH₂Cl₂ and basified with NH₄OH. The organic layer was separated, dried (MgSO₄), filtered and the solvent was evaporated. The residue (0.42g) was purified by column chromatography over silica gel (eluent: CH₂Cl₂/CH₃OH/NH₄OH 85/15/2; 15-40 μm). The pure fractions were collected and the solvent was evaporated. The residue (0.35g) was dissolved in CH₃OH and converted into the ethanedioic acid salt. The precipitate was filtered off and dried. This fraction was taken up in water and CH₂Cl₂ and alkalized with K₂CO₃ 10%. The organic layer was
- was purified by column chromatography over silica gel (eluent: CH₂Cl₂/CH₃OH/ NH₄OH 75/28/1; 15-40 μm). The pure fractions were collected and the solvent was evaporated, yielding 0.13g of compound (156).

separated, dried (MgSO₄), filtered and the solvent was evaporated. The residue (0.21g)

Example B2

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A mixture of intermediate (27) (0.02 mol) in hydrochloric acid (6N) (85ml) was stirred and refluxed at 50°C overnight and then brought to room temperature. The solvent was evaporated. The residue was taken up in K₂CO₃ 10% and extracted with CH₂Cl₂. The organic layer was separated, dried (MgSO₄), filtered and the solvent was evaporated, yielding 5g (69%) of N-(4-piperidinyl)-3-(8-quinolinylmethyl)-3H-imidazo[4,5-b]pyridin-2-amine (comp. 41).

Example B3

A mixture of intermediate (41) (0.00668 mol) in a solution of ammonia in methanol (7N) (70ml) was hydrogenated at room temperature under a 3 bar pressure for 5 hours

with Raney nickel (2.7g) as a catalyst. After uptake of hydrogen (2 equiv.), the catalyst was filtered through celite, washed with CH_2Cl_2 and CH_3OH and the filtrate was evaporated. The residue was taken up in CH_2Cl_2 and a small amount of CH_3OH . The organic solution was washed with H_2O , dried (MgSO₄), filtered and the solvent was evaporated. The residue was crystallized from EtOAc. The precipitate was filtered off and dried, yielding 1.6g (60%) of N-[1-(2-aminoethyl)-4-piperidinyl]-1-(4-quinolinyl-methyl)-1H-benzimidazol-2-amine; mp. 196°C (comp. 24).

Example B4

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A mixture of intermediate (36) (0.00351 mol) in hydrazine (2.5ml) and ethanol (30ml) was stirred and refluxed for 20 minutes and brought to room temperature. Ice water was added. The mixture was extracted with CH₂Cl₂ and separated into its layers. The aqueous layer was washed twice with CH₂Cl₂. The combined organic layer was dried (MgSO₄), filtered and the solvent was evaporated. The residue was taken up in diethyl ether. The precipitate was filtered off and dried, yielding 1g of (±)-N-[1-[1-(aminomethyl)-2-methylpropyl]-4-piperidinyl]-1-[ethoxy(8-quinolinyl)methyl]-1H-benzimidazol-2-amine; mp. 202°C (comp. 100).

Example B5

Intermediate (32) (0.1382 mol) was added at 55°C to a mixture of (±)-1-[ethoxy(3-methoxy-2-quinolinyl)methyl]-*N*-(4-piperidinyl)-1*H*-benzimidazol-2-amine (0.0346 mol) and potassium carbonate (0.242 mol) in acetonitrile (108ml) and DMF (20ml) (1 equiv of intermediate (32) was added every hour). The mixture was stirred at 55°C for 1 hour and filtered. The filtrate was poured out into H₂O and the mixture was extracted with EtOAc. The organic layer was separated, washed with a saturated NaCl solution, dried (MgSO₄), filtered and the solvent was evaporated. The residue was purified by column chromatography over silica gel (eluent: CH₂Cl₂/CH₃OH/ NH₄OH 98/2/0.4 and 96/4/0.5; 20-45 μm). Two fractions were collected and their solvents were evaporated, yielding 2.5g (23%) of (±)-1,1-dimethylethyl [1-[[4-[[1-[ethoxy(3-methoxy-2-quinolinyl)methyl]-1*H*-benzimidazol-2-yl]amino]-1-piperidinyl]methyl]-2-methylpropyl]carbamate (comp. 38).

30 Example B6

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A mixture of 1-[4-[[1-(2-quinolinylmethyl)-1*H*-benzimidazol-2-yl]amino]-1-piperidinyl]-3-methyl-2-butanone (0.0158 mol) and benzenemethanamine (0.0474 mol) in methanol (150ml) was hydrogenated at 40°C under a 3 bar pressure for 48 hours with palladium on activated carbon (0.7g) as a catalyst. After uptake of hydrogen (1 equiv), the catalyst was filtered through celite, washed with CH₂Cl₂/ CH₃OH and the filtrate was evaporated. The residue (11.5g) was purified by column chromatography

over silica gel (eluent: CH₂Cl₂/CH₃OH/NH₄OH 94/6/0.5; 20-45 μm). The pure fractions were collected and the solvent was evaporated, yielding 4g of residue. This fraction was converted into the hydrochloric acid salt with 2-propanol/ HCl. The precipitate was filtered off and dried, yielding 5.1g of product. This fraction was converted into the free base and then purified by column chromatography over C₁₈ (eluent: CH₃OH/NH₄OAc 60/40 and 80/20; column: KROMASIL C18). Two pure fractions were collected and their solvents were evaporated, yielding 0.8g of fraction 1 and 2g of fraction 2. Fraction 1 was crystallized from diethyl ether. The precipitate was filtered off and dried, yielding 0.5g of (±)-*N*-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-1-(2-quinolinylmethyl)-1*H*-benzimidazol-2-amine; mp. 135°C (comp. 6). Fraction 2 was dissolved in 2-propanol and converted into the hydrochloric acid salt (1:4). The precipitate was filtered off and dried, yielding 2.2g of (±)-*N*-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-1-[(1,2,3,4-tetrahydro-2-quinolinyl)methyl]-1*H*-benzimidazol-2-amine tetrahydrochloride monohydrate; mp. 230°C (comp. 46).

- a) A dispersion of sodium hydride in a mineral oil (60%) (0.01 mol) was added portionwise at 0° C under N_2 flow to a mixture of intermediate (38) (0.005 mol) in DMF (25ml). The mixture was stirred at room temperature for 1 hour. A solution of 2-(bromomethyl)-3-methoxyquinoline (0.0055 mol) in DMF (10ml) was added
- dropwise. The mixture was stirred at room temperature for 2 hours, hydrolized with K₂CO₃ 10% and extracted with EtOAc. The organic layer was separated, washed with NaCl, dried (MgSO₄), filtered and the solvent was evaporated, yielding 4.5g (>100%) of (±)-1,1-dimethylethyl [1-[[4-[[1-[(3-methoxy-2-quinolinyl)methyl]-1*H*-benzimi-dazol-2-yl]-amino]-1-piperidinyl]methyl]-2-methylpropyl]carbamate (comp. 14).
- b) A dispersion of sodium hydride in a mineral oil (60%) (0.014 mol) was added portionwise at 0°C under N₂ flow to a mixture of intermediate (38) (0.007 mol) in DMF (30ml). The mixture was stirred at 5°C for 1 hour. A solution of (±)-2,8-di-bromo-5,6,7,8-tetrahydroquinoline (0.0084 mol) in DMF (10ml) was added dropwise. The mixture was stirred at room temperature for 2 hours. H₂O and EtOAc were added.
- The organic layer was separated, washed with a saturated NaCl solution, dried (MgSO₄), filtered and the solvent was evaporated. The residue (5.6g) was purified by column chromatography over silica gel (eluent: CH₂Cl₂/CH₃OH/NH₄OH 97/3/0.5; 20-45 μm). The pure fractions were collected and the solvent was evaporated, yielding 1.1g (25%) of (±)-1,1-dimethylethyl [1-[[4-[[1-(2-bromo-5,6,7,8-tetrahydro-8-
- quinolinyl)-1*H*-benzimidazol-2-yl]amino]-1-piperidinyl]methyl]-2-methylpropyl]-carbamate (comp. 55).

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c) A mixture of intermediate 84 (0.0145 mol), 8-bromomethylquinoline (0.0174 mol) and K₂CO₃ (0.029 mol) in CH₃N (70ml) was stirred and refluxed for 4 hours, then brought to room temperature. The solvent was evaporated. The residue was taken up in H₂O and extracted twice with CH₂Cl₂. The organic layer was separated, dried (MgSO₄),

filtered and the solvent was evaporated. The residue was crystallized from diethyl 5 ether/CH₃CN. The precipitate was filtered off and dried, yielding 5.07g of compound 79 (74%).

Example B8

- c) $(\pm)-N-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-1-[(5,6,7,8-tetrahydro-3-methoxy-1-1-(5,6,7,8-tetrahydro-3-methylbutyl)-4-piperidinyl]-1-[(5,6,7,8-tetrahydro-3-methoxy-1-1-(5,6,7,8-tetrahydro-3-methylbutyl)-4-piperidinyl]-1-[(5,6,7,8-tetrahydro-3-methoxy-1-1-(5,6,7,8-tetrahydro-3-methoxy-1-1-(5,6,7,8-tetrahydro-3-methylbutyl)-4-piperidinyl]-1-[(5,6,7,8-tetrahydro-3-methoxy-1-1-(5,6,7,8-tetrahydro-3-methoxy-1-1-(5,6,7,8-tetrahydro-3-methoxy-1-1-(5,6,7,8-tetrahydro-3-methoxy-1-1-(5,6,7,8-tetrahydro-3-methoxy-1-1-(5,6,7,8-tetrahydro-3-methoxy-1-1-(5,6,7,8-tetrahydro-3-methoxy-1-1-(5,6,7,8-tetrahydro-3-methoxy-1-1-(5,6,7,8-tetrahydro-3-methoxy-1-1-(5,6,7,8-tetrahydro-3-methoxy-1-1-(5,6,7,8-tetrahydro-3-methoxy-1-1-(5,6,7,8-tetrahydro-3-methoxy-1-1-(5,6,7,8-tetrahydro-3-methoxy-1-1-(5,6,7,8-tetrahydro-3-methoxy-1-1-(5,6,7,8-tetrahydro-3-methox)-1-(5,6,7,8-tetrahydro-3-methoxy-1-1-(5,6,7,8-tetrahydro-3-methox)-1-($ 2-quinolinyl)methyl]-1H-benzimidazol-2-amine tetrahydrochloride monohydrate 10 (0.00218 mol) was basified with K2CO3 10%. The mixture was extracted with CH₂Cl₂. The organic layer was separated, dried (MgSO₄), filtered and the solvent was evaporated, to give A'. A mixture of A' in dichloromethane (50ml) was cooled to 0°C. A solution of tribromoborane in dichloromethane (0.01526 mol) was added dropwise.
- The mixture was stirred at room temperature overnight, poured out on ice, basified 15 with a concentrated NH₄OH solution, decanted and extracted with CH₂Cl₂. The organic layer was separated, dried (MgSO₄), filtered and the solvent was evaporated. The residue (1.1g) was purified by column chromatography over silica gel (eluent: CH₂Cl₂/CH₃OH/NH₄OH 90/10/0.5; 20-45 μm). The desired fractions were collected and the solvent was evaporated. The residue was converted into the hydrochloric acid 20 salt (1:4) with HCl/2-propanol. The precipitate was filtered off and dried, yielding
 - $0.5g (37\%) \text{ of } (\pm)-N-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-1-[(5,6,7,8-tetrahydro-$ 3-hydroxy-2-quinolinyl)methyl]-1H-benzimidazol-2-amine tetrahydro-chloride monohydrate; mp. 240°C (comp. 63).

- a) A mixture of compound 158 (0.0089 mol) in HCl 3N (40ml) was stirred at 100°C for 12 hours, then brought to room temperature and poured out on ice and NH₄OH. EtOAc was added. The precipitate was filtered off, washed with EtOAc and dried, yielding 2g of compound 159.
- b) A mixture of compound 168 (0.00895 mol) in HCl 3N (35ml) was stirred at 100°C 30 for 24 hours. The solvent was evaporated. The residue was taken up in EtOAc. The mixture was basified with NH4OH. The organic layer was separated, dried (MgSO4), filtered and the solvent was evaporated. Part of this fraction (0.7g) was crystallized from CH₃CN. The precipitate was filtered off and dried, yielding 0.3g of compound 35 167.

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c) A mixture of compound 176 (0.00373 mol) in HCl 3N (20ml) was stirred at 100°C for 12 hours, brought to room temperature, poured out on ice, basified with NH₄OH and extracted with EtOAc. The organic layer was separated, dried (MgSO₄), filtered and the solvent was evaporated. This fraction was dissolved in 2-propanol and converted into the hydrochloric acid salt (1:3). The precipitate was filtered off and dried, yielding 1.5g of compound 173 (77%).

Example B10

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A mixture of intermediate

(prepared according to A1b)), 1,2-ethanediamine (0.02 mol) and NaCN (0.0002 mol) in CH₃OH (7ml) was heated at 45°C for 4 hours and then brought to room temperature. Water was added. The mixture was extracted with CH₂Cl₂. The organic layer was separated, dried (MgSO₄), filtered and the solvent was evaporated. The residue (0.65g) was purified by column chromatography over silica gel (eluent:

 $CH_2Cl_2/CH_3OH/NH_4OH$ 90/10/1; 35-70 µm). The pure fractions were collected and the solvent was evaporated, yielding 0.42g of compound 170 (56%)

15 Example B11

A mixture of intermediate

(prepared according to A14a)) and formic acid/NH₃ (0.0462 mol) in formamide (35ml) was stirred at 140°C for 30 min and then brought to room temperature. CH₂Cl₂ was added. The organic layer was separated, washed with K₂CO₃ 10%, dried (MgSO₄), filtered and the solvent was evaporated. The residue (4g) was purified by column chromatography over silica gel (eluent: CH₂Cl₂/CH₃OH/NH₄OH 97/3/0.1; 15-40 μm). Two pure fractions were collected and their solvents were evaporated. The second fraction was crystallized from CH₃CN and diethyl ether. The precipitate was filtered off and dried, yielding: 1.37g of compound 137 (46%).

Isopropyl titanate (IV) (0.0294 mol) was added at room temperature to a mixture of intermediate 85 (0.0245 mol) and 1-acetylpiperazine (0.027 mol) in CH₂Cl₂ (50ml) and ethanol (50ml). The mixture was stirred at room temperature for 7 hours. NaBH3CN (0.0245 mol) was added portionwise. The mixture was stirred at room temperature for 12 hours. H₂O was added. The mixture was filtered over celite and washed with CH₂Cl₂. The filtrate was separated into its layers. The organic layer was dried (MgSO₄), filtered and the solvent was evaporated. The residue (6.7g) was purified by column chromatography over silica gel (eluent: CH₂Cl₂/CH₃OH/NH₄OH 95/5/0.2; 15-40 μm). The pure fractions were collected and the solvent was evaporated. This fraction was crystallized from 2-propanone. The precipitate was filtered off and dried, yielding: 10 0.64g of compound 176.

Tables 4 to 13 list the compounds of formula (I) which were prepared according to one of the above examples.

15 Table 4

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Comp No.	Ex. No.	a	Rª	R ^b	*	R ^C	Physical data
1	Bla	СН	Н	Н	2	н	HC1 (1:2)
2	Blb	CH	н	Н	2	**	
3	Blc	СН	н	н	2	CH ₂ CH ₂ NH ₂	HCl(1:4);H ₂ O(1:1)
4	Bla	CH	н	Н	8	Н	
5	Bla	СН	н	н	2	н	
6	В5	СН	н	н	2	CH ₂ CH(2-propyl)NH ₂	
7	В3	СН	н	н	8	CH(2-propyl)CH ₂ NH ₂	
8	В3	СН	н	Н	2	CH(2-propyl)CH ₂ NH ₂	H ₂ O (1:1)
9	Bla	СН	Н	8-Cl	2	н	HCl (1:2)
10	Blc	СН	Н	Н	8	CH ₂ CH(2-propyl)NH ₂	
11	В3	СН	Н	8-CI	2	CH(2-propyl)CH ₂ NH ₂	
12	Bla	СН	4-OH_	Н	2	Н	

Comp No.	Ex. No.	a	Rª	R ^b	*	R ^C	Physical data
13	В3	СН	н	8-C1	2	CH ₂ CH(2-propyl)NH ₂	
14	B6a	CH	3-OCH ₃	н	2	(C=O)OC(CH ₃) ₃	
15	Blc	СН	3-OCH ₃	н	2	CH ₂ CH(2-propyl)NH ₂	
16	B6a	N	3-CH₃	H	2	***	
17	Bla	СН	н	н	8	Н	HC1 (1:3)
18	Bla	N	н	н	8	н	
19	Blc	N	н	Н	8	CH ₂ CH(2-propyl)NH ₂	HCl(1:3); H ₂ 0(1:3
20	Bla	N	3-OCH ₃	н	2	H	
21	B4	N	3-OCH ₃	н	2	***	
22	Blc	N	3-OCH ₃	н	2	CH ₂ CH(2-propyl)NH ₂	
23	Bla	СН	Н	Н	4	Н	
24	B2	СН	H	H	4	CH2CH2NH2	
88	Bla	N	2-CH ₃	3-CH ₃	8	н	
89	Blc	N	2-CH ₃	3-CH ₃	8	CH ₂ CH(2-propyl)NH ₂	HCl(1:4);H ₂ O(1:2)
90	Bla	CH	2-CH ₃	H	8	H	
91	Blc	CH	2-CH ₃	H	8	CH ₂ CH(2-propyl)NH ₂	H ₂ O (1:1)
92	B2	CH	2-CH ₃	H	8	CH₂CH₂NH₂	
104	В3	CH	н	H	8	CH ₂ CH(2-propyl)NH ₂	
105	В3	CH	н	H	8	CH(2-propyl)CH ₂ NH ₂	
106	B1c	N	3-CH ₃	H	2	CH ₂ CH(2-propyl)NH ₂	H ₂ 0 (1:2)
109	B5	СН	H	Н	8	***	
110	B5	N	2-CH ₃	3-CH ₃	8	***	
111	B5	СН	2-CH ₃	Н	8	***	
112	B5	N	н	Н	8	***	
113	В7	СН	н	Н	8	***	

- * position bicyclic heterocycle
- ** (CH₂)₂NH(C=O)OC(CH₃)₃
- *** CH₂CH(2-propyl)NH(C=O)OC(CH₃)₃

Table 5

			K-				•	
Comp No.	Ex. No.	a	Rª	R ^b	*	R°	G	Physical data
25	Bla	СН	H	Н	2	H	CHOC₂H₅	
26	В3	СН	Н	н	2	CH(2-propyl)CH ₂ NH ₂	CHOC ₂ H ₅	H ₂ O (1:1)
27	В3	СН	Н	н	2	CH ₂ CH(2-propyl)NH ₂	CHOC ₂ H ₅	
28	Bla	СН	Н	н	2	H	***	
29	В3	СН	H	Н	2	CH(2-propyl)CH ₂ NH ₂	***	H ₂ O (1:1)
30	Bla	СН	H	Н	8	H	***	
31	В3	СН	Н	Н	8	CH ₂ CH(2-propyl)NH ₂	***	
32	В3	СН	H ·	н	8	CH(2-propyi)CH ₂ NH ₂	***	
33	Bla	CH	н	н	8	Н	CHOC ₂ H ₅	
34	Bla	CH	3-OCH₃	Н	2	Н	CHOC ₂ H ₅	
35	Bla	N	н	Н	2	Н	CH ₂	
36	B4	N	Н	Н	2	**	CH ₂	
37	Blc	N	Н	Н	2	CH ₂ CH(2-propyl)NH ₂	CH ₂	HCl (1:4)
38	B4	CH	3-OCH ₃	Н	2	**	CHOC ₂ H ₅	
39 ⁽⁹⁾	Blc	СН	3-OCH ₃	Н	2	CH₂CH(2-propyl)NH₂	CHOC₂H₅	HCl (1:3); H ₂ O (1:2)
40	B2	N	н	н	2	CH₂CH₂NH₂	CH ₂	
41	Bla	N	н	'Η	8	Н	CH ₂	
42	Blc	N	н	н	8	CH ₂ CH(2-propyl)NH ₂	CH ₂	
43	Bla	СН	Н	CH ₃	8	Н	CH ₂	
44	Bla	CH	H	CH ₃	8	Н	CHOC ₂ H ₅	
45	B2	N	Н	Н	8	CH ₂ CH ₂ NH ₂	CH ₂	
100	В3	СН	Н	Н	8	CH(2-propyl)CH ₂ NH ₂	CHOC ₂ H ₅	
107	Blc	СН	Н	H	8	CH ₂ CH(2-propyl)NH ₂	CHOC ₂ H ₅	
115	B5	СН	н	CH₃	8	CH(CH ₃) ₂ O C(CH ₃) ₃	CH₂	
116_	Bld	СН	Н	CH ₃	8	CH ₂ CH(2-propyl)NH ₂	CH ₂	

Comp No.	Ex. No.	а	Rª	R ^b	*	R°	G	Physical data
117	Bld	СН	Н	CH ₃	8	CH=O	CH ₂	
118	Bld	СН	H	CH ₃	8	CH₂CH₂NH₂	***	H ₂ O(1:1)
119	Bld	СН	Н	CH ₃	8	CH ₂ CH(2-propyl)NH ₂	***	
120	В3	N	H	CH ₃	8	CH ₂ CH ₂ NH ₂	CH ₂	HCl(1:4);
								H ₂ O(1:3)
121	Bld	СН	H	CH ₃	8	CH=O	***	
122	Blc	N	H	CH₃	8	CH ₂ CH(2-propyl)NH ₂	CH ₂	HCI(1:4);
								H ₂ O(1:1)
123	Bld	CH	H	CH ₃	8	CH ₂ CH ₂ NH ₂	CH ₂	
124	Blc	CH	Н	н	8	CH ₂ CH ₂ NH ₂	***	HCl(1:3);
								H ₂ O(1:2)
125	Blc	CH	Н	CH ₃	8	CH ₂ CH ₂ NH ₂	CHCH₃	H ₂ O(1:1)
126	Bld	СН	3-OCH ₃	Н	2	CH ₂ CH ₂ NH ₂	CH ₂	H ₂ O(1:2)
127	Blc	CH	4-CH ₃	H	2	CH ₂ CH(2-propyl)NH ₂	CH ₂	HCl(1:4);
1								H ₂ O(1:1)
128	Blc	СН	Н	Н	8	CH ₂ CH ₂ NH ₂	·CH ₂	HCl(1:4);
								H ₂ O(1:1)
129	Blc	CH	Н	H	8	CH₂CH₂NH₂	CHCH ₃	H ₂ O(1:1)
130	Blc	CH	4-CH ₃	Н	2	CH₂CH₂NH₂	CH ₂	HCl(1:4);
								H ₂ O(1:2)
131	Blc	СН	н	Н	4	CH ₂ CH(2-propyl)NH ₂	CH ₂	HCl(1:4);
								H ₂ O(1:2)
131	Blb	CH	н	CH ₃	8	C(CH ₃) ₃	CH ₂	
						N O C(C)		
132	B1b	СН	н	Н	8	0	CH ₂	
						N O C(CH ₃) ₃		
133	B2	СН	н	Н	8	Н	СНСН₃	HCl(1:2);
	1			'				H ₂ O(1:2)
134	Blc	СН	Н	Н	2	CH ₂ CH ₂ NH ₂	СНСН₃	H ₂ O(1:1)
135	B2	СН	4-CH ₃	н	2	Н	CH ₂	HCl(1:2)
136	B2	N	Н	CH ₃	8	Н	CH ₂	
137	B11	СН	Н	Н	8	CH=O	CH_2	

^{*} position quinoline

^{**} CH₂CH(2-propyl)NH(C=O)OC(CH₃)₃

^{***} CHO(CH₂)₂OC₂H₅

Table 6

Comp. No.	Ex. No.	*	G	Rª	Physical data
46	В5	2	CH ₂	CH ₂ CH(2-propyl)NH ₂	HCl(1:4);H ₂ O(1:1)
47	В5	8	CH ₂	CH ₂ CH(2-propyl)NH ₂	HCl(1:4);H ₂ O(1:1)
48	B5	8	-	Н .	
49	B5	8	<u> </u>	CH ₂ CH(2-propyl)NH ₂	H ₂ O (1:1)

^{*} position bicyclic heterocycle

5 <u>Table 7</u>

Co. No.	Ex. No.	*	a	Rª	G	R ^b	R ^c	Physical data
50	Bla	8	СН	н	-	Н	Н	
51	B 5	8	СН	н	-	CH ₂ CH(2-propyl)NH ₂	Н	
52	Bla	8	N	H	-	Н	Н	HCl (1:3)
53	В3	8	N	H	-	CH(2-propyl)CH ₂ NH ₂	Н	
54 ⁽³⁾	В3	8	N	н	-	CH ₂ CH(2-propyl)NH ₂	Н	H ₂ O (1:1)
55	Вбь	8	CH	2-Br	-	**	Н	
56	Blc	8	СН	2-Br	-	CH2CH(2-propyl)NH2	Н	HCl(1:3);H₂O(1:3)
57	B6b	8	CH	2-CH ₃	-	**	Н	

				T		I .	1	T
Co. No.	Ex. No.	*	a	Rª	G	R ^b	R°	Physical data
58	B1c	8	СН	2-CH ₃	-	CH ₂ CH(2-propyl)NH ₂	Н	HCl(1:4);H ₂ O(1:1)
59	B6a	2	СН	H	CH₂	**	Н	
60	B1c	2	СН	H	CH ₂	CH ₂ CH(2-propyl)NH ₂	Н	HCl(1:4);H ₂ O(1:1)
61	Вба	2	СН	з-осн	CH ₂	**	Н	
62	Blc	2	СН	з-осн	CH ₂	CH ₂ CH(2-propyl)NH ₂	Н	HCl(1:4);H ₂ O(1:1)
63	В7	2	СН	з-он	CH ₂	CH ₂ CH(2-propyl)NH ₂	н	HCl(1:4);H ₂ O(1:1)
64	Bla	8	N	з-Сі	-	Н .	Н	
65	B4	8	N	з-Сі	-	**	Н	
66	B1c	8	N	з-Сі	-	CH ₂ CH(2-propyl)NH ₂	Н	HCl(1:3);H ₂ O(1:1)
67	B2	8	N	H	-	CH ₂ CH ₂ NH ₂	Н	HCl(1:3);H ₂ O(1:3)
68	Bla	8	N	2-CI	-	H	Н	
69	B4	8	N	2-Cl	-	**	Н	
70(10)	Blc	8	N	2-CI	-	CH ₂ CH(2-propyl)NH ₂	н	HCl(1:3);H ₂ O(1:1)
139	B1c	5	N	з-сі	-	CH ₂ CH ₂ NH ₂	CH ₃	HCl(1:3);H ₂ O(1:2)
140	Bld	5	N	H	-	CH ₂ CH(2-propyl)NH ₂	CH ₃	
141	Blc	5	N	2-Cl	-	CH ₂ CH ₂ NH ₂	CH ₃	HCl(1:3);H ₂ O(1:3)
142	Blc	5	N	2-CI	- 1	CH ₂ CH(2-propyl)NH ₂	CH ₃	•

- * position bicyclic heterocycle
- ** CH₂CH(2-propyl)NH(C=O)OC(CH₃)₃

Table 8

$$R^{a}$$
 R^{a}
 R^{b}
 R^{a}
 R^{a}
 R^{b}
 R^{a}
 R^{b}
 R^{a}
 R^{a

Comp. R^{a} R^{b} Ex. ь G R^{c} Physical data No No. 71 N N Н H Η 72 S N Н Н HBr(1:2);H₂O(2:1) 73 Bla o N Н Н

Comp. No	Ex. No.	а	ь	Rª	R ^b	G	R ^c	Physical data
74		N	N	Н	Н	CH ₂	н	
75		N	N	Н	н	CH ₂	CH ₂ CH ₂ NH ₂	H ₂ O (1:1)
76		0	СН	-	н	CH₂	н	
77		N	N	CH₃	н	CH ₂	н	
78	Blc	N	N	CH₃	Н	CH₂	CH₂CH₂NH₂	
79	ļ	S	СН	-	Н	CH₂	н	
80	Bla	s	N	 -	Н	CH ₂	н	HCl(1:2);H ₂ O(1:1)
81	B2	N	N	Н	Н	-	CH ₂ CH ₂ NH ₂	HCl(1:4)
82	Bla	N	N	Н	OCH₃	CH ₂	н	
83	В1ь	s	N	-	н	-	*	H ₂ O (1:1)
84	Blc	S	N	-	H	-	CH₂CH₂NH₂	-HCl(1:3);H ₂ O(1:1)
85	Blb	N	N	CH ₃	Н	CH ₂	*	
86	Blb	0	N	-	н	-	*	
87	Blc	0	N	-	Н		CH ₂ CH ₂ NH ₂	

^{*} CH₂CH₂NH(C=O)OC(CH₃)₃

Table 9

5.

$$R^a$$
— N — N H— N

Comp. No.	Ex. No.	Rª	Physical data		
102	Bla	Н	HCl (1:3)		
103	B5	CH ₂ CH(2-propyl)NH ₂	H ₂ O (1:1)		

Table 10

Rnr

Comp.	Ex.	R ^b	Rc	G-R ^a	Physical data
No.	No.				
93		Н	Н	-CH ₂ -N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N-N	
101		CH2CH2NH2	Н	-CH ₂ -N	
94		CH ₂ CH ₂ NH(C=O)O CH ₂ CH ₃	н	_CH ₂	
95		CH₂CH₂NH₂	Н	-CH ₂	
96	Bla	Н	Н	$-CH_2$ N CF_3	
97	В2	CH₂CH₂NH₂	Н	$-H_2C$ N N CF_3	HCl(1:3);H ₂ O(1:1)
98	Bla	Н	Н	-CH ₂	
99	Blc	CH ₂ CH(2-propyl)NH ₂	Н	—CH ₂	HCl(1:3);H ₂ O(1:3)
108	В5	CH₂CH(2-propyl)NH₂	Н	-H ₂ C	

Rnr

Comp.	Ex.	R ^b	R°	G-R ^a	Physical data
No.	No.			L	
114		*	Н	—CH ₂ ————————————————————————————————————	
143	В6	CH ₂ CH(2-propyl)NH ₂	СН₃	-H ₂ C	

^{*} CH₂CH(2-propyl)NH(C=O)OC(CH₃)₃

Table 11

5

Co No.	Ex. No.	a-a1-a2-a3	*	Rª	R°	R ^b	G	Physical data
144	Blc	CH=N-CH=C	8	Н	-	CH ₂ CH(2-propyl)NH ₂	CH ₂	HCl(1:3);
							-	H ₂ O(1:4)
145	Blc	CH=C-N=C	8	н	н	CH ₂ CH(2-propyl)NH ₂	CH₂	HCl(1:3);
]	H ₂ O(1:2)
146	Blc	CH=C-C=N	8	-	Н	CH ₂ CH(2-propyl)NH ₂	CH₂	HCl(1:3);
ł								H ₂ O(1:2)
147	B2	CH=C-CH=C	8	СН₃	CI	н	CH ₂	
148	В3	CH=N-CH=C	8	Н	-	CH₂CH₂NH₂	CHOC₂H₅	
149	B2	CH=C-CH=N	8	-	Н	н	CH ₂	HCl(1:2);
								H ₂ O(1:1)
150	Blc	CH=C-CH=C	7	CH₃	CI	CH2CH2NH2	CH₂	HCl(1:4);
			1					H ₂ O(1:2)
151	В3	CH=N-CH=C	8	Н	<u>.</u>	CH2CH2NH2	CH ₂	

Co No.	Ex. No.	a-a1-a2-a3	*	Rª	R°	R ^b	G	Physical data
152	B2	CH=N-CH=C	8	Н	-	Н	CH₂	HCl(1:4);
								H ₂ O(1:2)
153	B3	CH=C-CH=N	8		H	CH₂CH₂NH₂	CHOC ₂ H ₅	

o position bicyclic heterocycle

Table 12

0			Ř ^a	
Co No.	Ex. No.	Rª	R ^b	Physical data
154	Blc	Н	3-propylamine	HCl(1:3);H ₂ O(1:1)
155	Blb	Н	(H ₃ C) ₃ C N NH ₂	
156	Ble	H	H ₂ N NH	
157	В7с	Н		trans
			(H ₃ C) ₃ C HN	
158	В7с	Н	H ₃ C NH	
159	B9a	Н	2-ethylamine	
160	Blc	н	3-propylmethylamine	
161	Blc	Н		cis;HCl(1:3);H ₂ O(1:1)
	:		H ₂ N N H	
162	Blc	Н	H ₂ N CH ₃	HCl(1:4);H₂O(1:1)
163	B4	н	3-isobutylamine	
164	Blc	Н	2-ethylmethylamine	HCI(1:2)

-	No.	I	ļ	Physical data
165	Bla	Н	$\langle \rangle$	trans;H ₂ O(1:1)
			H ₂ N	
166	B9a	CH ₃	2-ethylamine	
167	В9ь	Н		cis
		_	H ₂ N	
168	B7c	Н		cis
169	В3	н	H ₃ C H ₃	HCl(1:3);H₂O(1:2)
109	6 0	п.	H ₂ N CH ₂ -	110.(1.5),1120(1.2)
170	B10	Н	H N CH ₂ —	(1)
	Dio	•	H ₂ N O	9.1
171	B10	н	CH (CH ₃) ₂ H CH CH	H ₂ O(1:1)
			H ₂ N O	·
172	Blc	н	H ₂ N CH ₂	HCl(1:4);H ₂ O(1:2)
173	В9с	Н	HN CH ₂ —	HCl(1:3)H₂O(1:2)
174	Blc	Н	H ₂ N CH CH ₃	
175	В7с	н		cis
			(H ³ C) ³ C—O , III HIN	
176	B12	Н	H ₂ N CH CH ₃	

Table 13

$$G-N$$
 NH
 Ra
 Ra

Co No.	Ex. No.	G	L	a.	R _a .	Physical data
177	Bld	2-ethylamine	N CH ₃	СН	H	HCl(1:3);H ₂ O (1:3)
178	Blc	2-ethylamine	CH ₃	N	Н	HCl(1:4);H₂O (1:4)
179	Blc	2-ethylamine	N CH ₃	СН	СНЗ	H ₂ O(1:1)
180	В1ь	(H ₃ C) ₃ C O N CH ₂	N CH ₃	СН	Н	
181	Blc	CH(CH ₃) ₂ H ₂ N CH ₂ —		СН	Н	HCl(1:3);H ₂ O (1:2)
182	Blc	2-ethylamine	NH O	СН	Н	HCl(1:3);H₂O (1:2)
183	Blc	2-ethylamine		СН	Н	
184	Blc	2-ethylamine	N N	СН	Н	HCl(1:4);H ₂ O (1:1)
185	Bld	2-ethylamine	COL	СН	Н	C ₂ H ₂ O ₄ (2:7)

Table 14: Physical data

Comp.	C]	H		N	melting point
No.	Theor.	Exp.	Theor.	Exp.	Theor.	Exp.	
1	61.40	60.70	5.85	6.04	16.27	15.54	
3	51.08	51.16	6.07	6.35	14.89	14.17	
4	73.92	73.29	6.49	6.52	19.59	19.38	206°C
6	73.27	73.12	7.74	7.73	18.99	18.77	135°C
7	73.27	71.85	7.74	7.80	18.99	18.61	188°C
8	70.40	69.73	7.88	7.40	18.24	17.56	80°C
9	70.10	030	,				> 250°C
10	73.27	72.82	7.74	7.58	18.99	18.63	172°C
11	,3.27	72.02		,,,,,			190°C
13	67.98	66.43	6.97	6.79	17.62	17.02	164°C
15	71.16	70.66	7.68	7.58	17.78	17.81	210°C
19	51.45	51.64	6.97	6.89	16.15	15.96	240°C
22	68.47	68.04	7.45	7.52	20.70	20.55	206°C
23	73.92	71.70	6.49	6.53	19.59	19.92	140°C
24	71.97	69.89	7.05	7.10	20.98	20.07	196°C
89	51.46	53.22	6.94	7.11	15.00	15.14	24°C
91	70.85	69.82	8.07	8.29	17.71	17.48	180°C
92	72.43	71.51	7.29	7.30	20.27	20.10	176°C
104	72.87	70.26	7.53	7.27	19.61	18.73	88°C
105	72.87	71.37	7.53	7.39	19.61	19.39	135°C
106	65.69	66.19	7.96	7.62	19.86	19.71	110°C
26	69.02	69.16	7.99	7.68	16.65	16.79	140°C
27	71.57	70.60	7.87	7.80	17.27	17.14	166°C
29	67.86	67.64	8.08	7.79	15.32	15.15	100°C
31	70.16	68.97	7.98	7.97	15.84	15.56	110°C
32	70.16	69.35	7.98	8.34	15.84	14.73	98°C
33	71.79	70.72	6.78	7.17	17.44	16.69	145°C
37	,						215°C
39		<u> </u>					209°C
40	68.80	66.01	6.78	6.60	24.42	23.31	138°C
42	70.40	69.14	7.50	7.50	22.10	21.68	180°C
43	74.36	73.02	6.78	6.65	18.85	18.41	155°C
44	72.26	71.53	7.03	7.26	16.85	16.40	186°C
45	68.80	66.74	6.78	6.64	24.42	23.77	178°C
100	. 71.57	71.16	7.87	7.93	17.27	17.44	202°C
107	71.57	69.77	7.87	7.85	17.27	16.40	78°C

Comp.	(C		Н		N	melting point
	Theor.	Exp.	Theor.	Exp.	Theor.	Exp.	
46							230°C
47							230°C
48	72.59	71.54	7.25	7.13	20.16	19.91	205°C
49	69.30	70.08	8.50	8.37	18.65	18.93	140°C
51	72.19	70.66	8.39	8.43	19.43	18.79	120°C
53	69.25	68.88	8.14	8.28	22.61	22.23	
54	66.49	66.30	8.26	7.77	21.71	21.53	144°C
56	46.27	47.19	6.57	6.44	12.45	12.16	> 250°C
58							210°C
60							212°C
62	52.51	53.38	7.24	7.63	13.12	12.37	240°C
63	51.76	52.74	7.08	7.32	13.41	12.93	240°C
66	50.43	50.60	6.60	6.58	16.47	16.28	> 250°C
67	47.62	46.73	6.90	6.83	17.67	17.19	230°C
70							238°C
80							210°C
81	48.38	47.77	5.61	5.61			
82	67.00	66.51	6.43	6.29	22.32	22.12	
83	61.15	62.11	6.71	6.60	16.46	16.88	
84	48.51	48.46	5.62	5.35	16.16	16.03	
87	67.00	66.42	6.43	6.55	22.32	21.80	
103	68.78	68.77	8.31	8.23	19.25	18.78	88°C
96	58.73	58.59	5.16	5.03	22.83	22.40	144°C
97	1						210°C
99	53.51	52.63	7.15	7.02	13.87	13.24	200°C
108	70.08	68.99	7.92	8.10	22.00	21.65	160°C
116							203°C
117			ĺ			j	218°C
141						ŀ	225°C
177					ļ		>260°C
139			ĺ				190°C
118		ĺ	ŀ		İ		48°C
144				ĺ			220°C
143	70.55	66.03	8.11	8.14	21.33	18.98	ļ
119						•	145°C
121			1				185°C
140							172°C
120				l			210°C

Comp.	C	;]	H		N	melting point
No.	Theor.	Exp.	Theor.	Exp.	Theor.	Exp.	
142	THEOI.	LAP.	THEOI.	DAP.	THOUS.		98°C
122							245°C
154							90°C
145							190°C
123							194°C
124							150°C
146							240°C
125							74°C
178							160°C
150							>250°C
126							90°C
127	:					-	200°C
128							210°C
157							185°C
159			6				140°C
151							212°C
160	73.02	72.95	6.71	6.70	20.27	20.35	
129							170°C
130					i i		150°C
131					1		>250°C
152							230°C
153							169°C
131							120°C
161							206°C
132							160°C
133							210°C
134							81°C
162							210°C
147				ļ			>250°C
163					[168°C
179							116°C
135	62.16	62.10	6.12	6.06	15.76	15.71	
164					.		146°C
136		1					188°C
165]]				112°C
166							114°C
149							210°C
180							247°C

Comp. No.	(H		N	melting point
	Theor.	Ехр.	Theor.	Exp.	Theor.	Ехр.	
167							167°C
181							235°C
182							>250°C
184	47.75	47.58	6.01	6.37	17.72	17.00	
169							180°C
170		:					73°C
171					l		72°C
172							178°C
173							190°C
137							196°C
175							228°C
176							168°C
185							158°C

C. Pharmacological example

Example C1: In vitro screening for activity against Respiratory Syncytial Virus. The percent protection against cytopathology caused by viruses (antiviral activity or IC_{50}) achieved by tested compounds and their cytotoxicity (CC_{50}) were both calculated from dose-response curves. The selectivity of the antiviral effect is represented by the selectivity index (SI), calculated by dividing the CC_{50} (cytotoxic dose for 50% of the cells) by the IC_{50} (antiviral activity for 50 % of the cells).

10 Automated tetrazolium-based colorimetric assays were used for determination of IC50 and CC₅₀ of test compounds. Flat-bottom, 96-well plastic microtiter trays were filled with 180 μl of Eagle's Basal Medium, supplemented with 5 % FCS (0% for FLU) and 20 mM Hepes buffer. Subsequently, stock solutions (7.8 x final test concentration) of compounds were added in 45 µl volumes to a series of triplicate wells so as to allow 15 simultaneous evaluation of their effects on virus- and mock-infected cells. Five fivefold dilutions were made directly in the microtiter trays using a robot system. Untreated virus controls, and HeLa cell controls were included in each test. Approximately 100 TCID₅₀ of Respiratory Syncytial Virus was added to two of the three rows in a volume of 50 μ l. The same volume of medium was added to the third row to measure the cytotoxicity of the compounds at the same concentrations as those used to measure the 20 antiviral activity. After two hours of incubation, a suspension (4 x 10⁵ cells/ml) of HeLa cells was added to all wells in a volume of 50µl. The cultures were incubated at

10

37°C in a 5% CO2 atmosphere. Seven days after infection the cytotoxicity and the antiviral activity was examined spectrophotometrically. To each well of the microtiter tray, 25 μ l of a solution of MTT (3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide) was added . The trays were further incubated at 37°C for 2 hours, after which the medium was removed from each cup. Solubilization of the formazan crystals was achieved by adding 100 μ l 2-propanol. Complete dissolution of the formazan crystals were obtained after the trays have been placed on a plate shaker for 10 min. Finally, the absorbances were read in an eight-channel computer-controlled photometer (Multiskan MCC, Flow Laboratories) at two wavelengths (540 and 690 nm). The absorbance measured at 690 nm was automatically subtracted from the absorbance at 540 nm, so as to eliminate the effects of non-specific absorption.

Particular IC₅₀, CC₅₀ and SI values are listed in Table 15 hereinbelow. Table 15

Co. No.	IC ₅₀ (μM)	CC ₅₀ (µM)	SI
42	0.0004	>10.05	>25119
31	0.0008	12.68	15849
56	0.0016	12.71	7943
145	0.00631	25.12	3981
6	0.0126	10.00	794
156	0.01259	19.95	1585
131	0.0316	19.94	631
53	0.1259	>9.95	>79
29	0.3162	10.12	32
148	1	25	25
97	1.5849	>99.85	>63

Claims

5

10

15

1. A compound of formula

$$Q = \begin{pmatrix} R^1 & & & \\ & & & \\ & & & \\ N & & & \\$$

a prodrug, N-oxide, addition salt, quaternary amine, metal complex or stereochemically isomeric form thereof wherein

-a¹=a²-a³=a⁴- represents a bivalent radical of formula

-CH=CH-CH=CH-

(a-1);

-N=CH-CH=CH-

(a-2);

-CH=N-CH=CH-

(a-3);

-CH=CH-N=CH-

(a-4); or

-CH=CH-CH=N-

(a-5);

wherein each hydrogen atom in the radicals (a-1), (a-2), (a-3), (a-4) and (a-5) may optionally be replaced by halo, C₁₋₆alkyl, nitro, amino, hydroxy,

 C_{1-6} alkyloxy, polyhalo C_{1-6} alkyl, carboxyl, amino C_{1-6} alkyl, mono- or di $(C_{1-4}$ alkyl)amino C_{1-6} alkyl, C_{1-6} alkyloxycarbonyl, hydroxy C_{1-6} alkyl, or a radical of formula

wherein =Z is =O, =CH-C(=O)-NR^{5a}R^{5b}, =CH₂, =CH-C₁₋₆alkyl, =N-OH or =N-O-C₁₋₆alkyl;

20 Q is a radical of formula

wherein Alk is C₁₋₆alkanediyl:

Y¹ is a bivalent radical of formula -NR²- or -CH(NR²R⁴)-;

25 X¹ is NR⁴, S, S(=O), S(=O)₂, O, CH₂, C(=O), C(=CH₂), CH(OH), CH(CH₃), CH(OCH₃), CH(SCH₃), CH(NR^{5a}R^{5b}), CH₂-NR⁴ or NR⁴-CH₂;

 X^2 is a direct bond, CH₂, C(=O), NR⁴, C₁₋₄alkyl-NR⁴, NR⁴-C₁₋₄alkyl; t is 2, 3, 4 or 5; u is 1, 2, 3, 4 or 5; v is 2 or 3; and

whereby each hydrogen atom in Alk and the carbocycles and the heterocycles defined in radicals (b-3), (b-4), (b-5), (b-6), (b-7) and (b-8) may optionally be replaced by R³; with the proviso that when R³ is hydroxy or C₁₋₆alkyloxy, then R³ can not replace a hydrogen atom in the α position relative to a nitrogen atom;

G is a direct bond or C₁₋₁₀alkanediyl optionally substituted with one, two or three substituents selected from hydroxy, C₁₋₆alkyloxy, arylC₁₋₆alkyloxy, C₁₋₆alkylthio, arylC₁₋₆alkylthio, arylCarbonyl, HO(-CH₂-CH₂-O)_n-, C₁₋₆alkyloxy(-CH₂-CH₂-O)_n-, amino, mono-or di(C₁₋₆alkyl)amino, C₁₋₆alkyloxycarbonylamino and aryl;

R¹ is a bicyclic heterocycle selected from quinolinyl, quinoxalinyl, benzofuranyl, benzothienyl, benzimidazolyl, benzoxazolyl, benzthiazolyl, pyridopyridyl, naphthiridinyl, 1*H*-imidazo[4,5-b]pyridinyl, 3*H*-imidazo[4,5-b]pyridinyl, imidazo[1,2-a]pyridinyl, 2,3-dihydro-1,4-dioxino[2,3-b]pyridyl or a radical of formula

$$(CH_{2})_{m} \qquad (CH_{2})_{m} \qquad (CH_$$

and said bicyclic heterocycles may optionally be substituted in either of the two cycles
with 1 or where possible more, such as 2, 3 or 4, substituents selected from halo,
hydroxy, amino, cyano, carboxy, C₁₋₆alkyl, C₁₋₆alkyloxy, C₁₋₆alkylthio, C₁₋₆alkyloxyC₁₋₆alkyl, aryl, arylC₁₋₆alkyl, arylC₁₋₆alkyloxy, hydroxyC₁₋₆alkyl, mono-or di(C₁₋₆alkyl)amino, mono-or di(C₁₋₆alkyl)aminoC₁₋₆alkyl, polyhaloC₁₋₆alkyl, C₁₋₆alkylcarbonylamino,
C₁₋₆alkyl-SO₂-NR^{5c}-, aryl-SO₂-NR^{5c}-, C₁₋₆alkyloxycarbonyl, -C(=O)-NR^{5c}R^{5d}, HO(CH₂-CH₂-O)_n-, halo(-CH₂-CH₂-O)_n-, C₁₋₆alkyloxy(-CH₂-CH₂-O)_n-, arylC₁₋₆alkyloxy(CH₂-CH₂-O)_n- and mono-or di(C₁₋₆alkyl)amino(-CH₂-CH₂-O)_n-;
each n independently is 1, 2, 3 or 4;
each m independently is 1 or 2;

each p independently is 1 or 2;

each R^2 independently is hydrogen, formyl, $C_{1\text{-}6}$ alkylcarbonyl, Hetcarbonyl, pyrrolidinyl, piperidinyl, homopiperidinyl, $C_{3\text{-}7}$ cycloalkyl substituted with $N(R^6)_2$, or $C_{1\text{-}10}$ alkyl substituted with $N(R^6)_2$ and optionally with a second, third or fourth

- substituent selected from amino, hydroxy, C₃₋₇cycloalkyl, C₂₋₅alkanediyl, piperidinyl, mono-or di(C₁₋₆alkyl)amino, C₁₋₆alkyloxycarbonylamino, aryl and aryloxy; R³ is hydrogen, hydroxy, C₁₋₆alkyl, C₁₋₆alkyloxy, arylC₁₋₆alkyl or arylC₁₋₆alkyloxy; R⁴ is hydrogen, C₁₋₆alkyl or arylC₁₋₆alkyl;
 - $R^{5a},\,R^{5b},\,R^{5c}$ and R^{5d} each independently are hydrogen or $C_{1\text{-}6}$ alkyl; or
- 10 R^{5a} and R^{5b}, or R^{5c} and R^{5d} taken together form a bivalent radical of formula -(CH₂)_s-wherein s is 4 or 5;
 - R^6 is hydrogen, C_{1-4} alkyl, formyl, hydroxy C_{1-6} alkyl, C_{1-6} alkylcarbonyl or C_{1-6} alkyloxycarbonyl;
- aryl is phenyl or phenyl substituted with 1 or more, such as 2, 3 or 4, substituents selected from halo, hydroxy, C₁₋₆alkyl, hydroxyC₁₋₆alkyl, polyhaloC₁₋₆alkyl, and C₁₋₆alkyloxy;

Het is pyridyl, pyrimidinyl, pyrazinyl, pyridazinyl.

- A compound according to claim 1 wherein -a¹=a²-a³=a⁴- is a radical of formula
 (a-1), (a-2) or (a-3).
 - 3. A compound according to claim 1 or 2 wherein Q is a radical of formula (b-5) wherein v is 2 and Y¹ is -NR²-.
- 4. A compound according to anyone of claims 1 to 3 wherein R² is C₁₋₁₀alkyl substituted with NHR⁶.
- A compound according to anyone of claims 1 to 4 wherein G is a direct bond or C₁₋₁₀alkanediyl optionally substituted with one, two or three substituents selected from hydroxy, C₁₋₆alkyloxy, arylC₁₋₆alkyloxy, HO(-CH₂-CH₂-O)_n-, C₁₋₆alkyloxy(-CH₂-CH₂-O)_n-, arylC₁₋₆alkyloxy(-CH₂-CH₂-O)_n-.
- A compound according to claim 1 wherein the compound is selected from (±)-N-[1-(2-aminoethyl)-4-piperidinyl]-4-methyl-1-[1-(8-quinolinyl)ethyl]-1H-benzimidazol-2-amine monohydrate; (±)-N-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-1-(2-bromo-5,6,7,8-tetrahydro-8-quinolinyl)-1H-benzimidazol-2-amine trihydrochloride trihydrate; (±)-N-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-1-[(2-ethoxyethoxy)-8-quinolinylmethyl]-4-methyl-1H-benzimidazol-

-87-

2-amine; (\pm) -N-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-1-(2-chloro-5,6,7,8tetrahydro-5-quinoxalinyl)-1H-benzimidazol-2-amine trihydrochloride trihydrate; (\pm) -N-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-1-[(1-methyl-1H-benzimidazol-4yl)methyl]-1H-benzimidazol-2-amine; (±)-N-[1-(2-amino-3-methylbutyl)-4-5 piperidinyl]-1-(ethoxy-8-quinolinylmethyl)-1H-benzimidazol-2-amine; (±)-N-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-4-methyl-1-(5,6,7,8-tetrahydro-5quinoxalinyl)-1H-benzimidazol-2-amine; (±)-N-[1-(2-aminoethyl)-4-piperidinyl]-7-methyl-3-(8-quinolinylmethyl)-3*H*-imidazo[4,5-b]pyridin-2-amine tetrahydrochloride trihydrate; (±)-N-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-7-10 methyl-3-(8-quinolinylmethyl)-3H-imidazo[4,5-b]pyridin-2-amine tetrahydrochloride monohydrate; (±)-N-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-1-(8-quinolinylmethyl)-1H-imidazo[4,5-c]pyridin-2-amine trihydrochloride dihydrate; N-[1-(2-aminoethyl)-4-piperidinyl]-4-methyl-1-(8-quinolinylmethyl)-1H-benzimidazol-2-amine; N-[1-(8-quinolinylmethyl)-1H-benzimidazol-2-yl]-1,3-15 propanediamine trihydrochloride monohydrate; (±)-N-[1-(2-aminoethyl)-4piperidinyl]-1-[(2-ethoxyethoxy)-8-quinolinylmethyl]-1H-benzimidazol-2-amine trihydrochloride dihydrate; (±)-N-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-1-(8quinolinylmethyl)-1H-imidazo[4,5-b]pyridine-2-amine trihydrochloride dihydrate; (\pm) -N-[1-[1-(aminomethyl)-2-methylpropyl]-4-piperidinyl]-1-[(2-ethoxyethoxy)-8-20 quinolinylmethyll-1H-benzimidazol-2-amine: (±)-N-[1-(2-aminoethyl)-4piperidinyl]-3-(2-quinolinylmethyl)-3H-imidazo[4,5-b]pyridin-2-amine trihydrochloride trihydrate; (±)-N-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-1-(1isoquinolinylmethyl)-1H-benzimidazol-2-amine trihydrochloride trihydrate; N-[1-(2-aminoethyl)-4-piperidinyl]-1-(5,6,7,8-tetrahydro-5-quinoxalinyl)-1H-25 benzimidazol-2-amine trihydrochloride trihydrate; (±)-N-[1-(2-amino-3methylbutyl)-4-piperidinyl]-3-(quinolinylmethyl)-3H-imidazo[4,5-b]pyridin-2amine; (±)-N-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-4-methyl-1-(8quinolinylmethyl)-1H-benzimidazol-2-amine; (±)-N-[1-(2-aminoethyl)-4piperidinyl]-1-(2-chloro-5,6,7,8-tetrahydro-5-quinoxalinyl)-4-methyl-1H-30 benzimidazol-2-amine trihydrochloride.trihydrate; (±)-N-[1-(2-aminoethyl)-4piperidinyl]-1-(5,6,7,8-tetrahydro-2,3-dimethyl-5-quinoxalinyl)-1H-benzimidazol-2-amine trihydrochloride trihydrate; (±)-N-[1-(2-amino-3-methylbutyl)-4piperidinyl]-1-[(2-ethoxyethoxy)-8-quinolinylmethyl]-1H-benzimidazol-2-amine; (±)-N-[1-(2-amino-3-methylbutyl)-4-piperidinyl-1-(3-chloro-5,6,7,8-tetrahydro-5-35 quinoxalinyl)-1H-benzimidazol-2-amine trihydrochloride monohydrate; (±)-N-[1-(2-aminoethyl)-4-piperidinyl]-1-(3-chloro-5,6,7,8-tetrahydro-5-quinoxalinyl)-4methyl-1H-benzimidazol-2-amine trihydrochloride dihydrate; (\pm)-N-[1-(2aminoethyl)-4-piperidinyl]-1-[(2-ethoxyethoxy)-8-quinolinylmethyl]-4-methyl-1H-benzimidazol-2-amine monohydrate; (±)-N-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-3-(8-quinolinylmethyl)-3H-imidazo[4,5-c]pyridin-2-amine trihydrochloride tetrahydrate; (±)-N-[1-(2-aminoethyl)-4-piperidinyl]-3-(8-quinolinylmethyl)-3H-imidazo[4,5-b]pyridin-2-amine; (±)-N-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-4-methyl-1-[(1-methyl-1H-benzimidazol-4-yl)methyl]-1H-benzimidazol-2-amine; (±)-N-[1-(2-amino-3-methylbutyl)-4-piperidinyl]-1-(2-chloro-5,6,7,8-tetrahydro-5-quinoxalinyl)-4-methyl-1H-benzimidazol-2-amine; a prodrug, N-oxide, addition salt, quaternary amine, metal complex or stereochemically isomeric form thereof.

- 7. A compound according to any one of claims 1 to 6 for use as a medicine.
- 8. A pharmaceutical composition comprising a pharmaceutically acceptable carrier, and as active ingredient a therapeutically effective amount of a compound as described in any one of claims 1 to 6.
 - 9. A process of preparing a composition as claimed in claim 8, <u>characterized in that</u>, a pharmaceutically acceptable carrier is intimately mixed with a therapeutically effective amount of a compound as described in any one of claims 1 to 6.

10. An intermediate of formula

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with R¹, G and -a¹=a²-a³=a⁴- defined as in claim 1, P being a protective group, and Q₁ being defined as Q according to claim 1 but being devoided of the R² or R⁶ substituent.

11. An intermediate of formula

$$(O \Longrightarrow) Q_3 \longrightarrow N \longrightarrow a^1 \longrightarrow a^2$$

$$(IX)$$

with R^1 , G and $-a^1 = a^2 - a^3 = a^4$ defined as in claim 1, and $(O=)Q_3$ being a carbonyl derivative of Q, said Q being defined according to claim 1, provided that it is devoided of the NR^2R^4 or NR^2 substituent.

5 12. An intermediate of formula

with R^1 , Q and $-a^1=a^2-a^3=a^4$ defined as in claim 1, and $(O=)G_2$ being a carbonyl derivative of G, said G being defined according to claim 1.

10 13. A process of preparing a compound as claimed in claim 1, characterized by,

a) reacting an intermediate of formula (II-a) or (II-b) with an intermediate of formula (III)

with R¹, G, Q and -a¹=a²-a³=a⁴- defined as in claim 1, and W₁ being a suitable leaving group, in the presence of a suitable base and in a suitable reaction-inert solvent;

b) deprotecting an intermediate of formula (IV)

$$P = Q_1 = \begin{bmatrix} R^1 \\ N \\ A \end{bmatrix} \begin{bmatrix} a^1 \\ a^2 \\ A \end{bmatrix} \begin{bmatrix} A^2 \\$$

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with R^1 , G, and $-a^1=a^2-a^3=a^4$ - defined as in claim 1, H-Q₁ being defined as Q according to claim 1 provided that R^2 or at least one R^6 substituent is hydrogen, and P being a protective group;

5 c) deprotecting and reducing an intermediate of formula (IV-a)

$$P \longrightarrow Q_{1a}(CH=CH) \longrightarrow N \longrightarrow a^{1 \atop a^{2} \atop a^{4} = a^{3}} \longrightarrow H \longrightarrow Q_{1} \longrightarrow N \longrightarrow a^{1 \atop a^{2} \atop a^{4} = a^{3}}$$

$$(IV-a) \longrightarrow (I-a) \longrightarrow (I-$$

with R^1 , G, and $-a^1=a^2-a^3=a^4$ defined as in claim 1, H-Q₁ being defined as Q according to claim 1 provided that R^2 or at least one R^6 substituent is hydrogen, Q_{1a}(CH=CH) being defined as Q₁ provided that Q₁ comprises an unsaturated bond, and P being a protective group;

d) deprotecting an intermediate of formula (V)

with R^1 , G, and $-a^1=a^2-a^3=a^4$ defined as in claim 1, and H_2N-Q_2 being defined as Q according to claim 1 provided that both R^6 substituents are hydrogen or R^2 and R^4 are both hydrogen;

e) deprotecting an intermediate of formula (VI)

$$P = P = Q_{2} - Q_{2} - Q_{2} - Q_{2} - Q_{3} - Q_{4} - Q_{3} - Q_{4} - Q_{5} - Q_{5$$

with R^1 , G, and $-a^1=a^2-a^3=a^4$ - defined as in claim 1, and H_2N-Q_2 being defined as Q according to claim 1 provided that both R^6 substituents are hydrogen or R^2 and R^4 are both hydrogen, and P being a protective group;

f) deprotecting an intermediate of formula (VII) or (VIII)

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$$P = Q_{1} \cdot (OP)$$

$$N = \begin{bmatrix} A^{1} & A^{2} & A^{$$

with R^1 , G, and $-a^1=a^2-a^3=a^4$ defined as in claim 1, $H-Q_1\cdot(OH)$ being defined as Q according to claim 1 provided that R^2 or at-least-one R^6 substituent-is-hydrogen-and provided that Q comprises a hydroxy moiety, $H_2N-Q_2\cdot(OH)$ being defined as Q according to claim 1 provided that both R^6 substituents are hydrogen or R^2 and R^4 are both hydrogen and provided that Q comprises a hydroxy moiety, and P being a protective group;

g) amination of an intermediate of formula (IX)

$$(O=)Q_3 \xrightarrow{N} A_{a}^{1} A_{a}^{2}$$
 amination
$$H_2N - Q_3H \xrightarrow{N} A_{a}^{1} A_{a}^{2}$$
 (I-a-1-2)

with R^1 , G, and $-a^1=a^2-a^3=a^4$ - defined as in claim 1, and H_2N-Q_3H being defined as Q according to claim 1 provided that both R^6 substituents are hydrogen or R^2 and R^4 are both hydrogen, and the carbon adjacent to the nitrogen carrying the R^6 , or R^2 and R^4 substituents contains at least one hydrogen, in the presence of a suitable amination reagent;

h) reducing an intermediate of formula (X)

NC-Q₄

$$\stackrel{A}{=}$$
 $\stackrel{A_1}{=}$
 $\stackrel{A_2}{=}$
 $\stackrel{A_1}{=}$
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 with R^1 , G, and $-a^1=a^2-a^3=a^4$ - defined as in claim 1, and $H_2N-CH_2-Q_4$ being defined as Q according to claim 1 provided that Q comprises a $-CH_2-NH_2$ moiety, in the presence of a suitable reducing agent;

i) reducing an intermediate of formula (X-a)

with G, and $-a^1=a^2-a^3=a^4$ defined as in claim 1, $H_2N-CH_2-Q_4$ being defined as Q according to claim 1 provided that Q comprises a $-CH_2-NH_2$ moiety, and $R^{1'}$ being defined as R^1 according to claim 1 provided that it comprises at least one substituent, in the presence of a suitable reducing agent and suitable solvent;

10 j) amination of an intermediate of formula (XI)

$$CH_2-Q_4$$
 R^1
 A_2
 A_3
 A_4
 A_3
 A_4
 A_4
 A_3
 A_4
 with R^1 , G, and $-a^1=a^2-a^3=a^4$ - defined as in claim 1, and H_2N -CH₂-CHOH-CH₂-Q₄-being defined as Q according to claim 1 provided that Q comprises a

CH2-CHOH-CH2-NH2 moiety, in the presence of a suitable amination reagent;

15 k) reacting an intermediate of formula (XII) with formic acid, formamide and ammonia

$$C_{1-4}\text{alkyl} - CH_2 - Q_1 - N - A_3 - A_4 - A_3$$

$$(XII)$$

$$(XII)$$

with R^1 , G, and $-a^1=a^2-a^3=a^4$ defined as in claim 1, and H-C(=O)- Q_1 being defined as Q according to claim 1 provided that R^2 or at least one R^6 substituent is formyl; amination of an intermediate of formula (XIII) by reaction with an intermediate of

20 l) amination of an intermediate of formula (XIII) by reaction with an intermediate of formula (XIV)

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$$(O=)Q_{5} \xrightarrow{R^{1}} A_{1} A_{2} A_{3} + R^{2a} - NH_{2} A_{2} A_{3} A_{3} A_{4} A_{3} A_{3}$$

$$(XIII) \qquad (XIV) \qquad (I-c)$$

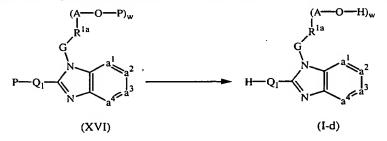
with R^1 , G, and $-a^1=a^2-a^3=a^4$ defined as in claim 1, and R^{2a} -NH-HQ₅ being defined as Q according to claim 1 provided that R^2 is other than hydrogen and is represented by R^{2a} , R^4 is hydrogen, and the carbon atom adjacent to the nitrogen atom carrying the R^2 and R^4 substituents, carries also at least one hydrogen atom, in the presence of a suitable reducing agent;

m) reducing an intermediate of formula (XV)

$$(R^{6})_{2}N_{-(C_{1}-9alkyl)-NH-HQ_{5}} - NH-HQ_{5} $

with R^1 , G, and $-a^1=a^2-a^3=a^4$ - defined as in claim 1, and $(R^6)_2N$ -[$(C_{1.9}alkyl)CH_2OH$]-NH-HQ₅ being defined as Q according to claim 1 provided that R^2 is other than hydrogen and is represented by $C_{1.10}alkyl$ substituted with $N(R_6)_2$ and with hydroxy, and the carbon atom carrying the hydroxy, carries also two hydrogen atoms, and provided that R^4 is hydrogen, and the carbon atom adjacent to the nitrogen atom carrying the R^2 and R^4 substituents, carries also at least one hydrogen atom, with a suitable reducing agent;

n) deprotecting an intermediate of formula (XVI), (XVI-a) or (XVI-b)



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with G, and $-a^1=a^2-a^3=a^4$ - defined as in claim 1, and H-Q₁ being defined as Q according to claim 1 provided that R² or at least one R⁶ substituent is hydrogen, and R^{1a}-(A-O-H)_w, R^{1a'}-(A-O-H)₂ and R^{1a''}-(A-O-H)₃ being defined as R¹ according to claim 1 provided that R¹ is substituted with hydroxy, hydroxyC₁₋₆alkyl, or HO(-CH₂-CH₂-O)_n-, with w being an integer from 1 to 4 and P or P₁ being a suitable protecting group, with a suitable acid.

o) amination of an intermediate of formula (XVII)

$$C_{1^{-4}alkyl} \longrightarrow C^{-Alk} \longrightarrow X^{1} \longrightarrow X^{1} \longrightarrow X^{2}A^{4}N $

with R^1 , G, $-a^1=a^2-a^3=a^4-$; Alk, X^1 R^2 and R^4 defined as in claim 1, in the presence of a suitable amination agent;

p) amination of an intermediate of formula (XIX)

$$H = C + C_{1-3}alkyl + NR^4 + A_{1-3}a^{1-3}a^{1-3} + Q_6N + CH_2 + C_{1-3}alkyl + NR^4 + A_{1-3}a^{1-3}a$$

with R^1 , G, and $-a^1=a^2-a^3=a^4$ defined as in claim 1, and Q_6N - CH_2 - C_{1-3} alkyl- NR^4

being defined as Q according to claim 1 provided that in the definition of Q, X^2 is C_{2-4} alkyl- NR^4 , in the presence of a suitable amination agent;

q) deprotecting an intermediate of formula (XXI)

$$P = O = G_1$$

$$Q = N$$

$$Q = A_1$$

$$Q = A_2$$

$$Q = A_3$$

$$Q = A_3$$

$$Q = A_4$$

$$Q = A_3$$

$$Q = A_4$$

$$Q$$

with R^1 , Q, and $-a^1=a^2-a^3=a^4$ - defined as in claim 1, and HO-G₁ being defined as G according to claim 1 provided that G is substituted with hydroxy or HO-(CH₂CH₂O-)_n;

r) reducing an intermediate of formula (XXII)

$$Q = \begin{pmatrix} R^1 \\ O =)G_2 \\ Q = \begin{pmatrix} R^1 \\ A = \\ A^2 \end{pmatrix} \begin{pmatrix} R^1 \\ A = \\ A^2$$

with R^1 , Q, and $-a^1=a^2-a^3=a^4$ defined as in claim 1, and H-G₂-OH being defined as G according to claim 1 provided that G is substituted with hydroxy and the carbon atom carrying the hydroxy substituent carries also at least one hydrogen, in the presence of a suitable reducing agent.

and, if desired, converting compounds of formula (I) into each other following artknown transformations, and further, if desired, converting the compounds of
formula (I), into a therapeutically active non-toxic acid addition salt by treatment
with an acid, or into a therapeutically active non-toxic base addition salt by
treatment with a base, or conversely, converting the acid addition salt form into the
free base by treatment with alkali, or converting the base addition salt into the free
acid by treatment with acid; and, if desired, preparing stereochemically isomeric
forms, metal complexes, quaternary amines or N-oxide forms thereof.

14. A product containing (a) a compound as defined in claim 1, and (b) another
 antiviral compound, as a combined preparation for simultaneous, separate or sequential use in the treatment or the prevention of viral infections.

15. A pharmaceutical composition comprising a pharmaceutically acceptable carrier and as active ingredients (a) a compound as defined in claim 1, and (b) another antiviral compound.



INTERNATIONAL SEARCH REPORT

Interna si Application No PCT/EP 00/05677

	1.0., 2.	10/ 056/ /
IPC 7	A61K31/4409 A61K31/436 A61P11/00 A61P31/12 C07	K31/498 D471/04 D491/056
	SEARCHED	
	cumentation searched (classification system followed by classification symbols)	
IPC 7	C07D A61K A61P	
Documenta	tion searched other than minimum documentation to the extent that such documents are included in the fields	searched
	ata base consulted during the international search (name of data base and, where practical, search terms us ternal, WPI Data, CHEM ABS Data	seu)
C. DOCUM	ENTS CONSIDERED TO BE RELEVANT	
Category °	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Α	WO 92 01697 A (JANSSEN PHARMACEUTICA NV) 6 February 1992 (1992-02-06) page 21, line 9 - line 12; claim 1	1,7
P,X	WO 99 44596 A (JANSSEN PHARMACEUTICA) 10 September 1999 (1999-09-10) page 11, line 18 - line 19; claim 1	1,7
Fur	ther documents are listed in the continuation of box C. Patent family members are list.	ted in annex.
"A" docum	ategories of cited documents: "T" later document published after the or priority date and not in conflict votered to be of particular relevance invention "X" document of particular relevance; the document but published on or after the international "X" document of particular relevance; the document published after the or priority date and not in conflict votations are provided to be of particular relevance.	with the application but r theory underlying the ne claimed invention
filing "L" docum which citatio "O" docum other "P" docum	cannot be considered novel or car involve an inventive step when the is cited to establish the publication date of another is cited to establish the publication date of another in or other special reason (as specified) The intervention of the intervention or other is combined with one or other special prior to the interventional filing date but than the priority date claimed The intervention or other special reason (as specified) The intervention of the intervention of the same pate of the same of the same pate of the same pate of the same pate of the same of the same pate	document is taken alone ne claimed invention n inventive step when the more other such docu- vious to a person skilled
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filing "L" docum which citatic "O" docum other "P" docum later t	ent which may throw doubts on priority claim(s) or is cited to establish the publication date of another or or other special reason (as specified) ent referring to an oral disclosure, use, exhibition or means ent published prior to the international filing date but han the priority date claimed involve an inventive step when the 'Y' document of particular relevance; it cannot be considered to involve an document of particular relevance; it cannot be considered to involve an inventive step when the 'Y' document of particular relevance; it cannot be considered to involve an inventive step when the 'Y' document of particular relevance; it cannot be considered to involve an inventive step when the 'Y' document of particular relevance; it cannot be considered to involve an inventive step when the 'Y' document of particular relevance; it cannot be considered to involve an inventive step when the 'Y' document of particular relevance; it cannot be considered to involve an inventive step when the 'Y' document of particular relevance; it cannot be considered to involve an inventive step when the 'Y' document of particular relevance; it cannot be considered to involve an inventive step when the 'Y' document of particular relevance; it cannot be considered to involve an inventive step when the 'Y' document of particular relevance; it cannot be considered to involve an inventive step when the 'Y' document of particular relevance; it cannot be considered to involve an inventive step when the 'Y' document of particular relevance; it cannot be considered to involve an inventive step when the 'Y' document or involve an involve an inventive step when the 'Y' document or involve an	e document is taken alone ne claimed invention n inventive step when the more other such docu- vious to a person skilled ent family

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Internal Application No
PCT/EP 00/05677

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A. CLASSIFICATION OF SUBJECT MATTER IPC 7 //(C07D471/04,221:00,221:00),(C07D471/04,235:00,221:00), (C07D491/056,319:00,221:00)							
According to International Patent Classification (IPC) or to both national classification	cation and IPC						
B. FIELDS SEARCHED							
Minimum documentation searched (classification system followed by classification system)	tion symbols)						
Documentation searched other than minimum documentation to the extent that							
Electronic data base consulted during the international search (name of data b	ase and, where practical, search terms used)						
C. DOCUMENTS CONSIDERED TO BE RELEVANT							
Category ° Citation of document, with indication, where appropriate, of the re	elevant passages Relevant to claim No.						
Further documents are listed in the continuation of box C.	χ Patent family members are listed in annex.						
Special categories of cited documents: "A" document defining the general state of the art which is not considered to be of particular relevance "E" earlier document but published on or after the international filling date "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "O" document referring to an oral disclosure, use, exhibition or other means "P" document published prior to the international filing date but later than the priority date claimed Date of the actual completion of the international search	T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention "X" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is taken alone "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combinad with one or more other such documents, such combination being obvious to a person skilled in the art. "&" document member of the same patent family Date of mailing of the international search report						
Name and mailing address of the ISA European Patent Office, P.B. 5818 Patentlaan 2 NL - 2280 HV Rijswijk Tel. (+31-70) 340-2040, Tx. 31 651 epo nl, Fax: (+31-70) 340-3016	Authorized officer Alfaro Faus, I						

1

INTERNATIONAL SEARCH REPORT i. armation on patent family members

Interna' il Application No PCT/EP 00/05677

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
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INTERNATIONAL SEARCH REPORT

International Application No. PCT/EP 00 \05677

FURTHER INFORMATION CONTINUED FROM PCT/ISA/ 210

Continuation of Box I.2

Present claims 1 to 15. relate to an extremely large number of possible compounds. Support within the meaning of Article 6 PCT and/or disclosure within the meaning of Article 5 PCT is to be found, however, for only a very small proportion of the compounds claimed. In the present case, the claims so lack support, and the application so lacks disclosure, that a meaningful search over the whole of the claimed scope is impossible. Consequently, the search has been carried out for those parts of the claims which appear to be supported and disclosed, namely those parts relating to the compounds of formula I where Q is a 1-R2-piperidy1-4-amino or amino(cyclo)alkylamino group and their intermediates as described in the examples of tables 3 to 13.

The applicant's attention is drawn to the fact that claims, or parts of claims, relating to inventions in respect of which no international search report has been established need not be the subject of an international preliminary examination (Rule 66.1(e) PCT). The applicant is advised that the EPO policy when acting as an International Preliminary Examining Authority is normally not to carry out a preliminary examination on matter which has not been searched. This is the case irrespective of whether or not the claims are amended following receipt of the search report or during any Chapter II procedure.

PATENT COOPERATION TREATY

	From the INTERNATIONAL BUREAU	
PCT COMMUNICATION IN CASES FOR WHICH NO OTHER FORM IS APPLICABLE	To: QUAGHEBEUR, Luc Janssen Pharmaceutica N.V. Patent Department - 3547 Turnhoutseweg 30 B-2340 Beerse BELGIQUE	JUL 1 7 2002 TECH CENTER 1600/2900
Date of mailing (day/month/year) 04 December 2001 (04.12.01)		<u> </u>
Applicant's or agent's file reference JAB 1500-PCT	REPLY DUE see paragraph 1 below	
International application No. PCT/EP00/05677	International filing date (day/month/year) 20 June 2000 (20.06.00)
Applicant JANSSEN PHARM	MACEUTICA N.V.	
	1 - Jan of mailing	
1. REPLY DUE within months/days from the	above date of maring	
NO REPLY DUE, however, see below	•	
MPORTANT COMMUNICATION		
INFORMATION ONLY		
2. COMMUNICATION:		ha abaya
It has been brought to the attention of the Inte identified application, the international public 2001 (04.01.01) erroneously indicated the nar Koenraad Jozef Lodenwijk Marcel. The corresponding to the Marcel.	ation No. WO 01/00013 published one of one inventor and applicant as	ANDRIES,
The International Bureau shall republish a co- corrected version of the corresponding PCT p	rrection in Section II of the PCT Ga camphlet will be published as early a	zette. A as possible.
A copy of this Notification is being sent to th designated/elected Offices concerned.	e receiving Office (RO/EP) and to t	he
The International Bureau of WIPO	Authorized officer Anman QIU	

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1211 Geneva 20, Switzerland

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P. INT COOPERATION TREA

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NOTIFICATION OF ELECTION

(PCT Rule 61.2)

To:

Commissioner
US Department of Commerce
United States Patent and Trademark
Office, PCT
2011 South Clark Place Room
CP2/5C24
Arlington, VA 22202
ETATS-UNIS D'AMERIQUE

Date of mailing (day/month/year)
02 February 2001 (02.02.01)

in its capacity as elected Office

International application No.
PCT/EP00/05677

Applicant's or agent's file reference JAB 1500-PCT

International filing date (day/month/year) 20 June 2000 (20.06.00)

Priority date (day/month/year) 28 June 1999 (28.06.99)

Applicant

JANSSENS, Frans, Eduard et al

1.	The designated Office is hereby notified of its election made:
	X in the demand filed with the International Preliminary Examining Authority on:
	18 November 2000 (18.11.00)
	in a notice effecting later election filed with the International Bureau on:
2.	The election X was
	was not
	made before the expiration of 19 months from the priority date or, where Rule 32 applies, within the time limit under Rule 32.2(b).

The International Bureau of WIPO 34, chemin des Colombettes 1211 Geneva 20, Switzerland

Authorized officer

Juan Cruz

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